St. John's University St. John's Scholar

Theses and Dissertations

2020

# CHEMICAL SYNTHESIS AND BIOLOGICAL EVALUATION OF BENZIMIDAZOLES AS CHEMOTHERAPEUTICS

Leonard Barasa Saint John's University, Jamaica New York

Follow this and additional works at: https://scholar.stjohns.edu/theses\_dissertations

Part of the Chemistry Commons

#### **Recommended Citation**

Barasa, Leonard, "CHEMICAL SYNTHESIS AND BIOLOGICAL EVALUATION OF BENZIMIDAZOLES AS CHEMOTHERAPEUTICS" (2020). *Theses and Dissertations*. 63. https://scholar.stjohns.edu/theses\_dissertations/63

This Dissertation is brought to you for free and open access by St. John's Scholar. It has been accepted for inclusion in Theses and Dissertations by an authorized administrator of St. John's Scholar. For more information, please contact fazzinol@stjohns.edu.

## CHEMICAL SYNTHESIS AND BIOLOGICAL EVALUATION OF BENZIMIDAZOLES AS CHEMOTHERAPEUTICS

A dissertation submitted in partial fulfillment of the requirements for the degree of

DOCTOR OF PHILOSOPHY

to the faculty of the

## DEPARTMENT OF GRADUATE DIVISION

of

## COLLEGE OF PHARMACY AND HEALTH SCIENCES

at

ST. JOHN'S UNIVERSITY

New York

by

Leonard Barasa

Date Submitted \_\_\_\_\_

Date Approved \_\_\_\_\_

LEONARD BARASA

PROF. SABESAN YOGANATHAN

© Copyright by Leonard Barasa 2020

All Rights Reserved

## ABSTRACT

# CHEMICAL SYNTHESIS AND BIOLOGICAL EVALUATION OF BENZIMIDAZOLES AS CHEMOTHERAPEUTICS

Leonard Barasa

Nitrogen-containing heterocycles are among the most important structural motifs of chemical substances, which are well represented among natural products, and pharmaceuticals. The presence of nitrogen in heterocycles help to modulates physicochemical properties and the pKa profile of therapeutic leads.

Benzimidazole is a heterocyclic structure and a privileged scaffold that is routinely used during drug discovery efforts. The benzimidazole scaffold has structural similarity to purine which makes it a useful structural motif for the development of pharmaceutical or biological interesting molecules. Benzimidazole derivatives possess a wide variety of biological activities, including anti-bacterial, anti-cancer, and anti-inflammatory activities. Development of synthetic methods to access benzimidazoles have become a focus of synthetic organic chemists, as they are useful building blocks for drug discovery efforts. We have developed two new synthetic methodologies to access benzimidazoles sand one new method to chemo-selectively alkylate indole-benzimidazole scaffold. We used these synthetic methodologies to synthesize new, drug-like benzimidazole compounds and evaluated their anticancer activity, and their ability to modulate Bone Morphogenetic Proteins (BMPs).

Our results indicate that several indole-based, lipid-based, and bis- benzimidazoles exhibit promising anticancer activity against several cancer cell lines. A new class of bisbenzimidazoles, show topoisomerase II inhibitory activity. In addition, substituted aryl benzimidazoles have been identified as new class of small molecules with promising BMP receptor agonistic activity, where they stimulated downstream cascade canonical Smad-signaling pathways in C2C12 cells. Our findings suggest that further development of these scaffolds could provide drug leads towards new chemotherapeutic agents and a new class of small molecule activators of BMP signaling pathway for the treatment of bone-fracture.

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

t.]

## ACKNOWLEDGEMENTS

ion With a sense of utmost gratitude, I would like to sincerely thank several people whose wit efforts have immensely contributed to the conclusion of my thesis during my term at St. h a John's University as a graduate student.

gre

der

's

att

ent

I would like to thank my mentor, Dr. Sabesan Yoganathan for his continuous support all at through my research. I also thank him for the knowledge he has provided me with and qu encouragement to think critically, which developed a skill for troubleshooting within me. ote His approach toward perceiving results and to develop on the results has led to a strong fro foundation in my research. I am forever grateful for him for mentoring me in this long m journey toward achieving my Ph.D. degree. I am also grateful to Dr. Vijaya Korlipara, Dr. the Sandra Reznik, Dr. Tanaji T. Talele and Dr. Aaron Muth for serving as committee members do in my defense. I am thankful to Dr. Vikas V. Dukhande, Dr. Woon-Kai Low, Dr. Jeanette cu C. Perron and Dr. Sandra Reznik for their collaboration in bioassays of the synthesized me compounds. I am grateful to Dr. Korlipara, Dr. Talele and Dr. Muth for providing me nt access to their equipment and chemicals from their labs. I am thankful to Dr. Vivek Gupta, or Dr. Steven M. Graham and Dr. Raymond S. Ochs for giving me an opportunity to work use with their LC-MS related projects. I am also thankful to lab mates for their support thi throughout my research. S

spa My sincere thanks go to St. John's University and especially the College of Pharmacy and
ce Health Sciences, and the Department of Pharmaceutical Sciences for providing me support
through weekly budgets, equipment for my research and also for providing me funding to
em complete my program. I am grateful to Dr. Carvalho, my supervisors Susana Solis, Joyce

ii

ph

asi

	der				
reat quote from	's				
he document or	att				
use this space to	Festa and Pratibha Agdern, for providing me the opportunity and a great experience in				
mphasize a key	teaching, both as an assistant and as a fellow. I also thank Suzette Weiss from the science				
oint. To place	supply department for her support in ordering supplies in a timely manner and for always				
his text box	wit being available for any help related to lab supplies.				
nywhere on the	h a				
age, just drag	gre				
t.]	at				
1	qu				
	ote				
	fro				
	m				
	the				
	do				
	cu				
	me				
	nt				
	or				
	use				
	thi				
	s				
	5				
	spa				
	ce				
	to				
	em				
	ph				
	asi				
	7e				

---

preat quote from the document or use this space to emphasize a key point. To place this text box inywhere on the page, just drag

der

's

att

ent

ion

## **TABLE OF CONTENTS**

ii ACKNOWLEDGEMENTSii
LIST OF TABLES
LIST OF FIGURESviii
gre LIST OF ABBREVIATIONSx
at CHAPTER I. INTRODUCTION1
qu 1.1 Nitrogen-Containing Heterocycles 1
ote 1.2 Benzimidazole: A Privileged Pharmacophore 1
fro 1.3 Synthetic limitations to access benzimidazoles
m 1.4 Synthetic limitation to N-alkylated 2-indolylbenzimidazoles
1.5 Benzimidazoles as Anti-Cancer Agents
1.5.1 Cancer and Current Treatment Options
1.5.2 Benzimidazole-based Anti-Cancer Drugs
cu 1.6 Benzimidazoles as Bone Morphogenetic Proteins (BMP) Modulators
me 1.6.1 Bone Morphogenetic Proteins
nt 1.6.2 Bone Morphogenetic Proteins in Fracture Repair
1.6.3 Small Molecules as Inhibitors of BMPs and their Therapeutic Application
1.6.4 Small Molecules as Activators of BMPs and their Therapeutic Application
CHAPTER II. DESIGN RATIONALE
thi 2.1 Benzimidazole derivatives as potential anti-cancer agents
s 2.1.1 2-substituted benzimidazole derivatives as potential anti-cancer agents
spa 2.1.2 Bis-benzimidazole derivatives as potential topoisomerase II inhibitors17
ce 2.2 Benzimidazole derivatives as BMPs agonists
to CHAPTER III. EXPERIMENTAL
3. 1 Chemical Synthesis
iv

 $\mathbf{p}\mathbf{h}$ 

asi

	der		
great quote from	,		
he document or	Ś		
se this space to	att	3.1.1 Materials and Instrumentation.	20
mphasize a kev	ent	3.1.2 General procedure for the synthesis of $41a - 41g$ , $43a - 43f$ , $45a - 45n$ , $46 - 50$	),
	ion	79a – 79f, 80a – 80f and 84a – 84n benzimidazole analogs	21
omt. To place	wit	3.1.3 General procedure for the synthesis of 64a – 64r, 66a – 66e, and 68a – 68k	
his text box	hа	benzimidazole analogs	40
nywhere on the	пu	3.1.4 General Procedure for the selective mono-alkylation of 2-indolyl-benzimidazol	les
age, just drag	gre	71a – 71j, 72a – 72f and 73a – 73e.	50
	at	3.1.5 General procedure for the synthesis of symmetrically or asymmetrically bis-	
t.]	qu	alkylated 2-indolylbenzimidazoles 75a – 75c and 77a – 77d.	58
	1	3.2 Cell culture	60
	ote	3.3. Cell viability determination (MTT assay)	61
	fro	3.4. Apoptosis / necrosis assay	61
	m	3.5 Western blot analysis	62
	the	3.6 Human topoisomerase II assay	62
	do	3.7 Materials for the BMP project	63
		3.8 Cell culture maintenance – C2C12 cells	64
	cu	3.9 Determination of cell concentration – C2C12 cells	65
	me	3.10 Determining cell viability using MT-Glo assay	65
	nt	3.11 Treatment of C2C12 cells and preparation of whole cell lysates	66
	or	3.12 Determination of Total Protein Concentration: Protein Assays	67
	use	3.13 Western Blotting to determine Smad phosphorylation	67
	thi (	CHAPTER IV. RESULTS AND DISCUSSION	69
	ci ii	4.1 O-(benzotriazole-1-yl)-N,N,N',N'-tetramethyluronium hexafluorophosphate	
	S	(HBTU) Promoted Synthesis of Benzimidazoles	69
	spa	4.1.1 One-pot, HBTU promoted strategy for the synthesis of benzimidazoles	71
	ce	4.1.2 Advantages of HBTU promoted approach in the synthesis of benzimidazoles	72

onium hexafluorophosphate
thesis of benzimidazoles71
ne synthesis of benzimidazoles72

v

4.1.4 A plausible mechanism for the HBTU promoted synthesis of benzimidazoles .....79 em

to

ph

- ze

	der	
great quote from	's	
he document or		
use this space to	4.2 A second library of aryl-benzimidazoles synthesized using the synthetic	
mphasize a key	ent methodology developed in our lab	80
point To place	ion 4.3 A Mild <i>N</i> -Alkylation Methodology for the Structure Diversification of Indolyl-	
	wit Benzimidazoles	83
his text box	4.3.1 Substrate scope evaluation of selective <i>N</i> -alkylation strategy	85
nywhere on the	4.3.2 Synthesis of symmetrical bis- <i>N</i> , <i>N</i> -alkylated 2-indolyl-benzimidazole	88
oage, just drag	gre 4.4 Synthesis of Bis-Benzimidazole derivatives	91
t.]	at 4.5 Synthesis of substituted phenylbenzimidazoles	94
	qu 4.6 Biological Evaluation of benzimidazoles	95
	ote 4.6.1 Evaluation of indole-based benzimidazoles for anti-cancer activity	95
	4.6.2 Evaluation of lipid-based benzimidazoles as potential anti-cancer agents	97
	4.6.3 HO-PI Assay	98
	4.6.4 Evaluation of bis-benzimidazole derivatives for anti-cancer activity	.100
	the 4.6.5 Evaluation of bis-benzimidazole derivatives as topoisomerase II inhibitors	.102
	do 4.6.6 Evaluation of aryl benzimidazoles as BMPs agonists	.104
	cu CONCLUSIONS	111
	me REFERENCES	113
	nt	
	or	
	0	
	use	
	thi	
	S	
	spa	
	ce	
	to	
	em	vi
	ph	*1
	asi	

reat quote from	der
he document or	's
se this space to	att LIST OF TABLES ent
emphasize a key point. To place his text box nywhere on the page, just drag t.]	ent ion Table 4.1. Optimization of conditions for the conversion of aryl-amide into benzimidazole. wit
	ce
	to em
	ph
	asi

	der
reat quote from	
ha document or	Ś
ne document of	att
se this space to	un
-	ent
mphasize a key	
aint To place	ion
onit. To place	wit
his text box	
	h a
nywhere on the	
age just drag	gre
lage, just mag	at
t.]	

att	LIST OF FIGURES
ent	LIST OF FIGURES
ion	Figure 1.1 Structures of common nitrogen-containing heterocycles
wit	Figure 1.2 Marketed drugs with benzimidazole scaffold
h a	Figure 1.3 Structures of PAD4 inhibitors
gre	Figure 1.4 Benzimidazole-based anti-cancer drugs
at	Figure 1.5. The canonical Smad-mediated and Smad-independent p38 MAPK pathways
qu	for BMP signal transduction are shown
ote	Figure 1.6 BMP signaling inhibitors of SMAD 1/5/8 phosphorylation
fro	Figure 1.7. BMPs activators or sensitizers in stem cell differentiation
m	Figure 2.1. 2-aryl and alkyl substituted benzimidazoles
the	Figure 2.2. Design strategy for bis-benzimidazole derivatives as potential Topo II
do	inhibitors
cu me	Figure 2.3. Design strategy for aryl-benzimidazole derivatives as potential BMPs agonists.
nt	
or	Figure 4.1 Structures of halogenated and N-Cbz protected amino acid based
use	benzimidazoles
thi	Figure 4.2 Synthesis of aryl-benzimidazoles
s	Figure 4.3 Synthesis of indole-based benzimidazoles
spa	Figure 4.4 Synthesis of benzimidazoles from alkyl-amides
ce	Figure 4.5 Synthesized bis-benzimidazole derivatives using chelidamic acid monohydrate
to	linker
em	
ph	viii

	der
reat quote from	's
he document or	att
se this space to	Figure 4.6 Synthesized bis-benzimidazole derivatives using 2,6-pyridine dicarboxylic aci
mphasize a key	ent linker
oint. To place	ion Figure 4.7 Apoptosis/pecrosis assay using fluorescence microscopy
his text box	wit
ills text box	Figure 4.8 Cell viability graphs of 79a, 80b and 80c against HeLa and MDA-MB231 tumo h a
nywhere on the	cell lines
bage, just drag	at Figure 4.9 Topo II agarose gel assay results
L•]	$_{\rm qu}$ Figure 4.10 Fluorescence imaging Immunoblotting Assay. Treatment of 82a, 82b, 82c, 82e
	ote 82h, 82g, 84a, 84b, and 84i in C2C12 cells 109
	fro Figure 4.11 Fluorescence Imaging. Treatment of 82c and 82e in C2C12 cells cause
	m translocation of pSmad into the nucleus
	the
	do
	cu
	me
	nt
	or
	use
	thi
	S
	spa
	ce
	to
	em
	ph
	asi

~

	der
great quote from	's
he document or	
se this space to	att
	ent
mphasize a key	ion
oint. To place	1011
1	wit
his text box	1
nywhere on the	ha
and inst dear	gre
age, just drag	at
t.]	aı

## LIST OF ABBREVIATIONS

	em		
/	ion	ATP	Adenosine Triphosphate
•	•,	ALK	Anaplastic Lymphoma Kinase
K	wit	BMP	Bone Morphogenetic proreins
•	h a	BMPRI	Bone Morphogenetic proreins Receptor I
	gre	BMPRII	Bone Morphogenetic proreins Receptor II
5	at	BMD	Bone Marrow Density
	qu	CLL	Chronic Lymphocytic Leukemia
	ote	Cbz	Carboxybenzyl
	ole	DNA	Deoxyribonucleic Acid
	fro	DMF	Dimethylformamide
	m	DM	Dorsomorphin
	the	RNA	Ribonucleic acid
	do	°C	Degree centigrade
	cu	CDCl <sub>3</sub>	Deuterated Chlorofoam
	me	CH <sub>3</sub> CN	Acetonitrile
	me	DIPEA	N,N-disopropylethylamine
	nt	DMSO	Dimethyl sulfoxide
	or	DMSO-d6	Deuterated Dimethyl sulfoxide
	use	DIC	N,N'-diisopropylcarbodiimide
	thi	EDCI	1-ethyl-3-(3-dimethylaminopropyl) carbodiimide
	S	h	hour
	600	Hz	Hertz
	spa	HBTU	O-(benzotriazole-1-yl)-N,N,N',N'-tetramethyluronium
	ce		hexafluorophosphate
	to	HCTU	O-(1H-6-chlorobenzotriazole-1-yl)-1,1,3,3-
	em		tetramethyluronium hexafluorophosphate
	ph		

Х

ze

	der		
great quote from	's		
he document or	ott		
use this space to	all H(	OBt	1-hydroxybenzotriazole
mphasize a key	ent H(	OCt	7-chloro-1-hydroxybenzotriazole
point. To place	ion HO	OAt	7-aza-1-hydroxybenzotriazole
his tout how	wit HI	RMS	High resolution mass spectometry
ills text box	<sub>ha</sub> Py	BOP	Tripyrrolidinophosphonium hexafluorophosphate
nywhere on the	gre J		Coupling Constant
oage, just drag	mi	in	Minutes
t.]	at NI	HS	N-hydroxysuccinimide
	qu Na	aHCO <sub>3</sub>	Sodium hydrogen carbonate
	ote rt		Room Temperature
	fro TI	LC	Thin layer chromatography
	m μN	M	Micro Molar
	the		
	do		
	cu		
	me		
	nt		
	or		
	use		
	thi		
	S		
	spa		
	ce		
	to		
	em		
	ph		
	asi		

xi

great quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

der

's

att

ent

## **CHAPTER I. INTRODUCTION**

## ion 1.1 Nitrogen-Containing Heterocycles

wit Nitrogen-containing heterocycles are the building blocks of life. They are the key h a constituents of both DNA and RNA.<sup>1</sup> Nitrogen-containing heterocycles are among the most gre important structural motifs of chemical substances, which are well represented among at natural products, key component of biologically active structures and among the most qu significant structural components of pharmaceuticals.<sup>2–8</sup> The presence of nitrogen in ote heterocyclic motifs help to modulates the physicochemical properties and the pKa profile fro of therapeutic leads.<sup>7,9</sup> Nitrogen-containing heterocycles (**Figure 1.1**), in light of their m importance, have drawn considerable attention of researchers and their synthesis has the always been a topic of interest to organic and medicinal chemists.



thi Figure 1.1 Structures of common nitrogen-containing heterocycles.

S

## 1.2 Benzimidazole: A Privileged Pharmacophore

ce

spa

Benzimidazole is a heterocyclic aromatic organic compound. It is an important scaffold to and a privileged structure that is routinely used in drug discovery.<sup>10</sup> Benzimidazole core em

1

ph

ze

preat quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

att

ent

t.]

The heterocyclic benzimidazole scaffold has structural similarity to purine (7, Figure 1.1) ion which makes it a useful structural motif for the development of molecules of wit pharmaceutical or biological interest.<sup>2,3</sup> Benzimidazole derivatives possess a wide variety h a of biological activities,<sup>10-29</sup> including anti-bacterial,<sup>30-40</sup> anti-HIV,<sup>41</sup> anti-cancer,<sup>42-54</sup> gre analgesic,<sup>55</sup> anti-inflammatory activities,<sup>56</sup> anti-ulcer,<sup>57</sup> acetylcholinesterase inhibitors,<sup>58</sup> at anti-tubercular,<sup>59</sup> anti-viral agent,<sup>60</sup> anthelmintic,<sup>61-64</sup> anti-malarial agent,<sup>65</sup> and antiqu hypertensive activities.<sup>66</sup> In addition, chemotherapeutic leads containing benzimidazole ote core are known to target several biological molecules, including topoisomerases, nucleic fro acids (DNA), polyADP ribose polymerase (PARP), microtubule, epidermal growth factor m receptors (EGFR), protein tyrosine kinases, and protein tyrosine phosphatases.<sup>12,67–70</sup> The the benzimidazole nucleus is used as a drug lead and many benzimidazole drugs are in the do market domain with Triclabendazole<sup>71</sup> (12) being the latest benzimidazole derivative drug cu to be approved by the FDA in the year 2019 for the treatment of fascioliasis (Figure 1.2). me

(5, Figure 1.1) is bicyclic in nature that consists of a fused benzene and an imidazole rings.



2

<sup>ce</sup> **Figure 1.2** Marketed drugs with benzimidazole scaffold.

to

em

ph

pr

ze

reat quote from he document or att se this space to ent mphasize a key ion oint. To place his text box h a nywhere on the gre age, just drag at

der

's

wit

t.]

qu substrate<sup>72,87,91,92</sup> in the presence of a catalyst or a promoter. Another classical approach

**1.3 Synthetic limitations to access benzimidazoles** 

ote involves reaction of an *o*-phenylenediamine with a carboxylic acid substrate under forcing fro conditions, in the presence of a mineral acid or acetic acid, and under refluxing m temperatures.<sup>93</sup> Other recent methods for the preparation of benzimidazole derivatives have

Development of synthetic methods to access benzimidazoles have become a focus of

synthetic organic chemists, as they are useful building blocks for drug discovery efforts.<sup>72</sup>

Various synthetic approaches are employed to generate structurally diverse benzimidazoles

to better understand SAR and optimize the desired medicinal properties of drug leads.<sup>73–90</sup>

One of the common synthetic approaches employed to access benzimidazoles involves

condensation-dehydration sequence of an o-phenylenediamine with an aldehyde

the

focused on the use of transition metal-catalysts.<sup>91,92,94-108</sup>

cu

do

The classically used approach (Phillip's method)<sup>109</sup> involves the condensation of an ome phenylenediamine with carboxylic acids or its derivatives, including heating the reagents nt together in the presence of aqueous hydrochloric acid (Scheme 1.1). Although effective, or this method is limited to structurally simple substrates that can withstand the acidic medium use at high temperatures.

thi

S

spa

ce

4N HCI (aq) 13 14

to

Scheme 1.1 A most commonly used synthetic method of benzimidazoles. em

ph

asi

ze

3

reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

t.]

att Due to the harsh reaction condition, the classical method shows very limited substrate ent scope, and cannot tolerate sensitive functional groups. In addition, limited availability of ion aldehyde substrates often hinders the application other available methods during medicinal wit chemistry efforts. Owing to the limited synthetic utility of existing methods, we envisioned h a developing a mild, functional group tolerant method for accessing diverse class of gre benzimidazole synthons (Scheme 1.2). In our proposed strategy, the substrates containing at carboxylic acid are coupled with 1,2-diaminobenzene derivatives using standard qu carbodiimide-based coupling conditions and then the amide precursor is converted into the ote benzimidazoles via a dehydrative cyclization step under a mild reaction condition.

fro

der

's

m



cu

Scheme 1.2. Proposed one-pot, two step synthesis of benzimidazole derivatives me

nt

#### 1.4 Synthetic limitation to N-alkylated 2-indolylbenzimidazoles or

use Indolylbenzimidazole, a hybrid benzimidazole derivative has emerged as an important scaffold during drug discovery efforts. N-alkylated indolylbenzimidazole scaffolds with thi different N-alkylation pattern have been found to be promising anticancer agents,<sup>110</sup> anti-S spa bacteria agents,<sup>38</sup> and highly effective reversible inhibitors of protein arginine deiminase 4 (PAD4, Figure 1.3).<sup>111–113</sup> PAD4 enzyme mediates the transformation of protein arginine ce into citrulline.<sup>112</sup> Citrullination of proteins has normal roles in gene regulation and to em pathological roles in immunological and inflammatory diseases.

4

ph

ze

der reat quote from 's he document or att se this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the gre age, just drag at t.] qu



ote Figure 1.3 Structures of PAD4 inhibitors.

fro

N-alkylation of indoles or benzimidazoles is typically carried out with strong bases, such m as potassium hydroxide. Such reagents that cannot differentiate the reactivity of an indole, the benzimidazole or any other reactive functional groups within a complex molecule.<sup>37,110,114</sup> do This reactivity profile makes it nearly impossible to chemo-selectively alkylate indolylcu benzimidazole hybrids. N-alkylation of indolyl-benzimidazoles generally involves the me separate N-alkylation of an indole's or benzimidazole's nitrogen prior to joining the two nt nuclei to form indolyl-benzimidazole hybrid structures (Scheme 1.3).<sup>37,110</sup> Such strategies or require a series of protecting group manipulations and are deemed less economical. Other use available methods for N-alkylation of heterocycles have typically focused on the use of thi ionic-liquids, other surfactant-type additives or metal catalyzed construction of S heterocycles.115-118 spa ce to em 5

ph

ze

der reat quote from he document or att use this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the gre age, just drag at t.]

's



qu Scheme 1.3. Separate *N*-alkylation of indole (20) and benzimidazole (21) motifs.

ote We envision developing a simple chemical method to chemo-selectively alkylate the fro indolylbenzimidazole scaffold. The method is a late-stage alkylation and takes advantage m of the pKa differences between the indolyl-nitrogen and the benzimidazole nitrogen. the

do

#### cu 1.5 Benzimidazoles as Anti-Cancer Agents

me

#### **1.5.1 Cancer and Current Treatment Options** nt

Cancer is a major public health problem worldwide and is the second leading cause of or death in the United States.<sup>119</sup> Cancer is a group of diseases characterized by the use uncontrolled growth and spread of abnormal cells, with the ability to spread to other parts thi of the body. If the spread is not controlled, it can lead to death.<sup>120</sup> According to the S American Cancer Society, approximately 21.7 million new cancer cases are expected by spa 2030.121 The mainstream treatment modalities,122 chemotherapy, radiotherapy, surgery and ce immunotherapy have experienced set-backs in the hard fought battle against cancer, with to toxicity and multidrug resistance (MDR) being the greatest stumbling blocks.<sup>123-125</sup> em

6

ph

asi

preat quote from the document or use this space to emphasize a key point. To place this text box mywhere on the page, just drag

t.]

der

's

ion

att Therefore, there is an urgent need for the development of new anti-cancer agents that ent exhibit improved efficacy and low adverse effects.

wit Many different types of drugs are used to treat cancer - either alone or in combination with h a other drugs or treatment options.<sup>126</sup> These drugs are very different in their chemical gre composition, how they are administered, how useful they are in treating certain types of cancer, and the side effects they exhibit.<sup>127</sup> Chlorambucil, an alkylating agent that works at by disrupting DNA replication process.<sup>128,129</sup> Side effects include bone marrow au ote suppression, nausea, mouth irritation, vomiting, skin reactions, hair loss, infertility, and fro diarrhea. Chlorambucil resistance in tumor cells has been reported to be secondary to: alterations in the transport of this agent, alterations in the kinetics of the DNA cross-links m formed by this agent; cytoplasmic metabolism of the chloroethyl alkylating moiety to the the inactive hydroxyethyl derivative via GSH/GST; and overexpression of metallothionein, do which confers resistance to cis-platinum and cross-resistance to melphalan.<sup>130</sup> 5cu Fluorouracil is an antimetabolite that interfere with DNA and RNA by acting as a substitute me for the normal building blocks of RNA and DNA.<sup>131</sup> Side effects include nausea, vomiting, nt hair loss, persistent hiccups, mucositis, and headache. Resistance to fluoropyrimidines is a or multifactorial event,<sup>132</sup> which includes transport mechanisms, metabolism, molecular use mechanisms, protection from apoptosis, and resistance via cell cycle kinetics. Etoposide is thi a topoisomerase II inhibitor,<sup>133</sup> forms a ternary complex with DNA and the topoisomerase S II enzyme, prevents re-ligation of the DNA strands, and by doing so causes DNA strands spa to break. Side effects include hair loss, low blood pressure, bone marrow suppression, and ce metallic food taste. Drug resistance towards Etoposide in human melanoma cells is to associated with drug-dependent apoptosis deficiency.<sup>134</sup> em

7

ph

asi

preat quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

## t.]

att **1.5.2 Benzimidazole-based Anti-Cancer Drugs** ent

Nitrogen-containing heterocyclic ring systems are employed to treat different types of ion cancer, and benzimidazole is one of them as exemplified by benzimidazole based antiwit cancer drugs<sup>135,136</sup> such as bendamustine  $(22)^{137}$ , is used for treatment of chronic h a lymphocytic leukemia (CLL), veliparib (23)<sup>138</sup> is an anti-cancer PARP inhibitor, gre selumetinib (24)<sup>139</sup> is an ATP-independent inhibitor of mitogen-activated protein kinase at (MEK or MAPK/ERK kinase) 1 and 2, galeterone (25)<sup>140</sup> is a steroidal anti-androgen qu which is used for the treatment of prostate cancer, and nocodazole  $(26)^{141}$  is a synthetic ote tubulin-binding agent that disrupts microtubule dynamics and this prevents mitosis and fro

induces apoptosis in tumor cells (Figure 1.4).

m

der

's



spa Figure 1.4 Benzimidazole-based anti-cancer drugs.

ce

to In an attempt to overcome adverse toxicity, drug-resistance and cancer-type based em specificity, scientists are striving to find new and better anti-cancer agents.<sup>135</sup> Since potent ph

ze

preat quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

t.]

# ote 1.6 Benzimidazoles as Bone Morphogenetic Proteins (BMP) Modulators

anticancer activity is associated with benzimidazole pharmacophore, it serves as a

promising template for the design of new anti-cancer leads. Current research efforts have

been heavily focused on the development of 2-substituted benzimidazoles, such as 2-aryl-

and 2-heteroaryl benzimidazoles as drug leads.<sup>12,38</sup> Moreover, 2-substituted bis-

benzimidazoles have been reported to exhibit remarkable cytotoxicity against various

## 1.6.1 Bone Morphogenetic Proteins

cancer cell lines as well.<sup>12,16,142,143</sup>

m

der

's

att

ent

ion

wit

h a

gre

at

qu

Bone Morphogenetic Proteins (BMPs) are a group of signaling molecules, which belong the to the TGF-β superfamily.<sup>144,145</sup> The bone morphogenetic protein (BMP) family of ligands do plays important roles in a multitude of processes during embryonic development and adult cu homeostasis by regulating cellular lineage commitment, morphogenesis, differentiation, me proliferation, and apoptosis of various types of cells throughout the body.<sup>146-148</sup> BMPs are nt promising molecules for tissue engineering and bone therapy.<sup>149,150</sup> Alternative strategy to or enhance BMP signaling by use of small molecules such as benzimidazoles<sup>151</sup> have been use reported. The use of small molecules which are simpler to synthesize and more costthi effective is preferred alternative to clinical use of recombinant BMPs. The recombinant S BMPs require harmful doses to achieve efficacy and is costly to synthesize the complex spa BMPs proteins.

ce

to Initially discovered for their ability to induce bone formation, BMPs are now known to em play crucial roles in all organ systems.<sup>152</sup> BMPs are important morphogens in

9

ph

ze

preat quote from the document or ase this space to emphasize a key point. To place this text box anywhere on the bage, just drag

der

's

att embryogenesis and development, and in maintenance of adult tissue homeostasis. Many ent processes in early development are dependent on BMP signaling for cell growth, apoptosis, ion and differentiation.<sup>153</sup> Due to their ubiquitous expression and importance as regulators wit throughout the body, deficiency in BMP production or functionality usually leads to h a marked defects or severe pathologies.<sup>145</sup> BMPs signal through cell surface receptor gre complexes that consist of two distinct transmembrane serine/threonine kinase receptors, at type I (BMPRI) and type II (BMPRII) which activate downstream signaling cascades qu

(Figure 1.5) in many developmental, physiological, and pathophysiological processes.<sup>154</sup>
 ote Intracellular signals for bone morphogenetic proteins (BMPs) and other members in the fro transforming growth factor (TGF)-β superfamily are mediated by Smad proteins.<sup>155</sup>
 m Receptor-regulated Smads (R-Smads) are activated by serine/threonine kinase receptors

the upon ligand binding. R-Smads then form hetero-oligomeric complexes with a commondo mediator Smad 4 (co-Smad).<sup>156</sup> This complex is then translocated into the nucleus to cu regulate the transcription of genes, broadly influencing growth and differentiation. Smads me 1, 5, and 8 are R-Smads activated by BMP receptors, whereas Smads 2 and 3 are activated nt by TGF-β and activin receptors.<sup>153</sup> Smad 4 is the only co-Smad isolated in mammals and or is shared by BMP and TGF- $\beta$ /activin signaling pathways. Smads 6 and 7 are anti-Smads, use which block signals by preventing the activation of R-Smads by serine/threonine kinase thi receptors. Anti-Smads are induced by ligand stimulation, suggesting that they constitute a S negative feedback loop in the signal transduction pathways of the TGF-B spa superfamily.152,155-159

10

ce

to

em

ph

asi ze



the

t.]

Figure 1.5. The canonical Smad-mediated and Smad-independent p38 MAPK pathways for BMP do signal transduction are shown [Image adapted from Chenard K. E et al., J. Biomed. Biotechnol. cu **2012**].<sup>153</sup>

me

nt

#### **1.6.2 Bone Morphogenetic Proteins in Fracture Repair**

or

Osteoporosis is a systemic disorder characterized as the depletion of bone mass with use structural deterioration of bone tissue.<sup>161</sup> This results in a decrease in bone mineral density thi (BMD) and a predisposition to fragility fractures. It is a widespread chronic metabolic S disease of the bone<sup>162</sup> and has been suggested to influence populations with different ethnic spa backgrounds, and the elderly are a high-risk group.<sup>163</sup> Osteoporotic fracture presents ce biomechanically impaired healing and seriously threatens human health as the global to population ages.<sup>164</sup> It is reported that about 21 million males and 137 million females age em 11 ph

ze

preat quote from the document or the document of the document of the document the document of the document of

att 50 years and older around the world are at high risk of osteoporotic fracture and these ent numbers are predicted to double by 2040.<sup>165</sup> Moreover, the World Health Organization ion predicted that half of global osteoporotic hip fractures will occur in Asia by 2050.<sup>166</sup> wit Bisphosphonates<sup>167</sup> are the most widely used drugs for the prevention and treatment of h a osteoporosis but the main concerns limiting their use are the rare side-effects, such as a gre typical femur fractures and osteonecrosis of the jaw, and unproven efficacy after 5 years at of treatment. Therefore, determining reasonable and effective approaches for diagnosis and qu treatment for osteoporotic fracture is of great significance.

ote

der

's

fro BMPs are unique growth factors that can induce the formation of bone tissue individually, and can induce the differentiation of bone marrow mesenchymal stem cells into m osteoblastic lineage and promote the proliferation of osteoblasts and chondrocytes.<sup>168</sup> the BMPs stimulate the target cells by specific membrane-bound receptors and signal do transduced through mothers against decapentaplegic (Smads) and mitogen-activated cu protein kinase (MAPK) pathways. It has been demonstrated that BMP-2, BMP-4, BMP-6, me BMP-7, and BMP-9 play an important role in bone formation and healing.<sup>179–182</sup> However, nt BMP-based therapy for fracture healing require high doses, costly to produce, and major or side-effects have been reported and their therapeutic use have been recently revisited.<sup>172</sup> use

- thi
- S

## **1.6.3 Small Molecules as Inhibitors of BMPs and their Therapeutic Application**

ce Synthetic small molecules have been widely used to control developmental signaling
 to pathways, as functional agonists or antagonists.<sup>173</sup> Compared to recombinant proteins,
 em synthetic small molecules can be more stable, easier to quantify for reproducible activity

12

ph

asi

reat quote from he document or se this space to ent mphasize a key ion oint. To place his text box nywhere on the age, just drag t.]

h a gre at

der

's

att

wit

ferroportin, the sole iron exporter in vertebrates. Loss of ferroportin activity prevents mobilization of iron to the bloodstream from intracellular stores in enterocytes, macrophages and hepatocytes.<sup>179</sup> Dorsomorphin (27, Figure 1.6) is the first known smallqu molecule inhibitor of BMP signaling, which inhibits BMP signals required for embryogenesis ote and iron metabolism.<sup>175,180</sup> Dorsomorphin selectively inhibits the BMP type I receptors fro

and dose-response, and far less expensive to produce.<sup>173,174</sup> To date, most of the small

molecules discovered to regulate BMP signaling are antagonists.<sup>175</sup> Recent work suggests

a role for BMP signals in regulating expression of hepcidin, a peptide hormone and central

regulator of systemic iron balance.<sup>176-178</sup> Hepcidin binds and promotes degradation of

ALK2, ALK3 and ALK6 and thus blocks BMP-mediated SMAD1/5/8 phosphorylation, m target gene transcription and osteogenic differentiation.<sup>181</sup> The role of dorsomorphin in the BMP signaling in iron homeostasis has been examined.<sup>175,182</sup> In vitro, dorsomorphin do inhibits BMP-, hemojuvelin- and interleukin 6-, stimulated expression of the systemic iron cu regulator hepcidin, which suggests that BMP receptors regulate hepcidin induction by all me of these stimuli. In vivo, systemic challenge with iron rapidly induced SMAD1/5/8 nt phosphorylation and hepcidin expression in the liver, whereas treatment with or dorsomorphin blocked SMAD1/5/8 phosphorylation, normalized hepcidin expression and use increased serum iron levels. These findings suggest an essential physiological role for thi hepatic BMP signaling in iron-hepcidin homeostasis.

S

spa

ce

- to
- em
- ph asi
- ze

der reat quote from he document or att use this space to ent ion point. To place his text box h a nywhere on the page, just drag at t.]



ote Figure 1.6 BMP signaling inhibitors of SMAD 1/5/8 phosphorylation.

fro

# <sup>m</sup> 1.6.4 Small Molecules as Activators of BMPs and their Therapeutic Application

the Recently, several small molecules have been identified as activators of the BMP pathway do (Figure 1.7). PD407824 (30) is a BMP sensitizer for human embryonic stem cu differentiation.<sup>173</sup> Ouinoline derivative KM11073 (31) enhances BMP-2-dependent me osteogenic differentiation of C2C12 cells via activation of p38 signaling and exhibits in nt vivo bone forming activity.<sup>183</sup> A01 (32) increases BMP-2 responsiveness by inhibiting or Smurf1-mediated Smad1/5 degradation.<sup>184</sup> SJ000063181 (33), SJ000291942 (34) and use SJ000370178 (35) are activators of the canonical BMP signaling pathway.<sup>147</sup> Sb 4 (36), Sb thi 5 (37), and Sb 6 (38) have been recently identified as agonists of BMP signaling S pathways.<sup>151</sup> However, most of the reported compounds show relatively low activity and spa fail to induce the generation of mature osteoblasts, which limits their therapeutic potential. ce Thus, we screened an in-house library of small molecules to discovery more effective BMP to activators. During the same time, we found a paper that reported a class of benzimidazoleem

14

ph

asi

great quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

der

's

h a

att based BMP agonists<sup>151</sup> (**37** and **38**) that are similar to the aryl-benzimidazoles we ent identified. Based on this literature validity, we focused our efforts in designing and ion synthesizing 2-substituted aryl benzimidazoles as potential agonists of BMPs signaling wit pathways.

gre CI Ο HC at qu ote PD407824 (30) KM11073 (**31**) fro m 0=0=0 the Ο do SJ000063181 (33) A01 (32) cu me  $NO_2$ Ĥ  $\cap$ nt SJ000291942 (34) SJ000370178 (35) or Br Br use thi N Sb 4 (36) Sb 5 (37) Sb 6 (38)

S

Figure 1.7. BMPs activators or sensitizers in stem cell differentiation. spa

- ce
- to
- am
- em
- ph
- asi
- ze

reat quote from he document or att se this space to ent mphasize a key oint. To place his text box nywhere on the age, just drag

t.]

**CHAPTER II. DESIGN RATIONALE** 

Benzimidazoles have revolutionized the drug discovery process by their diverse range of ion biological activities, which makes this scaffold an indispensable anchor for innovative drug wit discovery efforts. Therapeutic potential of benzimidazoles has attracted researchers to h a design and synthesize more potent derivatives with a wide range of pharmacological gre activities. We have used benzimidazole nucleus in designing different classes of at benzimidazole derivatives as drug leads with the aim of exploring their biological activities qu with regards to structure activity relationships (SAR) and if possible, deciphering their ote

mode of action.

fro

der

's

## m 2.1 Benzimidazole derivatives as potential anti-cancer agents.

the In our effort to address the challenges related to anti-cancer drug development, 2do substituted and bis-benzimidazole derivatives have been reported as remarkable cu cytotoxicity against various cancer cell lines.<sup>12,16</sup> With this understanding, we have me designed two classes of benzimidazole derivatives as potential anti-cancer leads.

nt

#### 2.1.1 2-substituted benzimidazole derivatives as potential anti-cancer agents

or

S

ze

Indole-based benzimidazoles have attracted the attention of medicinal chemists, 185, 186 and use our focus was to design this type of molecules as potential drug leads (Figure 2.1). thi

spa ce to em ph asi

16



fro Figure 2.1. 2-aryl and alkyl substituted benzimidazoles.

m In addition to indolyl-benzimidazoles derivatives, we also designed lipid-based the benzimidazoles (**Figure 2.1**) with the aim of studying the effect of lipid chain on the do activity of the benzimidazole scaffold. Lipid based benzimidazole derivatives is a new area cu of research that is yet to be explored.

me

```
2.1.2 Bis-benzimidazole derivatives as potential topoisomerase II inhibitors
nt
    Development of new anti-cancer Topo II inhibitors is necessary for improving cancer
or
    treatment.<sup>187</sup> Several benzimidazole derivatives are reported as novel Topo II
use
    inhibitors<sup>188,189</sup> and based on this information, we designed and synthesized a series of bis-
thi
    benzimidazoles as potential topoisomerase II inhibitors (Figure 2.2).
S
spa
ce
to
em
                                                                                            17
ph
```



Figure 2.2. Design strategy for bis-benzimidazole derivatives as potential Topo II cu inhibitors.

me

his

t.]

nt

## 2.2 Benzimidazole derivatives as BMPs agonists

or

Following a screening of a small set of aryl-benzimidazoles from an in-house library, we use identified 2-substituted aryl-benzimidazoles as promising agonists of BMP signaling thi pathway. During the same time, benzimidazoles Sb 5 (37) and Sb 6 (38) (Figure 2.3) were S reported as potential BMP agonists. However, these compounds are characterized by low spa activity and failed to induce the generation of mature osteoblasts, which limits their ce application to activate BMP signaling. We designed and synthesized a focused library of to em 18 ph

ze

preat quote from the document or the document of the document of the document the document of the document of the document of the document the document of the

der

's

att 2-arylbenzimidazole derivatives (**Figure 2.3**) to identify more effective agonists of BMP ent signaling pathway.



**Figure 2.3.** Design strategy for aryl-benzimidazole derivatives as potential BMPs agonists. fro



great quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

der

's

att

ent

## **CHAPTER III. EXPERIMENTAL**

## ion 3. 1 Chemical Synthesis.

# wit 3.1.1 Materials and Instrumentation.

h a All chemicals were procured from VWR International (Radnor, PA), Fisher Scientific gre (Hampton, NH), AK Scientific, Inc. (CA), Acros Organics (Geel, Belgium), Aldrich at Chemical Co. (Milwaukee, WI), Alfa Aesar (Ward Hill, MA), Arkpharm, Inc. (Arlington qu Heights, IL), Chem-Impex Int. Inc. (Wood Dale, IL), and were used without additional ote purification. Qualitative analysis of reactions was performed by thin layer chromatography fro (TLC) with silica gel G as the adsorbent (250 microns) on aluminum backed plates (Agela m Technologies) and Ultraviolet (UV) light at 254 nm or 365 nm for visualization purposes. the <sup>1</sup>H NMR experiments were performed using a Bruker 400 Ultrashield<sup>TM</sup> spectrometer (at do 400 MHz) equipped with a z-axis gradient probe. <sup>1</sup>H NMR chemical shifts were reported cu downfield from tetramethylsilane (TMS, an internal standard) in parts per million ( $\delta$  ppm) me for majority of the intermediates and all the target compounds. The <sup>1</sup>H NMR data are nt depicted as: chemical shift (multiplicity s (singlet), bs (broad singlet), d (doublet), t or (triplet), dd (doublet of doublets), dt (doublet of triplets), tt (triplet of triplets), m use (multiplet), H (number of protons) and J (coupling constant). Column chromatography thi purifications were performed using silica gel (40-63  $\mu$ m) purchased from Silicycle Inc. S

(Quebec City, CANADA). LR-LC/MS analyses were performed on single quadrupole an <sup>spa</sup> Agilent Technologies 1260 infinity series LC. The following method was used for <sup>ce</sup> verifying exact masses of compounds: Column = Agilent Poroshell 120 EC-C18 2.7  $\mu$ m, to 4.6 x 50 mm.; temperature = 300 K; solvent acetonitrile/water 70:30 (0.1% formic acid):

20

em

ph

ze

preat quote from the document or use this space to emphasize a key point. To place this text box inywhere on the page, just drag

der

's

gre

t.]

att flow rate 0.5 mL/min; isocratic; 3  $\mu$ L injection and each single run lasted for 5 min. ent Accurate mass measurements/high resolution mass spectra (HRMS) were obtained from ion the Columbia University Chemistry Department Mass Spectrometry Facility on a Waters wit Xevo G2-XS QToF mass spectrometer equipped with a H-Class UPLC inlet and a h a LockSpray ESI source.

# at 3.1.2 General procedure for the synthesis of 41a - 41g, 43a - 43f, 45a - 45n, 46 - 50, <sup>au</sup> 79a - 79f, 80a - 80f and 84a - 84n benzimidazole analogs

ote To a solution of commercially available carboxylic acid substrate (1.0 equiv) in 30 mL of fro toluene or DMF was added N-Diisopropylethylamine (1.9 equiv) and the solution was stirred for 10 min at room temperature. To the stirring solution, HBTU (2 equiv) was added m the and the reaction mixture stirred for another 10 min. To the stirring reaction mixture, Odo phenylenediamine (1.0 equiv) was added and stirred for 3-4 hours. Thereafter, the reaction cu was heated under reflux for 3-4 hours. The reaction was cooled to room temperature. The me solvent was removed in vacuo in the case of toluene, but for DMF, the reaction mixture was diluted with water and products were extracted using ethyl acetate (EtOAc). The nt organic layer was dried over anhydrous sodium sulfate, filtered and concentrated in vacuo. or use The crude product was purified using column chromatography using hexanes/EtOAc in an thi increasing polarity up to 1:1 mixture. The fractions containing the desired product were concentrated and crystallized in hexanes/EtOAc (1:1) to yield the product as a white solid. S

spa

## Tert-butyl (1H-benzo[d]imidazol-2-yl)methylcarbamate (41a),

ce

White solid, 0.47 g, 91%;  $R_f$  0.22 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3343.8, 2923.9, 1939.1, to 1738.9, 1683.4, 1527.6 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 12.21 (s, 1H, NH), 8.33 em

21

ph

asi
t.]

der

's

att (s, 1H, NH), 7.48 (m, 2H, Ar-H), 7.14 (m, 2H, Ar-H), 4.37 (d, *J*=5.9Hz, 2H, CH<sub>2</sub>), 1.35 (s, ent 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 156.2, 153.2, 79.7, 78.7, 28.7. LC-MS: ion (ESI) *m/z* calculated for C<sub>13</sub>H<sub>17</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 248.14, observed 248.20. wit

h a *Tert-butyl (S)-1-(methoxycarbonyl)-2-(1H-benzo[d]imidazol-2-yl)ethylcarbamate (41b),* gre White solid, 0.40 g, 92%; R<sub>f</sub> 0.41 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3300.1, 2979.8, 2618.5, at 1888.4, 1676.4, 1650.9cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.21$  (s, 1H, NH), 8.33 qu (s, 1H, NH), 7.49 (m, 2H, Ar-H), 7.27 (d, *J*=8.3Hz, 1H, Ar-H), 7.13 (m, 2H, Ar-H), 5.20 ote (q, *J*=6.4Hz, 1H, CH), 3.11 (dd, *J*=6.4, 6.4Hz, 1H, CH<sub>2</sub>), 3.06 (s, 3H, OCH<sub>3</sub>), 2.91 (dd, fro *J*=6.4, 6.4Hz, 1H, CH<sub>2</sub>), 1.39 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 169.8, m 155.9, 155.4, 125.3, 122.1, 79.7, 78.7, 60.2, 46.7, 37.4, 37.1, 35.3, 28.7, 21.2. LC-MS: the (ESI) *m/z* calculated for C<sub>16</sub>H<sub>22</sub>N<sub>3</sub>O<sub>4</sub> [M+H]<sup>+</sup> 320.16, observed 320.20.

### do 2-nonyl-1H-benzo[d]imidazole (41c),

<sup>cu</sup> White solid, 0.24 g, 90%;  $R_f$  0.47 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 2937.4, 2894.1, 2857.2, <sup>me</sup> 2562.8, 1949.0, 1926.9, 1886.9, 1739.9cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.17$ <sup>nt</sup> (s, 1H, NH), 7.45 (m, 2H, Ar-H), 7.10 (m, 2H, Ar-H), 2.79 (t, 2H, CH<sub>2</sub>), 1.75 (t, 2H, CH<sub>2</sub>), <sup>or</sup> 1.24 (s, 12H, CH<sub>2</sub>), 0.85 (t, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  155.6, 121.4, <sup>use</sup> 31.8, 29.4, 29.2, 29.1, 29.0, 28.0, 22.6, 14.4. LC-MS: (ESI) *m/z* calculated for C<sub>16</sub>H<sub>25</sub>N<sub>2</sub> <sup>thi</sup> [M+H]<sup>+</sup> 245.20, observed 245.20.

S

# 2-undecyl-1H-benzo[d]imidazole (41d), spa

White solid, 0.449 g, 92%;  $R_f$  0.24 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3296.9, 2952.9, 2921.1, 2848.8, 2775.9, 1935.9, 1738.8cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 12.18 (s, 1H, NH), 7.45 (m, 2H, Ar-H), 7.09 (m, 2H, Ar-H), 2.79 (t, 2H, CH<sub>2</sub>), 1.75 (t, 2H, CH<sub>2</sub>), 1.22

ph

asi

t.]

der

's

att (s, 16H, CH<sub>2</sub>), 0.84 (t, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 155.6, 122.1, 31.9, ent 29.7, 29.7, 29.7, 29.5, 29.4, 29.4, 29.4, 22.7, 14.1. LC-MS: (ESI) *m/z* calculated for ion C<sub>18</sub>H<sub>29</sub>N<sub>2</sub> [M+H]<sup>+</sup> 273.23, observed 273.20.

h a2-tridecyl-1H-benzo[d]imidazole (41e),

gre White solid, 0.46 g, 90%;  $R_f$  0.26 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3048.6, 2952.9, 2919.5, at 2849.0, 1897.4, 1778.9cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.15$  (s, 1H, NH), 7.45 qu (m, 2H, Ar-H), 7.10 (m, 2H, Ar-H), 2.78 (t, 2H, CH<sub>2</sub>), 1.75 (t, 2H, CH<sub>2</sub>), 1.26 (s, 21H, ote CH<sub>2</sub>), 0.85 (t, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  155.6, 122.1, 31.9, 29.7, 29.7, fro 29.7, 29.5, 29.4, 29.4, 29.4, 22.7, 14.1. **LC-MS**: (ESI) *m/z* calculated for C<sub>20</sub>H<sub>33</sub>N<sub>2</sub> [M+H]<sup>+</sup> m 301.26, observed 301.30.

### the 2-phenethyl-1H-benzo[d]imidazole (41f),

<sup>do</sup> White solid, 1.1 g, %; R<sub>f</sub> 0.24 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3028.9, 2677.1, 1928.2, 1644.1, <sup>cu</sup> 1623.9, 1591.2cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.26$  (s, 1H, NH), 7.48 (m, 2H, <sup>me</sup> Ar-H), 7.28 (m, 2H, Ar-H), 7.20 (m, 2H, Ar-H), 7.11 (m, 2H, Ar-H), 3.12 (s, 4H, (-CH<sub>2</sub>)<sub>2</sub>); <sup>nt</sup> <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  154.8, 141.5, 128.8, 128.8, 126.5, 33.8, 30.9. **LC-MS**: <sup>or</sup> (ESI) *m/z* calculated for C<sub>15</sub>H<sub>15</sub>N<sub>2</sub> [M+H]<sup>+</sup> 223.12, observed 223.10.

thi Benzyl 1-((1H-benzo[d]imidazol-2-yl)methylcarbamoyl)-2-phenylethylcarbamate (41g), White solid, 0.22 g, 68%;  $R_f$  0.38 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3280.7, 1691.9, 1648.3, 1539.5, 1439.8cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 12.21 (s, 1H, NH), 8.77 (d, J=5.0Hz, 1H, NH), 8.33 (s, 1H, NH), 7.52 (m, J=8.2, 3H, Ar-H), 7.22 (m, 14H, Ar-H), to 4.93 (m, 2H, CH<sub>2</sub>), 4.45 (s, 2H, CH<sub>2</sub>), 4.33 (m, 1H, CH), 3.095 (dd, J=3.4, 3.8Hz, 1H, CH<sub>2</sub>), em 2.84 (dd, J=3.4, 3.8 Hz, 1H, -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  172.3, 156.4, 152.3, ph

ze

use

t.]

der

's

att 138.6, 137.5, 129.7, 128.7, 128.5, 127.9, 126.7, 122.0, 79.7, 65.7, 49.1, 37.9, 37.7. LCent **MS**: (ESI) m/z calculated for C<sub>25</sub>H<sub>25</sub>N<sub>4</sub>O<sub>3</sub> [M+H]<sup>+</sup> 429.19, observed 429.20. ion

#### wit 2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (43a),

h a White solid, 0.46 g, 92%; R<sub>f</sub> 0.47 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 2991.5, 2829.5, 1705.8, gre 1674.6, 1623.9, 1602.3cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 12.93 (s, 1H, NH), 11.86 at (s, 1H, NH), 7.67 (d, *J*=7.2Hz, 1H, Ar-H), 7.56 (d, 1H, Ar-H), 7.34 (dd, *J*=8.8, 8.8Hz, 1H, qu Ar-H), 7.23 (m, 2H, Ar-H), 6.83 (dd, *J*=8.8, 8.8Hz, 1H, Ar-H), 3.79 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C ote NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 154.3, 146.7, 144.2, 135.2, 132.9, 129.4, 128.7, 122.9, fro 122.1, 118.9, 114.9, 114.1, 113.3, 113.2, 111.6, 102.2, 101.9, 55.7. **LC-MS**: (ESI) *m/z* m calculated for C<sub>16</sub>H<sub>14</sub>N<sub>3</sub>O [M+H]<sup>+</sup> 264.11, observed 264.10.

### the 5-fluoro-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (43b),

<sup>do</sup> White solid, 0.19 g, 53%;  $R_f$  0.38 (1:1/hexanes:EtOAc); IR: 3437.4, 3036.9, 1866.5, <sup>cu</sup> 1620.4, 1605.2, 1576.2 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 13.06 (s, 1H, NH), 11.86 <sup>me</sup> (s, 1H, NH), 7.59 (s, 1H, Ar-H), 7.41 (s, 1H, Ar-H), 7.35 (d, *J*=8.8Hz, 1H, Ar-H), 7.15 (s, <sup>nt</sup> 1H, Ar-H), 7.06 (t, *J*=7.9Hz, 2H, Ar-H), 6.84 (dd, *J*=2.4, 2.4Hz, 1H, Ar-H), 3.78 (s, 3H, <sup>or</sup> OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  154.3, 132.9, 129.1, 128.6, 114.2, 113.2, 102.2, <sup>use</sup> 102.1, 55.7. **LC-MS**: (ESI) *m/z* calculated for C<sub>16</sub>H<sub>13</sub>FN<sub>3</sub>O [M+H]<sup>+</sup> 282.10, observed <sup>thi</sup> 282.10.

S

## 5-chloro-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (43c), spa

ce White solid, 0.21 g, 59%;  $R_f$  0.47 (1:1/hexanes:EtOAc); IR: 3445.4, 2993.3, 2836.7, to 1870.7, 1618.2, 1576.3 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 13.14$  (s, 1H, NH), 11.91 em (s, 1H, NH), 7.69 (s, 1H, Ar-H), 7.65 (dd, J=8.6, 8.6Hz, 1H, Ar-H), 7.55 (dd, J=8.6, 8.6Hz, ph

ze

t.]

der

's

h a

att 1H, Ar-H), 7.34 (d, *J*=8.8Hz, 1H, Ar-H), 7.23 (m, 2H, Ar-H), 6.84 (dd, *J*=2.4, 2.4Hz, 1H, ent Ar-H), 3.78 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 154.3, 133.0, 128.8, 128.6, ion 114.4, 113.2, 102.4, 102.2, 79.7, 55.7. LC-MS: (ESI) *m/z* calculated for C<sub>16</sub>H<sub>13</sub>ClN<sub>3</sub>O wit [M+H]<sup>+</sup> 298.07, observed 298.10.

### gre 5-bromo-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (43d),

at White solid, 0.25 g, 70%; R<sub>f</sub> 0.42 (1:1/hexanes:EtOAc); IR: 3315.3, 2938.2, 2832.4, qu 1738.1, 1628.0, 1569.4 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 13.14$  (s, 1H, NH), 11.90 ote (s, 1H, NH), 7.8 (s, 1H, Ar-H), 7.71 (s, 1H, Ar-H), 7.60 (dd, J=8.6, 8.4Hz, 1H, Ar-H), 7.50 fro (dd, J=8.6, 8.4 Hz, 1H, Ar-H), 7.35 (m, 2H, Ar-H), 7.15 (dd, J=2.4, 1.7Hz, 2H, Ar-H), m 6.83 (dd, J=2.4, 2.4Hz, 1H, Ar-H), 3.79 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) the  $\delta$  154.3, 148.0, 147.7, 145.7, 143.3, 136.5, 134,4, 133.0, 128.7, 125.6, 125.1, 121.1, 120.4, do 115.1, 114.4, 113.2, 102.4, 102.2, 55.7. **LC-MS**: (ESI) *m/z* calculated for C<sub>16</sub>H<sub>13</sub>BrN<sub>3</sub>O cu [M+H]<sup>+</sup> 342.02, observed 342.10.

#### me 2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole-5-carbonitrile (43e),

<sup>nt</sup> White solid, 0.25 g, 70%; R<sub>f</sub> 0.40 (1:1/hexanes:EtOAc); IR: 3453.1, 3387.2, 3240.1, <sup>or</sup> 2970.2, 2218.1, 1738.0, 1623.1, 1595.9 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 13.49$ <sup>use</sup> (s, 1H, NH), 11.97 (s, 1H, NH), 8.13 (s, 1H, Ar-H), 7.73 (d, *J*=7.1 Hz, 1H, Ar-H), 7.61 (d, <sup>thi</sup> *J*=8.2 Hz, 1H, Ar-H), 7.34 (d, *J*=8.8 Hz, 1H, Ar-H), 7.23 (s, 1H, Ar-H), 7.15 (s, 1H, Ar-<sup>s</sup> H), 6.85 (dd, *J*=2.4, 2.4Hz, 1H, Ar-H), 3.79 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 <sup>spa</sup> MHz)  $\delta$  154.3, 133.3, 128.6, 128.2, 120.5, 114.9, 113.4, 104.3, 103.4, 102.3, 55.7. LC-<sup>ce</sup> MS: (ESI) *m/z* calculated for C<sub>17</sub>H<sub>13</sub>N<sub>4</sub>O [M+H]<sup>+</sup> 289.11, observed 289.10. to

25

om

em

ph

ze

preat quote from the document or the document the document of the document of the document of the document the document of the

der

's

att

2-(5-methoxy-1H-indol-2-yl)-5-methyl-1H-benzo[d]imidazole (43f), ent White solid, 0.32 g, 89%; R<sub>f</sub> 0.42 (9:1/CH<sub>2</sub>Cl<sub>2</sub>:MeOH); IR: 3461.6, 3009.7, 2970.5, ion  $1738.0, 1626.7, 1572.2 \text{ cm}^{-1}; {}^{1}\text{H} \text{NMR} (\text{DMSO-}d_6, 400 \text{ MHz}); \delta = 12.75 \text{ (s, 1H, NH)}, 11.79$ wit (s, 1H, NH), 7.52 (dd, J=8.2, 8.2Hz, 1H, Ar-H), 7.45 (s, 1H, Ar-H), 7.41 (dd, J=8.2, 8.2Hz, h a 1H, Ar-H), 7.33 (s, 1H, Ar-H), 7.13 (d, J=2.2Hz, 1H, Ar-H), 7.07 (s, 1H, Ar-H), 7.02 (m, gre 1H, Ar-H), 6.81 (dd, J=2.4, 2.4Hz, 1H, Ar-H), 3.78 (s, 3H, OCH<sub>3</sub>), 2.45 (s, 3H, -CH<sub>3</sub>); <sup>13</sup>C at NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 154.3, 146.6, 144.5, 133.2, 132.8, 131.0, 129.6, 124.4, qu 123.6, 118.6, 113.9, 111.3, 102.2, 101.6, 79.7, 55.7. LC-MS: (ESI) m/z calculated for ote  $C_{17}H_{16}N_{3}O$  [M+H]<sup>+</sup> 278.13, observed 278.10. fro

Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-3-methylbutylcarbamate (45a), White solid, the 0.48 g, 96%; R<sub>f</sub> 0.40 (1:1/hexanes:EtOAc); IR: 3345.2, 2958.5, 1680.1, 1524.6, 1443.6, do 1365.3, 1316.2 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.16$  (s, 1H, NH), 7.50 (m, 2H, cu Ar-H), 7.30 (d, 1H, *J*=8.1Hz, Ar-H), 7.14 (m, 2H, Ar-H), 4.84 (m, 1H, CH), 3.37 (s, 1H, me NH), 1.74 (m, 2H, CH<sub>2</sub>), 1.60 (m, 1H, CH), 1.34 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 0.92 (s, 6H, C(CH<sub>3</sub>)<sub>2</sub>); nt <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  156.6, 155.8, 143.5, 134.5, 122.1, 121.4, 118.9, 111.7, or 78.5, 48.1, 43.5, 28.7, 24.8, 23.2, 22.3. LC-MS: (ESI) *m/z* calculated for C<sub>17</sub>H<sub>26</sub>N<sub>3</sub>O<sub>2</sub> use [M+H]<sup>+</sup> 304.20, observed 304.20.

<sup>thi</sup> *Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-2-methylbutylcarbamate (45b)*, White solid, <sup>s</sup> 0.48 g, 96%; R<sub>f</sub> 0.56 (1:1/hexanes:EtOAc); IR: 3323.4, 2965.7, 1681.3, 1526.9, 1443.3, <sup>spa</sup> 1364.9, 1331.9cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.17$  (s, 1H, NH), 7.51 (m, 2H, <sup>ce</sup> Ar-H), 7.21 (d, 1H, *J*=8.8Hz, Ar-H), 7.15 (m, 2H, Ar-H), 4.64 (t, 1H, *J*=8.2Hz, CH), 3.39 <sup>to</sup> (s, 1H, NH), 1.74 (m, 2H, CH<sub>2</sub>), 1.60 (m, 1H, CH), 1.34 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 0.86 (d, 3H, em

26

ph

asi

t.]

der

's

h a

att J=2.2Hz, C(CH<sub>3</sub>)), 0.74 (d, 3H, J=6.7Hz, C(CH<sub>3</sub>)); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$ ent 156.0, 155.8, 155.5, 121.9, 118.9,111.8, 78.5, 54.4, 53.8, 38.8, 28.6, 26.2, 25.3, 16.0, 15.4, ion 11.8, 11.5. **LC-MS**: (ESI) m/z calculated for C<sub>17</sub>H<sub>26</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 304.20, observed wit 304.20.

gre *Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-2-methylpropylcarbamate (45c)*. White solid, at 0.48 g, 94%; R<sub>f</sub> 0.56 (1:1/hexanes:EtOAc); IR: 3325.7, 2964.0, 1681.4, 1529.7, 1443.3, qu 1365.3, 1302.5cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.16$  (s, 1H, NH), 7.51 (m, 2H, ote Ar-H), 7.19 (d, 1H, *J*=8.8Hz, Ar-H), 7.14 (m, 2H, Ar-H), 4.57 (t, 1H, *J*=7.8Hz, CH), 3.37 fro (s, 1H, NH), 2.21 (m, 1H, C(CH)), 1.38 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 0.92 (d, 3H, *J*=5.4Hz, C(CH<sub>3</sub>)), m 0.79 (d, 3H, *J*=5.2Hz, C(CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  155.9, 155.5, 121.6, the 118.9, 111.7, 78.5, 55.7, 32.6, 28.6, 19.7, 19.1. **LC-MS**: (ESI) *m/z* calculated for do C<sub>16</sub>H<sub>24</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 290.19, observed 290.20.

cu Tert-butyl 2-(1H-benzo[d]imidazol-2-vl)pyrrolidine-1-carboxylate (45d). White solid, 1.0 me g, 98%; Rf 0.38 (1:1/hexanes:EtOAc); IR: 2976.3, 1696.1, 1660.9, 1428.3, 1383.9, 1360.1, nt 1322.9, 1307.7 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 8.34$  (s, 1H, NH), 7.49 (m, 2H, or Ar-H), 7.13 (m, 2H, Ar-H), 4.95 (dd, 1H, J=6.7, 3.6Hz, CH), 3.61 (m, 1H, CH<sub>2</sub>), 3.41 (m, use 1H, CH<sub>2</sub>), 2.29 (m, 2H, CH<sub>2</sub>), 1.94 (m, 2H, CH<sub>2</sub>), 1.39 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSOthi d<sub>6</sub>, 100 MHz) δ 157.5, 156.9, 154.3, 153.8, 121.7, 115.2, 79.7, 79.2, 78.8, 56.2, 55.7, 47.2, S 46.9, 38.7, 33.7, 32.4, 28.6, 28.2, 24.3, 23.6. LC-MS: (ESI) m/z calculated for C<sub>16</sub>H<sub>22</sub>N<sub>3</sub>O<sub>2</sub> spa [M+H]<sup>+</sup> 288.17, observed 288.20.

ce

 Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-2-(4-(benzyloxy)phenyl)ethylcarbamate (45e),

 to

 White solid, 0.42 g, 84%; Rf 0.40 (1:1/hexanes:EtOAc); IR: 3325.7, 2982.9, 1676.9,

 ph

asi

t.]

der

's

att 1511.5, 1447.4, 1367.3, 1309.9, 1277.7, 1240.9, 1170.7, 1071.1, 1017.7, 962.1, 863.3, ent 813.8, 738.6cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz): δ = 12.22 (s, 1H, NH), 7.41 (m, 8H, ion Ar-H), 7.16 (m, 4H, Ar-H), 6.90 (d, 2H, *J*=7.7Hz, Ar-H), 5.04 (s, 2H, CH<sub>2</sub>), 4.95 (q, 1H, wit *J*=6.4Hz, CH<sub>2</sub>), 3.40 (s, 1H, NH), 3.29 (m, 1H, CH<sub>2</sub>), 3.02 (t, 1H, *J*=12.9Hz, CH), 1.32 (s, h a 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 157.4, 155.8, 155.6, 137.7, 130.7, 128.9, gre 128.2, 128.1, 114.9, 78.5, 69.6, 51.5, 28.6. LC-MS: (ESI) *m/z* calculated for C<sub>27</sub>H<sub>30</sub>N<sub>3</sub>O<sub>3</sub> at [M+H]<sup>+</sup> 444.23, observed 444.20.

Tert-butyl 1-(1H-benzo/d]imidazol-2-yl)-2-(benzyloxy)ethylcarbamate (45f), White solid, 0.42 g, 84%; Rf 0.34 (1:1/hexanes:EtOAc); IR: 3295.8, 2869.7, 1687.7, 1532.4, fro 1454.2, 1366.6, 1309.5 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.29$  (s, 1H, NH), 7.52 m (m, 2H, Ar-H), 7.37 (d, 2H, J=8.2Hz, Ar-H), 7.29 (m, 4H, 7.4Hz, Ar-H), 7.16 (m, 2H, Arthe H), 5.08 (m, 1H, CH), 4.53 (s, 1H, NH), 3.86 (m, 2H, CH<sub>2</sub>), 3.38 (s, 2H, CH<sub>2</sub>), 1.41 (s, 9H, do C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 155.8, 153.8, 138.7, 128.6, 127.9, 78.8, 72.4, cii 71.4, 49.7, 28.7. LC-MS: (ESI) *m/z* calculated for C<sub>21</sub>H<sub>26</sub>N<sub>3</sub>O<sub>3</sub> [M+H]<sup>+</sup> 368.20, observed me 368.20. nt

or *Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-2-(benzylthio)ethylcarbamate (45g)*, White <sup>use</sup> solid, 1.0 g, 98 %;  $R_f$  0.45 (1:1/hexanes:EtOAc); IR: 3316.1, 2980.9, 1672.6, 1515.4, <sup>thi</sup> 1441.1cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.32$  (s, 1H, NH), 7.58 (d, 1H, *J*=7.4Hz, <sup>s</sup> Ar-H), 7.48 (m, 2H, Ar-H), 7.44 (m, 4H, Ar-H), 7.22 (m, 1H, Ar-H), 7.18 (m, 2H, Ar-H), <sup>spa</sup> 5.01 (q, 1H, *J*=6.6Hz, CH), 3.75 (s, 2H, CH<sub>2</sub>), 3.11 (dd, 1H, *J*=6.2, 6.1Hz, CH<sub>2</sub>), 2.85 (dd, <sup>ce</sup> 1H, J=8.3, 8.3Hz, CH<sub>2</sub>), 1.43 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  155.8, <sup>to</sup> 154.7, 143.4, 138.8, 134.6,129.4, 128.8, 127.3, 122.5, 121.7, 119.1, 111.9, 78.9, 49.4, 35.7, em

28

ph

asi

der

's

ion

t.]

att 35.4, 28.7. LC-MS: (ESI) m/z calculated for C<sub>21</sub>H<sub>26</sub>N<sub>3</sub>O<sub>2</sub>S [M+H]<sup>+</sup> 384.17, observed ent 384.20.

wit *Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-3-(benzyloxy)butylcarbamate (45h)*, White h a solid, 0.47 g, 94%; R<sub>f</sub> 0.40 (1:1/hexanes:EtOAc); IR: 3324.3, 2977.1, 1678.7, 1528.7cm<sup>-</sup> gre <sup>1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.25$  (s, 1H, NH), 7.49 (m, 2H, Ar-H), 7.30 (d, 1H, at *J*=9.4Hz, Ar-H), 7.18 (m, 6H, Ar-H), 7.03 (d, 1H, *J*=8.9Hz, Ar-H), 5.02 (m, 1H, CH), 4.94 qu (m, 1H, CH), 4.47 (m, 2H, CH<sub>2</sub>), 4.01 (m, 1H, NH), 3.37 (m, 2H, CH<sub>2</sub>), 1.34 (s, 9H, ote C(CH<sub>3</sub>)<sub>3</sub>), 1.13 (d, 3H, *J*=5.6Hz, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  155.9, 154.1, fro 139.1, 138.9, 128.5, 127.9, 127.8, 127.7, 79.6, 78.9, 76.5, 76.4, 70.7, 70.5, 53.4, 28.6, 17.0, m 16.6. **LC-MS**: (ESI) *m/z* calculated for C<sub>22</sub>H<sub>28</sub>N<sub>3</sub>O<sub>3</sub> [M+H]<sup>+</sup> 382.21, observed 382.20.

<sup>the</sup> *Compound (45i)*, White solid, 1.00 g, 96%;  $R_f 0.29 (1:1/hexanes:EtOAc)$ ; IR: 3317.9, <sup>do</sup> 1693.9, 1677.2, 1530.9, 1439.8, 1393.0, 1364.3cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta =$ <sup>cu</sup> 12.15 (s, 1H, NH), 7.50 (m, 2H, Ar-H), 7.28 (d, 1H, *J*=7.3Hz, Ar-H), 7.14 (m, 2H, Ar-H), <sup>me</sup> 6.79 (s, 1H, NH), 4.73 (m, 1H, CH), 3.37 (s, 1H, NH), 2.90 (m, 2H, CH<sub>2</sub>), 1.84, (m, 2H, <sup>nt</sup> CH<sub>2</sub>), 1.38 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  156.3, 156.0, 155.8, 121.8, <sup>or</sup> 78.6, 77.8, 49.9, 34.1, 29.6, 28.7, 28.7, 23.3. **LC-MS**: (ESI) *m/z* calculated for C<sub>22</sub>H<sub>35</sub>N<sub>4</sub>O<sub>4</sub> <sup>use</sup> [M+H]<sup>+</sup> 419.27, observed 419.30.

thi

STert-butyl2-(phenoxycarbonyl)-1-(1H-benzo[d]imidazol-2-yl)ethylcarbamate(45j),spaWhite solid, 0.45 g, 90%; Rf 0.46 (1:1/hexanes:EtOAc); IR: 3320.1, 2980.3, 1737.4,ce1679.4, 1524.8, 1440.8, 1391.5, 1367.3, 1337.2cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta =$ to12.30 (s, 1H, NH), 7.56 (d, 1H, J=8.00Hz, Ar-H), 7.47 (m, 2H, Ar-H), 7.31 (m, 5H, Ar-emH), 7.16 (m, 2H, Ar-H), 5.21 (q, 1H, J=7.4Hz, CH), 5.10 (s, 2H, CH<sub>2</sub>), 3.40 (s, 1H, NH),ph29

ze

t.]

der

's

att 3.22 (m, 1H, CH<sub>2</sub>), 2.95 (m, 1H, CH<sub>2</sub>), 1.41 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 ent MHz) δ 170.6, 155.5, 154.6, 136.6, 134.9, 128.8, 128.3, 128.1, 121.6, 119.1, 111.9, 79.7, ion 78.9, 65.9, 46.6, 38.6, 28.7. LC-MS: (ESI) *m/z* calculated for C<sub>22</sub>H<sub>26</sub>N<sub>3</sub>O<sub>4</sub> [M+H]<sup>+</sup> 396.19, wit observed 396.20.
h a

gre Tert-butyl 3-((benzyloxy)carbonyl)-1-(1H-benzo[d]imidazol-2-yl)propylcarbamate (45k), White solid, 0.40 g, 80%; Rf 0.33 (1:1/hexanes:EtOAc); IR: 3312.3, 2977.1, 1737.5, at 1677.3, 1525.8, 1453.8, 1391.4, 1366.9 cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.22$ au ote (s, 1H, NH), 7.57 (dd, 1H, J=6.9, 6.9Hz, Ar-H), 7.46 (dd, 1H, J=6.9, 6.9Hz, Ar-H), 7.39 (m, 5H, Ar-H), 7.15 (m, 2H, Ar-H), 5.08 (s, 2H, CH<sub>2</sub>), 4.85 (m, 1H, CH), 3.38 (s, 1H, NH), fro 2.49 (d, 2H, *J*=12.0Hz, CH<sub>2</sub>), 2.29 (m, 1H, CH<sub>2</sub>), 2.09 (m, 1H, CH<sub>2</sub>), 1.40 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); m <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 172.7, 155.8, 155.5, 143.4, 136.6, 134.7, 128.9, 128.4, the 128.4, 122.3, 121.5, 119.0, 111.8, 79.7, 78.7, 65.9, 49.0, 30.6, 29.3, 28.7. LC-MS: (ESI) do m/z calculated for C<sub>23</sub>H<sub>28</sub>N<sub>3</sub>O<sub>4</sub> [M+H]<sup>+</sup> 410.21, observed 410.20. cu

me Tert-butvl 1-(1H-benzo/d/imidazol-2-vl)-2-carbamovlethylcarbamate (451), White solid, nt 0.48 g, 96%; Rf 0.41 (1:1/hexanes:EtOAc); IR: 3296.2, 2983.9, 1728.0, 1676.4, 1526.9, or 1446.2, 1394.7, 1371.1, 1303.5cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.43$  (s, 1H, use NH), 8.32 (d, 1H, J=1.2Hz, NH), 7.79 (d, 1H, J=8.3Hz, Ar-H), 7.60 (dd, 1H, J=7.2, 7.4Hz, thi Ar-H), 7.48 (dd, 1H, J=7.2, 7.4Hz, Ar-H), 7.19 (t, 1H, J=7.6Hz, Ar-H), 5.19 (m, 1H, CH), S 3.39 (s, 2H, NH<sub>2</sub>), 3.13 (m, 1H, CH<sub>2</sub>), 1.44 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 spa MHz) & 155.5, 152.9, 143.1, 134.9, 122.8, 121.8, 119.2, 118.8, 112.0, 79.6, 79.3, 46.5, ce 28.6, 22.5. LC-MS: (ESI) *m/z* calculated for C<sub>15</sub>H<sub>21</sub>N<sub>4</sub>O<sub>3</sub> [M-NH<sup>+</sup><sub>4</sub>] 305.16, observed to 287.20.

30

em

ph

ze

preat quote from the document or the document of the document

der

's

att 1,2,3,4-Tetrahydro-1-oxo-pyrido[1,2a]benzimidazole (45m), White solid, 0.48 g, 96%; Rf ent 0.53 (1:1/hexanes:EtOAc); IR: 3376.1, 2973.6, 1738.1, 1682.0, 1610.6, 1548.1, 1514.3, ion 1445.9, 1356.8, 1330.4, 1303.4cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 8.14$  (m, 1H, Arwit H), 7.71 (m, 1H, Ar-H), 7.56 (d, 1H, J=8.5Hz, Ar-H), 7.39 (m, 2H, Ar-H), 5.12 (m, 1H, h a CH), 3.04 (m, 1H, NH), 2.88 (m, 1H, CH), 2.17 (m, 2H, CH<sub>2</sub>), 1.45 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 0.92 gre (s, 6H, C(CH<sub>3</sub>)<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 169.6, 155.6, 155.5, 142.7, 131.6, at 125.4, 125.3, 119.9, 115.2, 78.9, 45.9, 31.8, 28.7, 27.6. LC-MS: (ESI) m/z calculated for qu  $C_{16}H_{19}N_{3}O_{3}$  [M+H]<sup>+</sup> 302.15, observed 334.20. ote

fro *Tert-butyl 1-(1H-benzo[d]imidazol-2-yl)-2-phenylethylcarbamate (45n)*, White solid, 1.0 m g, 99%; R<sub>f</sub> 0.40 (1:1/hexanes:EtOAc); IR: 3307.6, 1681.9, 1531.8, 1457.6, 1433.8, 1367.1, the 1335.5, 1311.8cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.22$  (s, 1H, NH), 7.51 (m, 1H, do Ar-H), 7.40 (d, 1H, *J*=8.3Hz, Ar-H), 7.20 (m, 7H, Ar-H), 4.99 (m, 1H, CH), 3.36 (s, 1H, cu NH), 3.07 (t, 1H, *J*=11.6Hz, CH<sub>2</sub>), 1.21 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) me δ 155.8, 155.6, 138.6, 138.9, 129.7, 128.5, 126.7, 78.5, 51.3, 28.6. LC-MS: (ESI) *m/z* nt calculated for C<sub>20</sub>H<sub>24</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 338.19, observed 338.20.

#### or Tert-butyl 2-(4-(benzyloxy)phenyl)-1-(5-chloro-1H-benzo[d]imidazol-2-yl)

<sup>use</sup> *ethylcarbamate (46)*, White solid, 0.45 g, 90%;  $R_f$  0.47 (7:3/hexanes:EtOAc); IR: 3318.2, <sup>thi</sup> 2930.1, 1675.5, 1607.9, 1510.4, 1445.3, 1371.1cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta =$ 

<sup>s</sup> 12.41 (s, 1H, NH), 8.15 (s, 1H, Ar-H), 7.82 (m, 1H, Ar-H), 7.71 (m, 1H, Ar-H), 7.48 (m, <sup>spa</sup> 9H, Ar-H), 7.11 (m, 4H, Ar-H), 5.05 (m, 1H, CH), 3.37 (s, 2H, CH<sub>2</sub>), 3.27 (m, 1H, CH<sub>2</sub>),
<sup>ce</sup> 3.01 (m, 1H, CH<sub>2</sub>), 1.31 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 157.4, 137.7, to

31

em

ph

asi

t.]

der

's

att

ion

130.7, 128.9, 128.2, 128.1, 118.4, 114.9, 78.6, 69.6, 28.6. LC-MS: (ESI) m/z calculated ent for C<sub>27</sub>H<sub>29</sub>ClN<sub>3</sub>O<sub>3</sub> [M+H]<sup>+</sup> 478.19, observed 478.20.

wit Tert-butyl 2-(4-(benzyloxy)phenyl)-1-(5-bromo-1H-benzo/d]imidazol-2h ayl)ethylcarbamate (47), White solid, 0.46 g, 92%; Rf 0.47 (7:3/hexanes:EtOAc); IR: gre 3314.6, 2914.3, 1675.6, 1613.5, 1512.9, 1443.4, 1371.9cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz): δ = 12.44 (s, 1H, NH), 7.77 (s, 1H, Ar-H), 7.60 (d, *J*=8.3Hz, 1H, Ar-H), 7.48 (m, at 7H, Ar-H), 7.15 (dd, J=8.1, 8.2Hz, 2H, Ar-H), 6.89 (dd, J=8.1, 8.2Hz, 2H, Ar-H), 5.05 au (m, 1H, CH), 3.37 (s, 2H, CH<sub>2</sub>), 3.27 (m, 1H, CH<sub>2</sub>), 3.01 (m, 1H, CH<sub>2</sub>), 1.31 (s, 9H, ote C(CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 157.4, 155.7, 144.9, 137.7, 130.7, 133.7, fro 129.4, 128.9, 128.2, 128.1, 124.9, 121.4, 120.7, 114.9, 114.5, 78.6, 69.6, 51.5, 38.9, 28.6. m **LC-MS**: (ESI) m/z calculated for C<sub>27</sub>H<sub>29</sub>BrN<sub>3</sub>O<sub>3</sub> [M+H]<sup>+</sup> 522.14, observed 522.10. the

do Benzyl 1-(1H-benzo/d/imidazol-2-yl)-2-carbamoylethylcarbamate (48), White solid, 0.41 cu g, 82%; R<sub>f</sub> 0.40 (1:1/hexanes:EtOAc); IR: 3303.1, 1693.3, 1528.7, 1440.3, 1328.9cm<sup>-1</sup>; <sup>1</sup>H me NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.52$  (s, 1H, NH), 8.31 (d, 1H, *J*=9.2Hz, NH), 7.61 (dd, nt 1H, J=7.6, 7.6Hz, Ar-H), 7.49 (dd, 1H, J=7.6, 7.6Hz, Ar-H), 7.33 (m, 5H, Ar-H), 7.23 (m, or 1H, Ar-H), 7.18(m, 2H, Ar-H), 5.27 (g, 1H, J=7.4Hz, CH), 5.13 (g, 2H, J=12.7Hz, CH<sub>2</sub>), use 3.38 (s, 4H, NH<sub>2</sub>, CH<sub>2</sub>), 3.33 (d, 1H, J=5.0Hz, CH), 3.18 (m, 1H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, thi 100 MHz) & 156.3, 152.6, 143.0, 137.2, 135.0, 128.9, 128.4, 128.3, 122.9, 121.9, 119.2, S 118.8, 112.0, 66.4, 47.0, 22.4. LC-MS: (ESI) *m/z* calculated for C<sub>18</sub>H<sub>19</sub>N<sub>4</sub>O<sub>3</sub> [M+H]<sup>+</sup> spa 339.15, observed 321.10.

32

ce

to

em

ph

ze

preat quote from the document or the document of the document of the document the document of the document of

der

's

att **Benzyl 1-(1H-benzo[d]imidazol-2-yl)-2-phenylethylcarbamate (49)**, White solid, 1.0 g, ent 98%; R<sub>f</sub> 0.54 (1:1/hexanes: EtOAc); IR: 3303.2, 1685.8, 1524.9, 1454.9, 1429.8, ion 1335.8cm<sup>-1</sup>; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 12.33$  (s, 1H, NH), 7.99 (d, 1H, *J*=8.2Hz, wit Ar-H), 7.53 (m, 2H, Ar-H), 7.25 (m, 13H, Ar-H), 5.00 (q, 2H, *J*=12.7Hz, CH), 3.40 (s, 4H, h a CH<sub>2</sub>), 3.11 (t, 1H, *J*=12.0Hz, NH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  156.3, 155.5, 138.6, gre 137.5, 129.7, 128.8, 128.6, 128.1, 127.9, 126.8, 65.7, 51.9. **LC-MS**: (ESI) *m/z* calculated at for C<sub>23</sub>H<sub>22</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 372.17, observed 372.20. qu

ote *Benzyl 1-(1H-benzo[d]imidazol-2-yl)-3-methylbutylcarbamate (50)*, White solid, 0.45 g, fro 90%; R/ 0.40 (1:1/hexanes: EtOAc); IR: 3255.3, 1693.1, 1530.7, 1326.9cm<sup>-1</sup>; <sup>1</sup>H NMR m (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 12.25 (s, 1H, NH), 7.83 (d, 1H, *J*=8.2Hz, Ar-H), 7.41 (m, 6H, the Ar-H), 7.15 (m, 2H, Ar-H), 5.06 (q, 2H, *J*=12.6Hz, CH<sub>2</sub>), 4.89 (q, 1H, *J*=7.7Hz, NH), 3.37 do (s, 2H, CH<sub>2</sub>), 1.79 (m, 1H, CH), 1.63 (m, 1H, CH), 0.92 (s, 6H, C(CH<sub>3</sub>)<sub>2</sub>); <sup>13</sup>C NMR cu (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  156.5, 156.4, 137.5, 128.8, 128.3, 128.2, 66.0, 48.6, 43.1, 24.8, me 23.2, 22.2. **LC-MS**: (ESI) *m/z* calculated for C<sub>20</sub>H<sub>24</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 338.19, observed 338.20.

<sup>nt</sup> 2,6-di(1H-benzo[d]imidazol-2-yl)pyridin-4(1H)-one (79a), a white solid, 0.16 g, 92 %; <sup>1</sup>H <sup>or</sup> NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 8.09$  (s, 1H, Ar-H), 7.88 (d, J=6.2 Hz, 2H, Ar-H), 7.56 <sup>use</sup> (d, J=6.2 Hz, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta = 167.4$ , 148.0, 134.0,126.2, <sup>thi</sup> 115.2, 112.7 . **HRMS** (ESI) *m/z* calculated for C<sub>19</sub>H<sub>14</sub>N<sub>5</sub>O [M+H]<sup>+</sup> 328.1193, observed <sup>s</sup> 328.1207.

spa

2,6-bis(5-fluoro-1H-benzo[d]imidazol-2-yl)pyridin-4(1H)-one (79b), a white solid, 0.20
g, 93 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 7.91 (s, 1H, Ar-H), 7.82 (s, 1H, Ar-H), 7.65
(d, J=8.6 Hz, 1H, Ar-H), 7.52 (d, J=8.6 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ
ph

ze

t.]

= 167.2, 149.2, 145.6, 136.1, 133.9, 128.5, 117.7, 116.9, 112.3. **HRMS** (ESI) m/zent calculated for  $C_{19}H_{12}F_2N_5O [M+H]^+$  364.1004, observed 364.1012. ion

wit

der

's

att

h a 2,6-bis(5-chloro-1H-benzo[d]imidazol-2-yl)pyridin-4(1H)-one (79c), a white solid, 0.15 gre g, 89 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 8.24 (s, 1H, Ar-H), 7.88 (d, *J*=8.2 Hz, 1H, Ar-H), 7.80 (s, 1H, Ar-H), 7.69 (d, *J*=8.2 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) at  $\delta = 162.4, 150.2, 150.0, 149.4, 144.7, 140.6, 136.2, 134.1, 129.7, 129.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 124.6, 125.8, 125.8, 124.6, 125.8, 125.8, 125.8, 124.6, 125.8, 125$ au ote 116.8, 114.9. HRMS (ESI) m/z calculated for C19H12Cl2N5O [M+H]+ 396.0413, observed fro 396.0420.

m 2,6-bis(5-bromo-1H-benzo/d/imidazol-2-vl)pyridin-4(1H)-one (79d), a white solid, 0.10 the g, 90 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 7.91 (s, 1H, Ar-H), 7.82 (s, 1H, Ar-H), 7.65 do (d, J=8.6 Hz, 1H, Ar-H), 7.52 (d, J=8.6 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ cu = 167.2, 149.2, 145.6, 136.1, 133.9, 128.5, 117.7, 116.9, 112.3. HRMS (ESI) m/z me calculated for C<sub>19</sub>H<sub>12</sub>Br<sub>2</sub>N<sub>5</sub>O [M+H]<sup>+</sup> 483.9403, observed 483.9420, 487.9384.

*Compound (79e)*, a white solid, 0.13 g, 85 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 8.24$  (s, or 1H, Ar-H), 7.88 ( d, J=8.2 Hz, 1H, Ar-H), 7.80 (s, 1H, Ar-H), 7.69 (d, J=8.2 Hz, 1H, Aruse H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  = 166.8, 153.2, 148.2, 127.1, 120.1, 110.9, 105.5. thi **HRMS** (ESI) m/z calculated for C<sub>21</sub>H<sub>12</sub>N<sub>7</sub>O [M+H]<sup>+</sup> 378.1098, observed 378.1096.

S

nt

<sup>spa</sup> Compound (79f), a white solid, 0.2 g, 90 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 8.24$  (s, ce 1H, Ar-H), 7.88 ( d, J=8.2 Hz, 1H, Ar-H), 7.80 (s, 1H, Ar-H), 7.69 (d, J=8.2 Hz, 1H, Arto H), 3.87 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  = 167.2, 166.1, 150.4, 150.0, em

34

ph

asi

t.]

der

's

att 145.6, 127.9, 126.4, 126.1, 125.8, 116.8, 115.2, 112.5, 112.4, 52.7, 52.7. **HRMS** (ESI) m/z ent calculated for C<sub>23</sub>H<sub>18</sub>N<sub>5</sub>O<sub>5</sub> [M+H]<sup>+</sup> 444.1302, observed 444.1312. ion

wit 2-(6-(1H-benzo[d]imidazol-2-yl)pyridin-2-yl)-1H-benzo[d]imidazole (80a), a white solid, h a 0.2 g, 95 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 8.77$  ( d, J=7.8, Hz, 1H, Ar-H), 8.38 (t, gre J=7.8, 7.8 Hz, 1H, Ar-H), 7.85 (m, 2H, Ar-H), 7.56 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, at 100MHz)  $\delta = 165.3$ , 148.8, 147.4, 143.2, 140.6, 133.3, 127.5, 126.8, 126.6, 125.5, 115.2. gre HRMS (ESI) *m/z* calculated for C<sub>19</sub>H<sub>14</sub>N<sub>5</sub> [M+H]<sup>+</sup> 312.1244, observed 312.1257.

<sup>ote</sup> *5-fluoro-2-(6-(5-fluoro-1H-benzo[d]imidazol-2-yl)pyridin-2-yl)-1H-benzo[d]imidazole* <sup>fro</sup> *(80b)*, a white solid, 0.18 g, 84 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 8.54$  (d, *J*=7.9 Hz, <sup>m</sup> 1H, Ar-H), 8.35 (t, *J*=7.8, 8.0 Hz, 1H, Ar-H), 7.81 (dd, *J*=2.3, 4.6 Hz, 1H, Ar-H), 7.56 (d, <sup>the</sup> *J*=8.7 Hz, 1H, Ar-H), 7.34 (t, *J*=9.4, 9.2 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$ <sup>do</sup> = 161.5,159.1, 149.7, 145.0, 140.6, 124.2, 116.8, 114.2, 113.9, 101.6, 101.4. **HRMS** (ESI) <sup>cu</sup> *m/z* calculated for C<sub>19</sub>H<sub>12</sub>F<sub>2</sub>N<sub>5</sub> [M+H]<sup>+</sup> 348.1055, observed 348.1056.

me

nt 5-chloro-2-(6-(5-chloro-1H-benzo[d]imidazol-2-yl)pyridin-2-yl)-1H-benzo[d]imidazole or (80c), a white solid, 0.17 g, 89 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 11.00$  (s, 1H, NH), use 8.54 (d, J=7.9 Hz, 1H, Ar-H), 8.35 (t, J=7.8, 8.0 Hz, 1H, Ar-H), 7.81 (dd, J=2.3, 4.6 Hz, 1H, Ar-H), 7.56 (d, J=8.7 Hz, 1H, Ar-H), 7.34 (t, J=9.4, 9.2 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta = 161.5,159.1, 149.7, 145.0, 140.6, 124.2, 116.8, 114.2, 113.9,$ spa 101.6, 101.4. HRMS (ESI)*m/z*calculated for C<sub>19</sub>H<sub>12</sub>Cl<sub>2</sub>N<sub>5</sub> [M+H]<sup>+</sup> 380.0464, observed380.0464.

ce

to 5-bromo-2-(6-(5-bromo-1H-benzo[d]imidazol-2-yl)pyridin-2-yl)-1H-benzo[d]imidazole em (80d), a white solid, 0.18 g, 88 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 8.54 (d, J=7.9 Hz, ph

ze

t.]

der

's

att

ent

ion

wit

m/z calculated for C<sub>19</sub>H<sub>12</sub>Br<sub>2</sub>N<sub>5</sub> [M+H]<sup>+</sup> 467.9454, observed 467.9480, 471.9442. h a gre *Compound (80e)*, a white solid, 0.14 g, 88 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 8.54 (d, at *J*=7.9 Hz, 1H, Ar-H), 8.35 (t, *J*=7.8, 8.0 Hz, 1H, Ar-H), 7.81 (dd, *J*=2.3, 4.6 Hz, 1H, Arqu H), 7.56 (d, *J*=8.7 Hz, 1H, Ar-H), 7.34 (t, *J*=9.4, 9.2 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, ote 100MHz)  $\delta$  = 161.5,159.1, 149.7, 145.0, 140.6, 124.2, 116.8, 114.2, 113.9, 101.6, 101.4. fro **HRMS** (ESI) *m/z* calculated for C<sub>21</sub>H<sub>12</sub>N<sub>7</sub> [M+H]<sup>+</sup> 362.1149, observed 362.1150.

1H, Ar-H), 8.35 (t, J=7.8, 8.0 Hz, 1H, Ar-H), 7.81 (dd, J=2.3, 4.6 Hz, 1H, Ar-H), 7.56 (d,

J=8.7 Hz, 1H, Ar-H), 7.34 (t, J=9.4, 9.2 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ

= 161.5,159.1, 149.7, 145.0, 140.6, 124.2, 116.8, 114.2, 113.9, 101.6, 101.4. **HRMS** (ESI)

<sup>m</sup> *Compound (80f)*, a white solid, 0.12 g, 91 %; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 8.77$  ( d, the *J*=7.8, Hz, 1H, Ar-H), 8.38 (t, *J*=7.8, 7.8 Hz, 1H, Ar-H), 7.85 (m, 2H, Ar-H), 7.56 (m, 2H, do Ar-H), 3.92 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta = 168.1, 167.1, 152.9, 147.8,$ <sup>cu</sup> 147.7, 1140.0, 125.9, 122.9, 122.9, 52.0. **HRMS** (ESI) *m/z* calculated for C<sub>23</sub>H<sub>18</sub>N<sub>5</sub>O<sub>4</sub> <sup>me</sup> [M+H]<sup>+</sup> 428.1353, observed 428.1363.

nt

or 3-(1H-benzo[d]imidazol-2-yl)phenol (84a), White solid, 0.85 g, 92%;  $R_f$  0.40 use (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.85 (s, 1H, NH), 9.78 (s, 1H, thi OH), 7.61 (m, 4H, Ar-H), 7.33 (m, 3H, Ar-H), 6.9 (d, *J*=7.3 Hz, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  158.2, 151.8, 131.9, 130.5, 117.7, 117.4, 113.8. **HRMS**: (ESI) *m/z* spa calculated for C<sub>13</sub>H<sub>11</sub>N<sub>2</sub>O [M+H]<sup>+</sup> 211.0866, observed 211.0867.

<sup>ce</sup> 2-(3-bromophenyl)-1H-benzo[d]imidazole (84b), White solid, 1.2 g, 95%; R<sub>f</sub> 0.61 to (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.06 (s, 1H, NH), 8.39 (s, 1H, em Ar-H), 8.20 (d, *J*=7.8 Hz, 1H, Ar-H), 7.55 (m, 4H, Ar-H), 7.23 (m, 2H, Ar-H); <sup>13</sup>C NMR ph

ze

t.]

der

's

att (DMSO-d<sub>6</sub>, 100MHz) δ 150.1, 132.9, 132.9, 131.7, 129.4, 125.8, 122.7. **HRMS**: (ESI) *m/z* ent calculated for C<sub>13</sub>H<sub>11</sub>BrN<sub>2</sub> [M+H]<sup>+</sup> 273.0022, observed 273.0033, 275.0015. ion

wit 3-(1H-benzo[d]imidazol-2-yl)benzonitrile (84c), White solid, 2.1 g, 96%; R<sub>f</sub> 0.49 h a (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.15 (s, 1H, NH), 8.56 (s, 1H, gre Ar-H), 8.50 (d, *J*=7.8 Hz, 1H, Ar-H), 7.96 (d, *J*=7.6 Hz, 1H, Ar-H), 7.70 (m, 3H, Ar-H), at 7.25 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  149.7, 133.5, 131.8, 131.4, 130.8, qu 130.2, 118.9, 112.6. **HRMS**: (ESI) *m*/*z* calculated for C<sub>14</sub>H<sub>10</sub>N<sub>3</sub> [M+H]<sup>+</sup> 220.0869, ote observed 220.0874.

<sup>fro</sup> *2-m-tolyl-1H-benzo[d]imidazole (84d)*, White solid, 0.8 g, 89%; R<sub>f</sub> 0.51 (7:3/Hexanes:
<sup>m</sup> EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.91 (s, 1H, NH), 8.05 (s, 1H, Ar-H), 8.00
<sup>the</sup> (d, *J*=7.8 Hz, 1H, Ar-H), 7.55 (m, 3H, Ar-H), 7.30 (d, *J*=7.5 Hz, 1H, Ar-H), 7.20 (m, 2H,
<sup>do</sup> Ar-H), 2.42 (s, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 151.8, 138.6, 130.9, 130.6,
<sup>cu</sup> 129.3, 127.5, 124.1, 21.5. HRMS: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub> [M+H]<sup>+</sup> 209.1073,
<sup>me</sup> observed 209.1084.

nt

or 2-(3-nitrophenyl)-1H-benzo/d/imidazole (84e), White solid, 2.1 g, 82%; Rf 0.34 use (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 13.33$  (s, 1H, NH), 9.02 (s, 1H, thi Ar-H), 8.60 (d, J=7.8 Hz, 1H, Ar-H), 8.30 (d, J=8.1 Hz, 1H, Ar-H), 7.85 (m, 4H, Ar-H), S 7.25 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 149.5, 148.8, 132.9, 132.2, 131.1, spa 124.7, 121.3. **HRMS**: (ESI) m/z calculated for C<sub>13</sub>H<sub>10</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 240.0768, observed ce 240.0768. to em 37 ph

ze

der

's

at

qu

att 2-(3-fluorophenyl)-1H-benzo[d]imidazole (84f), White solid, 0.21 g, 78%; R<sub>f</sub> 0.45 ent (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 13.05$  (s, 1H, NH), 8.03 (d, ion J=7.9 Hz, 1H, Ar-H), 7.96 (m, 1H, Ar-H), 7.60 (m, 3H, Ar-H), 7.35 (m, 1H, Ar-H), 7.23 wit (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  150.4, 150.4, 133.0, 132.9, 131.7, 131.7, h a 123.0, 122.9. **HRMS**: (ESI) *m/z* calculated for C<sub>13</sub>H<sub>10</sub>FN<sub>2</sub> [M+H]<sup>+</sup> 213.0823, observed gre 213.0824.

t.]

2-(3-chlorophenyl)-1H-benzo[d]imidazole (84g), White solid, 0.21 g, 78%; R<sub>f</sub> 0.45 ote (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 13.06 (s, 1H, NH), 8.24 (s, 1H, fro Ar-H), 8.15 (d, *J*=7.2 Hz, 1H, Ar-H), 7.60 (m, 4H, Ar-H), 7.23 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 150.2, 134.2, 132.7, 131.4, 130.0, 126.5, 125.5, 122.9. HRMS: the (ESI) *m/z* calculated for C<sub>13</sub>H<sub>10</sub>ClN<sub>2</sub> [M+H]<sup>+</sup> 229.0527, observed 229.0535. do

cu 2-(3-(methylthio)phenyl)-1H-benzo[d]imidazole (84h), White solid, 0.32 g, 85%; R<sub>f</sub> 0.47 me (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.98 (s, 1H, NH), 8.07 (s, 1H, nt Ar-H), 7.96 (d, *J*=7.8 Hz, 1H, Ar-H), 7.50 (m, 3H, Ar-H), 7.37 (d, *J*=7.8 Hz, 1H, Ar-H), or 7.21 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  151.2, 139.7, 131.3, 129.9, 127.5, use 123.5, 123.4, 15.1. **HRMS**: (ESI) *m*/*z* calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub>S [M+H]<sup>+</sup> 241.0794, thi observed 241.0786.

<sup>s</sup> 2-(3-methoxyphenyl)-1H-benzo[d]imidazole (84i), White solid, 0.22 g, 94%; R<sub>f</sub> 0.53
<sup>spa</sup> (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.92 (s, 1H, NH), 7.79 (s, 1H, Ar-H), 7.66 (m, 2H, Ar-H), 7.45 (m, 2H, Ar-H), 7.21 (m, 2H, Ar-H), 7.05 (d, J=7.7 Hz, 1H, Ar-H), 3.87 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 160.1, 151.5, 131.9, em

38

ph

asi

t.]

der

's

ion

att 130.6, 119.2, 116.3, 111.9, 55.8. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub>O[M+H]<sup>+</sup> ent 225.1022, observed 225.1032.

wit 2-(3-(trifluoromethyl)phenyl)-1H-benzo[d]imidazole (84j), White solid, 0.22 g, 82%; R<sub>f</sub> h a 0.50 (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.18 (s, 1H, NH), 8.54 gre (s, 1H, Ar-H), 8.50 (d, *J*=7.8 Hz, 1H, Ar-H), 7.80 (m, 4H, Ar-H), 7.24 (m, 2H, Ar-H); <sup>13</sup>C at NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  150.1, 131.6, 130.6, 130.1, 126.7, 126.6, 125.9, 123.3, qu 123.2. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>10</sub>F<sub>3</sub>N<sub>2</sub> [M+H]<sup>+</sup> 263.0791, observed 263.0784.

ote 2-(3-isopropylphenyl)-1H-benzo[d]imidazole (84k), White solid, 0.35 g, 90%; R<sub>f</sub> 0.42 fro (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.92 (s, 1H, NH), 8.11 (s, 1H, <sup>m</sup> Ar-H), 8.01 (d, *J*=7.8 Hz, 1H, Ar-H), 7.50 (m, 3H, Ar-H), 7.34 (d, *J*=7.8 Hz, 1H, Ar-H), the 7.21 (m, 2H, Ar-H), 3.00 (m, 1H, -CH-), 1.27 (s, 6H, (CH<sub>3</sub>)<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  151.9, 149.6, 130.6, 129.4, 128.5, 124.8, 124.5, 49.1, 33.9, 24.3. **HRMS**: (ESI) <sup>cu</sup> *m/z* calculated for C<sub>16</sub>H<sub>17</sub>N<sub>2</sub> [M+H]<sup>+</sup> 237.1386, observed 237.1389. me

nt *Methyl 3-(1H-benzo[d]imidazol-2-yl)benzoate (84l),* White solid, 0.35 g, 90%; R<sub>f</sub> 0.42 or (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.17 (s, 1H, NH), 8.82 (s, 1H, use Ar-H), 8.45 (d, *J*=8.2 Hz, 1H, Ar-H), 8.06 (d, *J*=8.0 Hz, 1H, Ar-H), 7.70 (m, 3H, Ar-H), thi 7.24 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  166.3, 150.6, 131.3, 131.2, 130.9, s 130.7, 130.0, 127.5, 52.8. **HRMS**: (ESI) *m/z* calculated for C<sub>15</sub>H<sub>13</sub>N<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup> 253.0972, spa <sup>observed 253.1028.</sup>

<sup>ce</sup> 2-(3-(*methylsulfonyl*)*phenyl*)-1*H-benzo[d]imidazole (84m)*, White solid, 0.3 g, 96%;  $R_f$ to 0.23 (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.25 (s, 1H, NH), 8.74 em

39

ph

asi

t.]

der

's

att

ent

ion 2H, Ar-H), 3.38 (s, 3H, SO<sub>2</sub>CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 150.0, 144.1, 142.3, wit 135.6, 131.8, 131.5, 130.8, 128.4, 125.2, 123.6, 122.5, 119.6, 112.1, 44.0. HRMS: (ESI) h a m/z calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub>O<sub>2</sub>S [M+H]<sup>+</sup> 273.0692, observed 273.0692. gre 2-(3-(1H-pyrrol-1-yl)phenyl)-1H-benzo/d/imidazole (84n), White solid, 0.4 g, 91%; Rf at 0.45 (7:3/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 13.00$  (s, 1H, NH), 8.34 au ote (s, 1H, Ar-H), 8.06 (d, J=7.7 Hz, 1H, Ar-H), 7.72 (m, 3H, Ar-H), 7.45 (m, 3H, Ar-H), 7.24 fro (m, 2H, Ar-H), 6.35 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 151.1, 140.9, 132.0, 130.9, 123.7, 120.9, 119.5, 117.4, 111.3. HRMS: (ESI) m/z calculated for C<sub>17</sub>H<sub>14</sub>N<sub>3</sub> m the [M+H]<sup>+</sup> 260.1182, observed 260.1193.

(s, 1H, Ar-H), 8.51 (d, J=7.9 Hz, 1H, Ar-H), 8.05 (d, J= 8.0 Hz, 1H, Ar-H), 7.86 (t, J=7.8,

7.8 Hz, 1H, Ar-H), 7.72 (d, J=7.6 Hz, 1H, Ar-H), 7.58 (d, J=7.4 Hz, 1H, Ar-H), 7.25 (m,

<sup>do</sup> 3.1.3 General procedure for the synthesis of 64a – 64r, 66a – 66e, and 68a – 68k <sup>cu</sup> benzimidazole analogs.

me

The amide substrates were semi-synthetically prepared using carbodiimide-based coupling nt conditions. The amide substrate was dissolved in 30 mL of toluene or DMF, and added or N,N-diisopropylethylamine (1.0 equiv), HBTU (1.0 equiv), and heated to reflux for 6 use hours. The reaction was cooled to room temperature. The solvent was removed in vacuo in thi the case of toluene, but for DMF, the reaction mixture was diluted with water and products S were extracted using ethyl acetate (EtOAc). The organic layer was dried over anhydrous spa sodium sulfate, filtered and concentrated *in vacuo*. The crude product was purified using ce column chromatography using hexanes/EtOAc in an increasing polarity up to 1:1 mixture. to The impure fractions containing the desired product were concentrated and crystallized in em

40

ph

asi

t.]

der

's

att

hexanes/EtOAc 1:1 to yield the product as a white solid. All the products were ent characterized by 1D (<sup>1</sup>H and <sup>13</sup>C) NMR, and HRMS. ion

wit 2-phenyl-1H-benzo[d]imidazole (64a), White solid, 0.23 g, 96%; R<sub>f</sub> 0.61 (1:1/Hexanes: h a EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.94 (s, 1H, NH), 8.20 (d, 2H, *J*=7.4Hz, Argre H), 7.54 (m, 5H, Ar-H), 7.21 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 151.7, at 130.7, 130.3, 129.4, 126.9. **HRMS**: (ESI) *m/z* calculated for C<sub>13</sub>H<sub>11</sub>N<sub>2</sub> [M+H]<sup>+</sup> 195.0917, ou observed 195.0918.

<sup>ote</sup> 2-*p*-tolyl-1H-benzo[d]imidazole (64b), White solid, 0.21 g, 94%; R<sub>f</sub> 0.56 (1:1/Hexanes: fro EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.84$  (s, 1H, NH), 8.08 (d, 2H, *J*=8.1 Hz, <sup>m</sup> Ar-H), 7.59 (m, 2H, Ar-H), 7.35 (d, 2H, *J*=8.1 Hz, Ar-H), 7.20 (m, 2H, Ar-H), 2.38 (s, 3H, <sup>the</sup> CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  151.9, 140.0, 129.9, 127.9, 126.9, 21.4. **HRMS**: <sup>do</sup> (ESI) *m/z* calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub> [M+H]<sup>+</sup> 209.1073, observed 209.1074.

cu

me 2-(4-ethylphenyl)-1H-benzo[d]imidazole (64c), White solid, 0.23 g, 88%;  $R_f$  0.26 nt (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.85 (s, 1H, NH), 8.11 (d, or 2H, J=8.0 Hz, Ar-H), 7.59 (m, 2H, Ar-H), 7.38 (d, 2H, J=8.0 Hz, Ar-H), 7.20 (m, 2H, Aruse H), 2.65 (q, 2H, J=12, 4 Hz CH<sub>2</sub>), 1.22 (t, 3H, J=12, 4 Hz, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, thi 100MHz)  $\delta$  151.9, 146.2, 128.8, 128.2, 126.9, 28.5, 15.8. **HRMS**: (ESI) *m/z* calculated for  $C_{15}H_{15}N_{2}$  [M+H]<sup>+</sup> 223.1230, observed 223.1238.

<sup>spa</sup> 2-(3,4-dimethoxyphenyl)-1H-benzo[d]imidazole (64d), White solid, 0.32 g, 86%; R<sub>f</sub> 0.57
<sup>ce</sup> (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.69 (s, 1H, NH), 7.76 (m,
<sup>to</sup> 2H, Ar-H), 7.57 (m, Ar-H), 7.17 (m, 3H, Ar-H), 3.88 (s, 3H, OCH<sub>3</sub>), 3.84 (s, 3H, OCH<sub>3</sub>);
<sup>em</sup> <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 152.0, 150.7, 149.4, 123.3, 122.2, 119.7, 112.3, 110.2,
<sup>h</sup>

asi

t.]

der

's

ion

att 56.1, 56.0. **HRMS**: (ESI) m/z calculated for C<sub>15</sub>H<sub>15</sub>N<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup> 255.1128, observed ent 255.1129.

wit 2-(1H-benzo[d]imidazol-2-yl)-5-methoxyphenol (64e), White solid, 0.12g, 47%; R<sub>f</sub> 0.23 h a (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.39 (s, 1H, NH), 7.96 (d, gre 1H, J=8.6 Hz, Ar-H), 7.60 (m, 2H, Ar-H), 7.25 (m, 2H, Ar-H), 6.62 (m, 2H, Ar-H), 3.81 at (s, 3H, OCH<sub>3</sub>), 3.37 (s, 1H, OH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  162.7, 160.4, 152.6, qu 127.7, 106.9, 106.2, 101.9, 55.8. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup> ote 241.0972, observed 241.0988.

<sup>fro</sup> 2-(3,4,5-trimethoxyphenyl)-1H-benzo[d]imidazole (64f), White solid, 0.11g, 82%; R<sub>f</sub> 0.27
<sup>m</sup> (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.86 (s, 1H, NH), 7.67 (d, the 1H, *J*=7.5 Hz, Ar-H), 7.54 (m, 2H, Ar-H), 7.21 (m, 2H, Ar-H), 3.91 (s, 6H, OCH<sub>3</sub>), 3.38
<sup>do</sup> (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 153.7, 151.7, 144.2, 139.4, 135.4, 125.9,
<sup>cu</sup> 122.9, 122.1, 119.2, 111.6, 104.3, 60.6, 56.5. HRMS: (ESI) *m/z* calculated for C<sub>16</sub>H<sub>17</sub>N<sub>2</sub>O<sub>3</sub>
<sup>me</sup> [M+H]<sup>+</sup> 285.1234, observed 285.1236.

nt

or 2-(4-fluorophenyl)-1H-benzo[d]imidazole (64g), White solid, 0.15 g, 80%;  $R_f$  0.34 use (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.94 (s, 1H, NH), 8.25 (m, 2H, Ar-H), 7.65 (m, 2H, Ar-H), 7.42 (m, 2H, Ar-H), 7.20 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  164.8, 162.3, 150.7, 144.3, 135.5, 131.8, 130.8, 130.7, 129.2, spa 129.1, 127.3, 127.3, 123.0, 122.2, 119.3, 116.6, 116.4, 116.1, 115.9, 111.8. HRMS: (ESI) ce m/z calculated for C<sub>13</sub>H<sub>10</sub>FN<sub>2</sub> [M+H]<sup>+</sup> 213.0823, observed 213.0829.

to 2-(4-nitrophenyl)-1H-benzo[d]imidazole (64h), White solid, 0.17 g, 79%;  $R_f$  0.61 em (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.29 (s, 1H, NH), 8.42 (m, ph

ze

t.]

der

's

att

h a

4H, Ar-H), 7.72 (d, 1H, *J*=7.6 Hz, Ar-H), 7.60 (d, 1H, *J*=7.6 Hz, Ar-H), 7.23 (m, 2H, Arent H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 149.5, 148.3, 144.3, 136.5, 135.7, 127.8, 124.7, ion 124.0, 122.8, 119.9, 112.3. **LC-MS**: (ESI) *m/z* calculated for C<sub>13</sub>H<sub>10</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> wit 240.0768, observed 240.0771.

gre 2-(4-bromophenyl)-1H-benzo[d]imidazole (64i), White solid, 0.30 g, 94%; R<sub>f</sub> 0.51 at (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.01 (s, 1H, NH), 8.14 (d, qu 2H, *J*=9.1 Hz, Ar-H), 7.61 (m, 2H, Ar-H), 7.22 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, ote 100MHz) δ 150.7, 132.4, 129.9, 128.8, 123.7, 123.3, 122.4, 119.4, 111.9. HRMS: (ESI) fro *m/z* calculated for C<sub>13</sub>H<sub>10</sub>BrN<sub>2</sub> [M+H]<sup>+</sup> 273.0022, observed 273.0016, 274.9995.

<sup>m</sup> 2-(4-methoxyphenyl)-1H-benzo[d]imidazole (64j), White solid, 0.54 g, 90%;  $R_f$  0.63 the (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.77$  (s, 1H, NH), 8.14 (d, do 2H, *J*=6.8 Hz, Ar-H), 7.57 (m, 2H, Ar-H), 7.18 (m, 2H, Ar-H), 7.11 (d, 2H, J=6.8 Hz, Ar-<sup>cu</sup> H), 3.84 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  161.1, 151.8, 128.5, 123.2, me 114.8, 55.8. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>13</sub>N<sub>2</sub>O[M+H]<sup>+</sup> 225.1022, observed nt 225.1025.

or

use 2-(3,5-dimethoxy-4-methylphenyl)-1H-benzo[d]imidazole (64k), White solid, 0.61 g, 88 thi %; R<sub>f</sub> 0.701 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.90 (s, 1H, NH), 7.66 (d, 2H, J=7.4 Hz, Ar-H), 7.55 (d, 2H, J=7.4 Hz, Ar-H), 7.49 (s, 2H, Ar-H), 7.20 (m, 2H, Ar-H), 3.91 (s, 6H, OCH<sub>3</sub>), 2.07 (s, 3H, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  158.5, to (ESI) *m/z* calculated for C<sub>16</sub>H<sub>17</sub>N<sub>2</sub>O<sub>2</sub>[M+H]<sup>+</sup> 269.1285, observed 269.1298. em

43

ph

asi

preat quote from the document or the document of the document of the document the document of the document of

der

's

att 2-(3-iodo-4-methylphenyl)-1H-benzo[d]imidazole (64l), White solid, 0.71 g, 86 %; R<sub>f</sub> ent 0.67 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.96 (s, 1H, NH), 8.64 ion (s,1H, Ar-H), 8.10 (d, 2H, J=7.9 Hz, Ar-H), 7.66 (d, 1H, J=7.4 Hz, Ar-H), 7.52 (d, 1H, wit J=7.4 Hz, Ar-H), 7.49 (d, 1H, j=7.9 Hz, Ar-H), 7.20 (m, 2H, Ar-H), 2.43 (s, 3H, CH<sub>3</sub>); <sup>13</sup>C h a NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  150.0, 144.1, 143.0, 136.6, 135.4, 130.8, 130.0, 129.9, gre 126.9, 126.7, 123.2, 122.3, 119.3, 111.8, 102.2, 27.9. HRMS: (ESI) *m/z* calculated for at C<sub>14</sub>H<sub>12</sub>IN<sub>2</sub>[M+H]<sup>+</sup> 335.0040, observed 335.0065. qu

ote 2-(3-bromo-4-methylphenyl)-1H-benzo[d]imidazole (64m), White solid, 0.93 g, 82 %;  $R_f$ fro 0.58 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.98 (s, 1H, NH), 8.40 m (s,1H, Ar-H), 8.10 (d, 2H, *J*=7.9 Hz, Ar-H), 7.66 (d, 1H, *J*=7.4 Hz, Ar-H), 7.52 (d, 1H, the J=7.4 Hz, Ar-H), 7.49 (d, 1H, j=7.9 Hz, Ar-H), 7.20 (m, 2H, Ar-H), 2.43 (s, 3H, CH<sub>3</sub>); <sup>13</sup>C do NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  150.0, 144.1, 139.5, 132.1, 130.3, 130.1, 126.0, 125.1, cu 123.2, 122.3, 119.4, 111.9, 22.8. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>12</sub>BrN<sub>2</sub>[M+H]<sup>+</sup> me 287.0178, observed 287.0201, 289.0182.

<sup>nt</sup> 2-(2,3-dimethoxyphenyl)-1H-benzo[d]imidazole (64n), White solid, 0.22 g, 91%;  $R_f$  0.51 <sup>or</sup> (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.23 (s, 1H, NH), 7.85(m,1H, <sup>use</sup> Ar-H), 7.64 (m, 2H, Ar-H), 7.21 (m, 4H, Ar-H), 3.85 (s, 6H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-<sup>thi</sup> d<sub>6</sub>, 100MHz)  $\delta$  152.4, 149.0, 148.8, 119.2, 118.1, 114.4, 112.9, 56.2. **HRMS**: (ESI) *m/z* <sup>s</sup> calculated for C<sub>15</sub>H<sub>15</sub>N<sub>2</sub>O<sub>2</sub>[M+H]<sup>+</sup> 255.1128, observed 255.1144.

spa

ce 2-(6-chloropyridin-2-yl)-1H-benzo[d]imidazole (64o), White solid, 0.16 g, 87%;  $R_f$  0.49 to (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.53 (s, 1H, NH), 7.63 (m, 4H, Ar-H), 7.22 (m, 2H, Ar-H), 6.72 (d, 1H, J=8.0Hz, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 44

ze

preat quote from the document or the document of the document of the document the document of the document of

der

's

att 100MHz) δ 159.1, 151.9, 146.7, 144.5, 141.5, 138.6, 135.1, 125.6, 123.2, 122.1, 120.9, ent 119.5, 112.4, 109.3, 107.3. HRMS: (ESI) *m/z* calculated for C<sub>12</sub>H<sub>9</sub>ClN<sub>3</sub>[M+H]<sup>+</sup> 230.0480, ion observed 230.0499.

h a 2-(3,5-dimethoxyphenyl)-1H-benzo[d]imidazole (64p), White solid, 0.32 g, 84%; R<sub>f</sub> 0.51 gre (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.90$  (s, 1H, NH), 7.44 (m, at 4H, Ar-H), 7.21 (m, 3H, Ar-H), 3.85 (s, 6H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  qu 161.3, 151.5, 132.4, 104.7, 102.5, 56.0. **HRMS**: (ESI) *m/z* calculated for ote C<sub>15</sub>H<sub>15</sub>N<sub>2</sub>O<sub>2</sub>[M+H]<sup>+</sup> 255.1128, observed 255.1128.

fro *4-(1H-benzo[d]imidazol-2-yl)benzene-1,3-diol (64q),* White solid, 0.12g, 47%; R<sub>f</sub> 0.23 <sup>m</sup> (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.39 (s, 1H, NH), 7.96 (d, the 1H, *J*=8.6 Hz, Ar-H), 7.60 (m, 2H, Ar-H), 7.25 (m, 2H, Ar-H), 6.62 (m, 2H, Ar-H), 3.81 <sup>do</sup> (s, 3H, OCH<sub>3</sub>), 3.37 (s, 1H, OH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 162.7, 160.4, 152.6, <sup>cu</sup> 127.7, 106.9, 106.2, 101.9, 55.8. HRMS: (ESI) *m/z* calculated for C<sub>13</sub>H<sub>11</sub>N<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup> <sup>me</sup> 227.0815, observed 227.0827.

nt

*2-(1H-benzo[d]imidazol-2-yl)benzene-1,3-diol (64r)*, White solid, 0.12g, 47%; R<sub>f</sub> 0.23
use (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 13.39 (s, 1H, NH), 7.96 (d,
<sup>1</sup>H, *J*=8.6 Hz, Ar-H), 7.60 (m, 2H, Ar-H), 7.25 (m, 2H, Ar-H), 6.62 (m, 2H, Ar-H), 3.81
(s, 3H, OCH<sub>3</sub>), 3.37 (s, 1H, OH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 162.7, 160.4, 152.6,
<sup>1</sup>27.7, 106.9, 106.2, 101.9, 55.8. LC-MS: (ESI) *m/z* calculated for C<sub>13</sub>H<sub>11</sub>N<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup>
<sup>2</sup>27.0815, observed 227.0827.

to 2-(1H-indol-2-yl)-1H-benzo[d]imidazole (66a), White solid, 0.19g, 96 %;  $R_f$  0.203 em (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.98 (s, 1H, NH), 12.03 (s, ph

ze

t.]

der

's

att 1H, NH), 7.66 (m, 2H, Ar-H), 7.56 (d, *J*=7.4 Hz, 1H, Ar-H), 7.48 (m, 1H, Ar-H), 7.24 (m, ent 4H, Ar-H), 7.05 (m, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 146.6, 144.2, 137.7, ion 135.2, 129.1, 128.3, 123.3, 123.1, 122.2, 121.3, 120.2, 118.9, 112.4, 111.6, 102.1. HRMS: wit (ESI) *m/z* calculated for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub>[M+H]<sup>+</sup> 234.1026, observed 234.1039. h a

gre 2-(1H-indol-3-yl)-1H-benzo[d]imidazole (66b), White solid, 0.2 g, 84%;  $R_f$  0.201 at (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.07 (s, 1H, NH), 12.04 (s, qu 1H, NH), 8.53 (m, 1H, Ar-H), 8.15 ( s, 1H, Ar-H), 7.52 (m, 2H, Ar-H), 7.15 (m, 3H, Arote H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  150.0, 136.9, 126.5, 125.7, 122.7, 121.9, 121.6, fro 120.7, 112.4, 107.1. **HRMS**: (ESI) *m/z* calculated for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub>[M+H]<sup>+</sup> 234.1026, m observed 234.1028.

<sup>the</sup> 2-(1H-indol-5-yl)-1H-benzo[d]imidazole (66c), White solid, 0.19g, 94 %; R<sub>f</sub> 0.302 <sup>do</sup> (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.75$  (s, 1H, NH), 11.37 (s, <sup>cu</sup> 1H, NH), 8.42 (s, 1H, Ar-H), 8.00 (d, 1H, J=8.6 Hz, Ar-H), 7.52 (m, 4H, Ar-H), 7.16 (m, <sup>me</sup> 2H, Ar-H), 6.58 (t, 1H, J=7.4, 2.2 Hz, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  153.6, <sup>nt</sup> 137.3, 128.2, 127.2, 121.9, 121.7, 120.5, 119.2, 112.3, 102.4. **HRMS**: (ESI) *m/z* calculated <sup>or</sup> for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub>[M+H]<sup>+</sup> 234.1026, observed 234.1040.

thi 2-(1H-indol-6-yl)-1H-benzo[d]imidazole (66d), White solid, 0.41 g, 96%;  $R_f$  0.20 s (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.81 (s, 1H, NH), 11.47 (s, spa 1H, NH), 8.11 (d, 2H, J=8.0 Hz, Ar-H), 7.59 (m, 2H, Ar-H), 7.38 (d, 2H, J=8.0 Hz, Ar-H), rea 7.20 (m, 2H, Ar-H), 6.52 (m, 2H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  153.4, 143.5, to 136.3, 135.7, 130.5, 129.5, 127.9, 126.9, 124.6, 123.5, 120.8, 119.9, 119.0, 118.2, 116.9, em ph

ze

asi

use

t.]

der

's

att

ion

116.8, 112.3, 110.4, 101.9, 101.8, 79.7. **HRMS**: (ESI) *m/z* calculated for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub> [M+H]<sup>+</sup> ent 234.1026, observed 234.1026.

wit 2-(1H-indol-7-yl)-1H-benzo[d]imidazole (66e), White solid, 0.5g, 90%;  $R_f$  0.302 h a (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 13.07 (s, 1H, NH), 11.47 (s, gre 1H, NH), 7.95 (d, 1H, *J*=7.4 Hz, Ar-H), 7.76 (m, 2H, Ar-H), 7.55 (m, 2H, Ar-H), 7.24 (m, at 3H, Ar-H), 6.58 (t, 1H, *J*=7.4, 2.2 Hz, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  151.8, qu 133.5, 129.3, 127.2, 123.1, 122.9, 122.1, 119.4, 119.1, 113.3, 111.7, 101.9. HRMS: (ESI) ote *m/z* calculated for C<sub>15</sub>H<sub>12</sub>N<sub>3</sub>[M+H]<sup>+</sup> 234.1026, observed 234.1033.

fro 2-heptadecyl-1H-benzo[d]imidazole (68a), White solid, 1.5 g, 96 %; R<sub>f</sub> 0.62 <sup>m</sup> (9:1/CH<sub>2</sub>CL<sub>2</sub>:MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400MHz): δ = 7.55 (m, 2H, Ar-H), 7.23 (m, 2H, the Ar-H), 2.91 (t, 2H, *J*=15.2, 7.2Hz, CH<sub>2</sub>), 1.84 (m, 2H, CH<sub>2</sub>), 1.26 (s, 28H, CH<sub>2</sub>), 0.89 (t, 3H, *J*=12.6, 5.8Hz, CH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100MHz) δ 155.6, 122.1, 31.9, 29.7, 29.7, 29.7, <sup>cu</sup> 29.7, 29.5, 29.4, 29.4, 29.4, 22.7, 14.1. **HRMS**: (ESI) *m/z* calculated for C<sub>24</sub>H<sub>41</sub>N<sub>2</sub> [M+H]<sup>+</sup> <sup>me</sup> 357.3264, observed 357.3277.

nt

or 2-pentadecyl-1H-benzo[d]imidazole (68b), White solid, 1.2 g, 92 %; R<sub>f</sub> 0.56 use  $(9:1/CH_2CL_2:MeOH)$ ; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400MHz):  $\delta = 7.55$  (m, 2H, Ar-H), 7.23 (m, 2H, Ar-H), 2.91 (t, 2H, *J*=15.2, 7.2Hz, CH<sub>2</sub>), 1.84 (m, 2H, CH<sub>2</sub>), 1.26 (s, 28H, CH<sub>2</sub>), 0.89 (t, 3H, *J*=12.6, 5.8Hz, CH<sub>3</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100MHz)  $\delta$  155.6, 138.3, 122.1, 114.6, 31.9, spa 29.7, 29.7, 29.7, 29.5, 29.4, 29.4, 29.4, 22.7, 14.1. HRMS: (ESI) *m/z* calculated for ce  $C_{22}H_{37}N_2$  [M+H]<sup>+</sup> 329.2951, observed 329.2958.

to 2-heptyl-1H-benzo[d]imidazole (68c), White solid, 0.92 g, 94 %;  $R_f$  0.61 em (9:1/CH<sub>2</sub>CL<sub>2</sub>:MeOH); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.18 (s, 1H, NH), 7.45 (m, 47

asi

t.]

der

's

att 2H, Ar-H), 7.10 (m, 2H, Ar-H), 2.78 (t, 2H, *J*=15.2, 7.2Hz, CH<sub>2</sub>), 1.75 (m, 2H, CH<sub>2</sub>), 1.26 ent (s, 8H, CH<sub>2</sub>), 0.85 (t, 3H, J=12.6, 5.8Hz, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 155.6, ion 122.1, 31.9, 29.7, 29.7, 29.7, 29.5, 29.4, 29.4, 29.4, 22.7, 14.1. LC-MS: (ESI) *m/z* wit calculated for C<sub>14</sub>H<sub>21</sub>N<sub>2</sub> [M+H]<sup>+</sup> 217.1699, observed 217.1698.
h a

gre *Tert-butyl 2-(1H-benzo[d]imidazol-2-yl)ethylcarbamate (68d)*, White solid, 0.36 g, 92 %; at R<sub>f</sub> 0.67 (9:1/CH<sub>2</sub>CL<sub>2</sub>:MeOH); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.24$  (s, 1H, NH), qu 7.45 (m, 2H, Ar-H), 7.10 (m, 3H, Ar-H), 3.38 (m, 2H, CH<sub>2</sub>), 2.94 (t, 2H, *J*=7.6, 14.8Hz, ote CH<sub>2</sub>), 1.38 (s, 9H, CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  162.8, 155.9, 153.2, 134.3, fro 127.7, 121.6, 120.8, 117.1, 112.3, 102.9, 94.3, 78.2, 29.8, 28.7. **HRMS**: (ESI) *m/z* m calculated for C<sub>14</sub>H<sub>20</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 262.1550, observed 262.1552.

<sup>the</sup> 2-(*but-3-enyl*)-1*H-benzo[d]imidazole (68e),* White solid, 0.75 g, 92 %; R<sub>f</sub> 0.44 <sup>do</sup> (9:1/CH<sub>2</sub>CL<sub>2</sub>:MeOH); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.23 (s, 1H, NH), 7.47 (m, <sup>cu</sup> 2H, Ar-H), 7.10 (m, 2H, Ar-H), 5.88 (m, 1H, CH), 5.11 (dd, 1H, *J*=4.0, 17.0Hz, CH), 5.00 <sup>me</sup> (dd, 1H, J=4.0, 17.0Hz, CH), 2.90 (t, 2H, *J*=7.2, 15.2Hz, CH<sub>2</sub>), 2.52 (m, 2H, CH<sub>2</sub>); <sup>13</sup>C <sup>nt</sup> NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  154.9, 137.9, 121.6, 115.9, 32.0, 28.4. **HRMS**: (ESI) *m/z* <sup>or</sup> calculated for C<sub>11</sub>H<sub>13</sub>N<sub>2</sub> [M+H]<sup>+</sup> 173.1073, observed 173.1075.

use

thi **2-(3-phenylpropyl)-1H-benzo[d]imidazole (68f),** White solid, 0.28 g, 95%; R<sub>f</sub> 0.410 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.24$  (s, 2H, NH), 7.45 (m, <sup>spa</sup> 2H, Ar-H), 7.21 (m, 8H, Ar-H), 2.82 (t, *J*=15.0, 7.4Hz, 2H, CH<sub>2</sub>), 2.66 (t, *J*=15.0, 7.4Hz, <sup>ce</sup> 2H, CH<sub>2</sub>), 2.10 (p, *J*=15.0, 7.4Hz, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  155.3, <sup>to</sup> 142.1, 128.9, 128.8, 126.3, 121.5, 35.1, 29.7, 28.5. **HRMS**: (ESI) *m/z* calculated for <sup>em</sup> C<sub>16</sub>H<sub>17</sub>N<sub>2</sub>[M+H]<sup>+</sup> 237.1386, observed 237.1387.

ph

asi

48

t.]

der

's

att **2-(4-nitrobenzyl)-1H-benzo[d]imidazole (68g),** White solid, 0.21 g, 90%; R<sub>f</sub> 0.57 ent (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.42$  (s, 2H, NH), 8.20 (d, ion 2H, *J*=6.8Hz, Ar-H), 7.62 (d, 2H, *J*=6.8Hz, Ar-H), 7.50 (m, 2H, Ar-H), 7.13 (m, 2H, Ar-wit H), 4.3 (s, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  152.7, 146.8, 146.1, 130.7, 124.1, h a 122.0, 35.0. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>12</sub>N<sub>3</sub>O<sub>2</sub>[M+H]<sup>+</sup> 254.0924, observed gre 254.0936.

qu 2-(4-bromobenzyl)-1H-benzo[d]imidazole (68h), White solid, 0.8 g, 92%;  $R_f$  0.64 ote (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.31 (s, 2H, NH), 7.50 (m, fro 4H, Ar-H), 7.21 (m, 4H, Ar-H), 4.17 (s, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  153.5, m 137.5, 131.5, 121.8, 120.2, 34.7. HRMS: (ESI) *m*/*z* calculated for C<sub>14</sub>H<sub>12</sub>BrN<sub>2</sub>[M+H]<sup>+</sup> the 287.0178, observed 287.0177, 289.0158.

<sup>do</sup> 2-(2-bromobenzyl)-1H-benzo[d]imidazole (68i), White solid, 1.2 g, 94%; R<sub>f</sub> 0.51 <sup>cu</sup> (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.30 (s, 2H, NH), 7.64 (d, <sup>me</sup> 1H, *J*=8.0Hz, Ar-H), 7.44 (m, 2H, Ar-H), 7.34 (d, 1H, *J*=8.0Hz, Ar-H), 7.21 (m, 3H, Ar-<sup>nt</sup> H), 4.32 (s, 2H, CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  152.6, 137.3, 133.0, 132.0, 129.3, <sup>or</sup> 128.4, 124.5, 35.8. **HRMS**: (ESI) *m/z* calculated for C<sub>14</sub>H<sub>12</sub>BrN<sub>2</sub> [M+H]<sup>+</sup> 287.0178, <sup>use</sup> observed 287.0182, 289.0165.

thi

s 2-(2-(1H-indol-3-yl)ethyl)-1H-benzo[d]imidazole (68j), White solid, 0.36 g, 93%; R<sub>f</sub> 0.70 spa (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.28 (s, 1H, NH), 10.81 (s, 1H, NH), 7.57 (d, 1H, *J*=7.8 Hz, Ar-H), 7.47 (m, 2H, Ar-H), 7.60 (d, 1H, *J*=7.8 Hz, Ar-H), 7.10 (m, 4H, Ar-H), 6.97 (t, 1H, *J*=7.4, 14.8Hz, Ar-H), 3.21 (m, 4H, CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  155.4, 136.7, 127.5, 122.7, 121.6, 121.4, 118.7, 118.7, 114.0, 49

ze

der

's

ion

t.]

att 111.8, 30.1, 23.9. **HRMS**: (ESI) *m/z* calculated for C<sub>17</sub>H<sub>16</sub>N<sub>3</sub>[M+H]<sup>+</sup> 262.1339, observed ent 262.1339.

wit 2-(2-(1H-indol-3-yl)propyl)-1H-benzo[d]imidazole (68k), White solid, 0.21g, 92%; R<sub>f</sub> h a 0.52 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.25 (s, 1H, NH), 10.83 gre (s, 1H, NH), 7.53 (d, 1H, *J*=7.9 Hz, Ar-H), 7.48 (m, 1H, Ar-H), 7.360 (d, 1H, *J*=8.0 Hz, at Ar-H), 7.10 (m, 4H, Ar-H), 6.97 (t, 1H, *J*=7.4, 14.8Hz, Ar-H), 2.89 (t, 2H, *J*=7.4, 15.2Hz, qu CH<sub>2</sub>), 2.79 (t, 2H, *J*=7.4, 15.2Hz, CH<sub>2</sub>), 2.17 (p, 2H, *J*=7.7, 15.1Hz, CH<sub>2</sub>); <sup>13</sup>C NMR ote (DMSO-d<sub>6</sub>, 100MHz) δ 155.6, 136.8, 127.7, 122.9, 121.5, 121.3, 118.8, 114.4, 111.8, 28.9, fro 28.9, 24.8. **HRMS**: (ESI) *m*/*z* calculated for C<sub>18</sub>H<sub>18</sub>N<sub>3</sub>[M+H]<sup>+</sup> 276.1495, observed m 276.1519.

the 3.1.4 General Procedure for the selective mono-alkylation of 2-indolyl-benzimidazoles do 71a - 71j, 72a - 72f and 73a - 73e.

cu

A solution of 2-indolylbenzimidazole derivative **66a** (1.0 equiv.) in acetonitrile (3 mL) me was cooled to 0 °C in ice/water bath. To the solution was added potassium carbonate nt powder (3 equiv.) and allowed to stir for 20 min. Then, alkyl or benzyl bromide (1 equiv) or was added and the reaction mixture allowed to stir for another 20 minutes at 0 °C before use brought to room temperature. The reaction stirred for another 3 h. Upon the reaction thi completion, (as per TLC analysis), the reaction mixture was diluted with water and S extracted using ethyl acetate (3 x 100 mL). The combined organic layer was dried over spa anhydrous sodium sulfate, filtered and concentrated in vacuo. The crude product was ce purified using column chromatography using hexanes/EtOAc in increasing polarity up to to

50

em

ph

ze

preat quote from the document or the document of the document of the document the document of the document of

der

's

att

7:3 mixture. The fractions containing the desired product were concentrated and ent crystallized in hexanes/EtOAc (7:3) to yield the desired products as off-white solid. ion

wit *1-benzyl-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71a)*, *a* white solid, 0.10 g, 90%; R<sub>f</sub> h a 0.50 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.03 (s, 1H, NH), 7.66 gre (m, 1H, Ar-H), 7.56 (m, 3H, Ar-H), 7.25 (m, 5H, Ar-H), 7.05 (m, 5H, Ar-H), 5.83 (s, 2H, at -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 146.7, 143.0, 137.2, 136.8, 134.2, 128.4, 127.3, qu 126.4, 126.3, 126.1, 123.6, 123.3, 122.9, 121.4, 120.2, 119.3, 112.4, 111.2, 102.9, 79.7, ote 46.7. **HRMS**: (ESI) *m/z* calculated for C<sub>22</sub>H<sub>18</sub>N<sub>3</sub>[M+H]<sup>+</sup> 324.1495, observed 324.1501.

fro *Compound (71b)*, a white solid, 0.19 g, 92%; R<sub>f</sub> 0.60 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR <sup>m</sup> (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.14 (s, 1H, NH), 7.80 (m, 2H, Ar-H), 7.66 (m, 1H, Ar-H), the 7.25 (m, 10H, Ar-H), 6.88 (s, 1H, Ar-H), 5.78 (s, 2H, -CH<sub>2</sub>), 2.41 (s, 3H, -CH<sub>3</sub>); <sup>13</sup>C NMR <sup>do</sup> (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  147.3, 142.7, 137.7, 137.1, 136.5, 132.9,129.9, 128.5, 126.7, <sup>cu</sup> 126.0, 123.8, 123.3, 123.1, 121.3, 120.2, 119.4, 111.9, 110.1, 103.5, 48.3, 21.2. **HRMS**: <sup>me</sup> (ESI) *m/z* calculated for C<sub>23</sub>H<sub>20</sub>N<sub>3</sub>[M+H]<sup>+</sup> 338.1652, observed 338.1656.

nt

or  $I-(4-tert-butylbenzyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71c), a white solid, 0.2 g, 84%; R_f 0.64 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d_6, 400MHz): <math>\delta = 12.03$  (s, 1H, NH), 7.66 (m, 1H, Ar-H), 7.56 (m, 3H, Ar-H), 7.25 (m, 5H, Ar-H), 7.05 (m, 5H, Ar-H), 5.83 (s, 2H, -CH<sub>2</sub>), 1.20 (s, 9H, (-CH<sub>3</sub>)<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d\_6, 100MHz)  $\delta$  150.3, 146.9, 143.0, 137.2, 136.8, 134.2, 128.4, 127.3, 126.4, 126.3, 126.1, 123.6, 123.3, 122.9, 121.4, 120.2, 119.3, 112.4, 111.2, 102.9, 79.7, 47.4, 34.6, 31.5, 22.6. HRMS: (ESI) *m/z* calculated for C<sub>26</sub>H<sub>26</sub>N<sub>3</sub>[M+H]<sup>+</sup> 380.2121, observed 380.2121.

51

em

ph

asi

preat quote from the document or the document of the document of the document the document of the document of

der

's

att *I-(4-(trifluoromethoxy)benzyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71d)*, a white ent solid, 0.3 g, 89%; R<sub>f</sub> 0.70 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.68 ion (s, 1H, NH), 7.80 (m, 2H, Ar-H), 7.66 (m, 1H, Ar-H), 7.25 (m, 10H, Ar-H), 6.88 (s, 1H, wit Ar-H), 5.81 (s, 2H, -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  148.9, 148.9, 148.9, 147.4, h a 142.7, 137.3, 136.2, 134.6, 128.4, 127.6, 126.4, 124.3, 124.0, 123.5, 123.4, 121.8, 121.3, gre 120.4, 119.4. 119.2, 116.7, 112.1, 109.9, 103.4, 47.8. **HRMS**: (ESI) *m/z* calculated for at C<sub>23</sub>H<sub>17</sub>F<sub>3</sub>N<sub>3</sub>O[M+H]<sup>+</sup> 480.1318, observed 408.1319. qu

ote *1-(4-(trifluoromethylthio)benzyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71e)*, a white fro solid, 0.5 g, 90%; R<sub>f</sub> 0.60 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta =$ m 12.06 (s, 1H, NH), 7.77 (m, 8H, Ar-H), 7.25 (m, 8H, Ar-H), 6.95 (s, 1H, Ar-H), 6.50 (m, the 2H, -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  146.8, 142.9, 141.2, 137.3, 137.2, 136.8, do 128.5, 128.4, 128.1, 127.0, 123.7, 123.5, 123.2, 122.4, 121.4, 120.2, 119.4, 112.4, 111.1, cu 103.0, 47.3. **HRMS**: (ESI) *m/z* calculated for C<sub>23</sub>H<sub>17</sub>F<sub>3</sub>N<sub>3</sub>S[M+H]<sup>+</sup> 424.1090, observed me 424.1090.

<sup>nt</sup> *4-((2-(1H-indol-2-yl)-1H-benzo/d/imidazol-1-yl)methyl)benzonitrile (71f)*, a white solid,
 or 0.4 g, 84%; R<sub>f</sub> 0.50 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.06 (s,
 <sup>use</sup> 1H, NH), 7.75 (m, 9H, Ar-H), 7.31 (m, 9H, Ar-H), 7.00 (s, 1H, Ar-H), 5.84 (s, 2H, -CH<sub>2</sub>);
 <sup>thi</sup> <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 146.8, 145.8, 144.6, 143.1, 142.9, 142.9, 138.1, 137.2,
 <sup>s</sup> 136.7, 133.3, 133.2, 132.8, 128.4, 127.7, 127.6, 127.5, 127.5, 127.3, 126.9, 123.7, 123.5,
 <sup>spa</sup> 123.2, 121.9, 121.4, 120.2, 119.4, 119.0, 112.4, 111.6, 111.4, 111.0, 110.8, 110.3, 106.3,
 <sup>ce</sup> 103.0, 79.7, 47.5. HRMS: (ESI) *m/z* calculated for C<sub>23</sub>H<sub>17</sub>N<sub>4</sub>[M+H]<sup>+</sup> 349.1448, observed
 <sup>to</sup> 349.1451.

52

em

ph

asi

t.]

der

's

qu

att *1-(4-nitrobenzyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71g)*, a yellow solid, 0.7 g, ent 86%; R<sub>f</sub> 0.43 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.08 (s, 1H, ion NH), 8.20 (d, *J*=8.8 Hz, 2H, Ar-H), 7.56 (m, 5H, Ar-H), 7.25 (m, 6H, Ar-H), 7.00 (m, 2H, wit Ar-H), 6.06 (s, 2H, -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 147.3, 146.8, 145.2, 142.9, h a 137.3, 136.7, 128.3, 127.8, 126.9, 124.6, 123.7, 123.6, 123.3, 121.4, 120.2, 119.4, 112.4, gre 111.0, 103.1, 47.4. HRMS: (ESI) *m/z* calculated for C<sub>22</sub>H<sub>17</sub>N<sub>4</sub>O<sub>2</sub> [M+H]<sup>+</sup> 369.1346, at observed 369.1348.

ote *1-(4-(methylsulfonyl)benzyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71h)*, a white fro solid, 0.6 g, 86%;  $R_f$  0.19 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta =$ m 12.14 (s, 1H, NH), 7.97 (m, 2H, Ar-H), 7.66 (m, 1H, Ar-H), 7.25 (m, 8H, Ar-H), 6.72 (s, the 1H, Ar-H), 5.87 (s, 2H, -CH<sub>2</sub>), 3.09 (s, 3H, -CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  147.1, do 142.7, 142.3, 140.4, 136.9, 136.1, 128.5, 128.3, 127.0, 126.2, 124.2, 123.7, 123.5, 121.3, cu 120.6, 119.6, 111.9, 109.7, 103.3, 48.0, 44.5. **HRMS**: (ESI) *m/z* calculated for me C<sub>23</sub>H<sub>20</sub>N<sub>3</sub>O<sub>2</sub>S[M+H]<sup>+</sup> 402.1271, observed 402.1280.

<sup>nt</sup> *Methyl 4-((2-(1H-indol-2-yl)-1H-benzo[d]imidazol-1-yl)methyl)benzoate (71i)*, a white <sup>or</sup> solid, 0.9 g, 87%; R<sub>f</sub> 0.45 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = <sup>use</sup> 12.01 (s, 1H, NH), 7.97 (m, 2H, Ar-H), 7.66 (m, 1H, Ar-H), 7.25 (m, 9H, Ar-H), 6.84 (s, <sup>thi</sup> 1H, Ar-H), 5.98 (s, 2H, -CH<sub>2</sub>), 3.80 (s, 3H, -CH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  166.3, <sup>s</sup> 146.9, 142.9, 142.9, 137.2, 136.8, 130.3, 129.3, 128.3, 127.0, 126.8, 123.7, 123.5, 123.2, <sup>spa</sup> 121.4, 120.2, 119.4, 112.4, 111.1, 103.1, 79.7, 52.6. **HRMS**: (ESI) *m/z* calculated for <sup>ce</sup> C<sub>24</sub>H<sub>20</sub>N<sub>3</sub>O<sub>2</sub>[M+H]<sup>+</sup> 382.1550, observed 382.1556. to

53

em

ph

asi

preat quote from the document or the document the document of the document of the document of the document the document of the

der

's

att *1-(4-(trifluoromethyl)benzyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (71j)*, a white ent solid, 0.2 g, 79%; R<sub>f</sub> 0.60 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = ion 12.06 (s, 1H, NH), 7.75 (m, 3H, Ar-H), 7.56 (m, 2H, Ar-H), 7.31 (m, 4H, Ar-H), 7.17 (m, wit 1H, Ar-H), 7.02 (m, 1H, Ar-H), 6.89 (m, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ h a 146.8, 142.9, 142.2, 137.2, 136.8, 128.4, 128.4, 127.3, 126.9, 126.3, 126.3, 125.9, 123.7, gre 123.5, 123.2, 123.2, 121.4, 120.2, 119.4, 112.4, 111.1, 103.0, 47.4. HRMS: (ESI) *m/z* at calculated for C<sub>26</sub>H<sub>26</sub>N<sub>3</sub>[M+H]<sup>+</sup> 392.1369, observed 392.1369.

ote *1-(cyclopropylmethyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (72a)*, a white solid, 0.3 fro g, 92%; R<sub>f</sub> 0.53 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): δ = 12.00 (s, 1H, m NH), 7.66 (m, 3H, Ar-H), 7.30 (m, 1H, Ar-H), 7.20 (m, 4H, Ar-H), 7.05 (m, 2H, Ar-H), the 4.53 (d, *J*=6.6 Hz, 2H, -CH<sub>2</sub>) 1.33 (m, 1H, -CH), 0.45 (m, 4H, (CH<sub>2</sub>)<sub>2</sub>); <sup>13</sup>C NMR (DMSOdo d<sub>6</sub>, 100MHz) δ 146.4, 142.9, 137.2, 136.8, 128.6, 127.6, 123.5, 123.0, 122.7, 121.5, 120.2, cu 119.1, 112.4, 111.4, 102.9, 79.7, 48.1, 11.6, 4.0. **HRMS**: (ESI) *m/z* calculated for me C<sub>19</sub>H<sub>18</sub>N<sub>3</sub>[M+H]<sup>+</sup> 288.1495, observed 288.1495.

<sup>nt</sup> *1-(cyclohexylmethyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (72b)*, a white solid, 0.20 <sup>or</sup> g, 89%; R<sub>f</sub> 0.68 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 11.94 (s, 1H, <sup>use</sup> NH), 7.66 (m, 3H, Ar-H), 7.50 (m, 1H, Ar-H), 7.25 (m, 3H, Ar-H), 7.05 (m, 2H, Ar-H), <sup>thi</sup> 4.43 (d, *J*=7.4 Hz, 2H, -CH<sub>2</sub>) 1.92 (m, 1H, -CH), 1.54 (m, 5H, (-CH)<sub>5</sub>), 1.09 (m, 5H, (-<sup>s</sup> CH)<sub>5</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  146.4, 142.9, 137.2, 137.1, 128.6, 127.7, 123.5, <sup>spa</sup> 122.9, 122.6, 121.5, 120.1, 119.1, 112.4, 111.6, 102.7, 79.7, 50.2, 30.6, 26.2, 25.7. **HRMS**: <sup>ce</sup> (ESI) *m/z* calculated for C<sub>22</sub>H<sub>24</sub>N<sub>3</sub>[M+H]<sup>+</sup> 330.1965, observed 330.1965. <sup>to</sup>

54

10

em

 $\mathbf{p}\mathbf{h}$ 

ze

preat quote from the document or the document the document of the document of the document of the document the document of the

der

's

att *I-(cyclobutylmethyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (72c)*, a white solid, 0.40 ent g, 90%; R<sub>f</sub> 0.62 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 12.00 (s, 1H, ion NH), 7.66 (m, 3H, Ar-H), 7.30 (m, 1H, Ar-H), 7.20 (m, 6H, Ar-H), 7.00 (s, 1H, Ar-H), wit 4.53 (d, *J*=6.6 Hz, 2H, -CH<sub>2</sub>) 3.10 (m, 1H, -CH), 2.15 (m, 2H, -CH<sub>2</sub>), 1.95 (m, 4H, (-CH<sub>2</sub>)<sub>2</sub>); h a <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  146.7, 142.4, 136.8, 136.6, 128.6, 127.3, 123.7, 122.9, gre 122.7, 121.2, 120.2, 119.3, 111.9, 110.2, 102.9, 49.5, 36.1, 26.7, 18.5. **HRMS**: (ESI) *m/z* at calculated for C<sub>20</sub>H<sub>20</sub>N<sub>3</sub>[M+H]<sup>+</sup> 302.1652, observed 302.1652. qu

ote *1-allyl-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (72d)*, a white solid, 0.20 g, 94%; R<sub>f</sub> 0.55 fro (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 12.98$  (s, 1H, NH), 12.03 (s, m 1H, NH), 7.66 (m, 2H, Ar-H), 7.56 (d, *J*=7.4 Hz, 1H, Ar-H), 7.48 (m, 1H, Ar-H), 7.24 (m, the 4H, Ar-H), 7.05 (m, 1H, Ar-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  146.9, 144.2, 137.7, do 135.2, 129.1, 128.3, 123.3, 123.1, 122.2, 121.3, 120.2, 118.9, 112.4, 111.6, 102.1, 79.9, cu 47.7. **HRMS**: (ESI) *m/z* calculated for C<sub>18</sub>H<sub>16</sub>N<sub>3</sub>[M+H]<sup>+</sup> 274.1339, observed 274.1339.

<sup>me</sup> 2-(1H-indol-2-yl)-1-phenethyl-1H-benzo[d]imidazole (72e), a white solid, 0.31 g, 88%; <sup>nt</sup> R<sub>f</sub> 0.47 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 11.97$  (s, 1H, NH), <sup>or</sup> 7.66 (m, 4H, Ar-H), 7.48 (m, 10H, Ar-H), 4.77 (t, J=7.5, 15.1 Hz, 2H, -CH<sub>2</sub>), 3.16 (t, J=<sup>use</sup> 7.6, 15.1 Hz, 2H, -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  146.4, 142.8, 138.3, 137.2, <sup>thi</sup> 136.4, 129.4, 128.9, 128.5, 127.5, 127.1, 123.6, 122.9, 122.7, 121.5, 120.2, 120.1, 119.1, <sup>s</sup> 112.4, 111.1, 102.9, 102.7, 45.9, 35.6. **HRMS**: (ESI) *m/z* calculated for C<sub>23</sub>H<sub>20</sub>N<sub>3</sub>[M+H]<sup>+</sup> <sup>spa</sup> 338.1652, observed 338.1652.

ce

to 1-(3-(benzyloxy)propyl)-2-(1H-indol-2-yl)-1H-benzo[d]imidazole (72f), a white solid, $0.15 g, 90%; R<sub>f</sub> 0.49 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz): <math>\delta = 11.59$  (s, em 55

ph

asi

t.]

der

's
att 1H, NH), 7.66 (m, 3H, Ar-H), 7.48 (m, 12H, Ar-H), 4.75 (t, *J*= 7.2, 14.44 Hz, 2H, -CH<sub>2</sub>), ent 4.59 (s, 2H, -CH<sub>2</sub>), 3.66 (t, *J*= 5.4, 10.96 Hz, 2H, -CH<sub>2</sub>), 2.38 (p, *J*= 7.0, 12.96 Hz, 2H, ion CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 146.5, 142.5, 138.1, 136.8, 136.4, 128.7, 128.6, wit 127.8, 127.7, 127.0, 123.8, 122.9, 122.8, 121.3, 120.2, 119.3, 111.7, 109.8, 103.2, 66.8, h a 41.9, 30.1. HRMS: (ESI) *m/z* calculated for C<sub>25</sub>H<sub>24</sub>N<sub>3</sub>O[M+H]<sup>+</sup> 382.1914, observed gre 382.1916.

qu *1-benzyl-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (73a)*, *a* white solid, 0.6 g, ote
92%; R<sub>f</sub> 0.60 (1:1/hexanes:EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz): δ = 11.91 (s, 1H, fro NH), 7.76 (d, *J*=7.2Hz, 1H, Ar-H), 7.57 (d, *J*=7.4Hz, 1H, Ar-H), 7.36 (m, 6H, Ar-H), 7.34 m (dd, *J*=8.8, 8.8Hz, 1H, Ar-H), 7.23 (m, 2H, Ar-H), 6.83 (dd, *J*=8.8, 8.8Hz, 1H, Ar-H), 5.86 the (s, 2H, -CH<sub>2</sub>-), 3.74 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 154.3, 147.0, 142.9, do 137.3, 136.8, 132.5, 129.4, 128.7, 127.9, 127.4, 126.5, 123.3, 122.9, 119.2, 114.6, 113.2, cu 111.1, 102.8, 102.1, 79.7, 55.7, 47.7. HRMS: (ESI) *m/z* calculated for C<sub>23</sub>H<sub>20</sub>N<sub>3</sub>O [M+H]<sup>+</sup> me 354.1601, observed 354.1601.

### nt 1-(4-(trifluoromethyl)benzyl)-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole

or (73b), a white solid, 0.35 g, 90%; R<sub>f</sub> 0.60 (1:1/hexanes:EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 use MHz): δ = 11.91 (s, 1H, NH), 7.76 (d, *J*=7.2Hz, 1H, Ar-H), 7.57 (d, *J*=7.4Hz, 1H, Ar-H),
thi 7.36 (m, 6H, Ar-H), 7.34 (dd, *J*=8.8, 8.8Hz, 1H, Ar-H), 7.23 (m, 2H, Ar-H), 6.83 (dd,
<sup>s</sup> *J*=8.8, 8.8Hz, 1H, Ar-H), 5.86 (s, 2H, -CH<sub>2</sub>-), 3.74 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>,
spa 100 MHz) δ 154.3, 147.0, 142.9, 137.3, 136.8, 132.5, 129.4, 128.7, 127.9, 127.4, 126.5,
ce 123.3, 122.9, 119.2, 114.6, 113.2, 111.1, 102.8, 102.1, 79.7, 55.7, 47.7. HRMS: (ESI) *m/z*to calculated for C<sub>24</sub>H<sub>19</sub>F<sub>3</sub>N<sub>3</sub>O [M+H]<sup>+</sup> 422.1475, observed 422.1482.

56

em

ph

ze

preat quote from the document or the document of the document of the document the document of the document of

der

's

att 1-(cyclohexylmethyl)-2-(5-methoxy-1H-indol-2-yl)-1H-benzo/d/imidazole (73c), a white ent solid, 0.6 g, 78%;  $R_f$  0.68 (1:1/hexanes:EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta = 11.86$ ion (s, 1H, NH), 7.67 (d, J=7.2Hz, 1H, Ar-H), 7.56 (d, 1H, Ar-H), 7.34 (dd, J=8.8, 8.8Hz, 1H, wit Ar-H), 7.23 (m, 2H, Ar-H), 6.83 (dd, J=8.8, 8.8Hz, 1H, Ar-H), 4.41 (d, J=6.6Hz, 2H, h a CH<sub>2</sub>-), 3.79 (s, 3H, OCH<sub>3</sub>), 1.7 (m, 1H, -CH-), 1.55 (m, 5H, -CH-), 1.1 (m, 5H, -CH-); <sup>13</sup>C gre NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 154.3, 146.7, 144.2, 135.2, 132.9, 129.4, 128.7, 122.9, at 122.1, 118.9, 114.9, 114.1, 113.3, 113.2, 111.6, 102.2, 101.9, 55.7, 50.2, 38.5, 30.6, 26.2, qu 25.7. **HRMS**: (ESI) *m/z* calculated for C<sub>23</sub>H<sub>26</sub>N<sub>3</sub>O [M+H]<sup>+</sup> 360.2070, observed 360.2070. ote

fro 1-(cyclopropylmethyl)-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (73d),а white solid, 0.46 g, 82%; R<sub>f</sub> 0.53 (1:1/hexanes:EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz): m δ = 12.93 (s, 1H, NH), 11.86 (s, 1H, NH), 7.67 (d, *J*=7.2Hz, 1H, Ar-H), 7.56 (d, 1H, Arthe H), 7.34 (dd, J=8.8, 8.8Hz, 1H, Ar-H), 7.23 (m, 2H, Ar-H), 6.83 (dd, J=8.8, 8.8Hz, 1H, do Ar-H), 4.50 (d, J=6.6Hz, 2H, -CH<sub>2</sub>-), 3.78 (s, 3H, OCH<sub>3</sub>), 1.33 (m, 1H, -CH-), 0.48 (m. cii 4H, 2[-CH<sub>2</sub>-]); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz) δ 154.3, 146.7, 144.2, 135.2, 132.9, 129.4, me 128.7, 122.9, 122.1, 118.9, 114.9, 114.1, 113.3, 113.2, 111.6, 102.2, 101.9, 79.8, 55.7, 48.1, nt 11.6, 4.0. HRMS: (ESI) m/z calculated for C<sub>20</sub>H<sub>20</sub>N<sub>3</sub>O [M+H]<sup>+</sup> 318.1601, observed or use 318.1601.

<sup>thi</sup> *1-allyl-2-(5-methoxy-1H-indol-2-yl)-1H-benzo[d]imidazole (73e)*, a white solid, 0.25 g, <sup>s</sup> 96%; R<sub>f</sub> 0.56 (1:1/hexanes:EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  = 11.86 (s, 1H, <sup>spa</sup> NH), 7.67 (d, *J*=7.2Hz, 1H, Ar-H), 7.56 (d, 1H, Ar-H), 7.34 (dd, *J*=8.8, 8.8Hz, 1H, Ar-H), <sup>ce</sup> 7.23 (m, 2H, Ar-H), 6.83 (dd, *J*=8.8, 8.8Hz, 1H, Ar-H), 6.21 (m, 1H, -CH-), 5.21 (m, 2H, <sup>to</sup> -CH<sub>2</sub>-), 4.88 (m, 1H, -CH-), 3.79 (s, 3H, OCH<sub>3</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  154.3, em

57

ph

asi
reat quote from he document or att se this space to mphasize a key oint. To place his text box h a nywhere on the age, just drag t.]

der

's

146.7, 144.2, 135.2, 132.9, 129.4, 128.7, 122.9, 122.1, 118.9, 114.9, 114.1, 113.3, 113.2, ent 111.6, 102.2, 101.9, 55.7, 46.8. **HRMS**: (ESI) m/z calculated for C<sub>19</sub>H<sub>18</sub>N<sub>3</sub>O [M+H]<sup>+</sup> ion 304.1444, observed 304.1444. wit

gre 3.1.5 General procedure for the synthesis of symmetrically or asymmetrically bisat alkylated 2-indolylbenzimidazoles 75a - 75c and 77a - 77d.

qu There were two steps. Step 1, a solution of 2-indolylbenzimidazole 66a (1 equiv.) in 3 mL ote of CH<sub>3</sub>CN was cooled to 0 °C in ice/water bath, potassium hydroxide powder (1 equiv.) fro was added and the reaction mixture was left to stir for 20 min. Thereafter, alkyl or benzyl m bromide (1 equiv.) was added and the reaction mixture was stirred for 4 h at 0 °C. For the the benzylation, the reaction was done at 0 °C for 4 h. For the cyclopropylmethylation reaction, do the reaction was allowed to warm to rt and continued at rt for 6 h. Step 2, Once the monocu alkylation was complete, second equivalence of potassium hydroxide powder was added me to the reaction mixture and stirred at 0 °C for 20 min. Then, alkyl bromide (1 equiv.) was nt added and the reaction mixture was stirred at 0 °C for another 20 min, and subsequently or allowed to warm to rt and stirred for 7 h. Once the reaction was complete, crude reaction use mixture was diluted with water and extracted using ethyl acetate (3 x 100 mL). The thi combined organic layer was dried over anhydrous sodium sulfate, filtered and concentrated S in vacuo. The crude product was purified using column chromatography using spa hexanes/EtOAc in increasing polarity up to 7:3 mixture. The fractions containing the ce desired product were concentrated and crystallized in hexanes/EtOAc (7:3) to yield the to desired products as off-white solid.

- em
- ph

58

ze

preat quote from the document or the document the document of the document of the document of the document the document of the

att 1-benzyl-2-(1-benzyl-1H-indol-2-yl)-1H-benzo[d]imidazole (75a), a white solid, 0. 11g, ent 68%; R<sub>f</sub> 0.71 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta = 7.78$  (m, 1H, ion Ar-H), 7.66 (m, 1H, Ar-H), 7.25 (m, 2H, Ar-H), 7.12 (m, 6H, Ar-H), 7.03 (m, 6H, Ar-H), wit 5.92 (s, 2H, -CH<sub>2</sub>), 5.45 (s, 2H, -CH<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz) δ 163.0, 162.8, h a 160.6, 160.4, 145.9, 142.9, 138.1, 135.6, 134.9, 134.9, 133.4, 133.4, 129.0, 128.9, 128.8, gre 128.7, 127.8, 127.3, 123.9, 123.8, 123.1, 121.8, 120.9, 120.0, 116.1, 116.0, 115.7, 115.5, at 111.7, 111.5, 106.1, 79.7, 47.3, 46.7. **HRMS**: (ESI) *m/z* calculated for C<sub>29</sub>H<sub>22</sub>N<sub>3</sub>[M+H]<sup>+</sup> qu 414.1965, observed 414.1966.

ote

der

's

fro 2-(1-benzyl-1H-indol-2-yl)-1-(cyclopropylmethyl)-1H-benzo[d]imidazole (77a), a white m solid, 0.14 g, 62%; R<sub>f</sub> 0.81 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = the 7.66 (m, 2H, Ar-H), 7.30 (m, 7H, Ar-H), 7.10 (m, 6H, Ar-H), 5.81 (s, 2H, -CH<sub>2</sub>), 4.05 (d, do J=6.6 Hz, 2H, -CH<sub>2</sub>) 1.33 (m, 1H, -CH), 0.45 (m, 4H, (CH<sub>2</sub>)<sub>2</sub>); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, cu 100MHz) δ 146.4, 142.9, 137.2, 136.8, 128.6, 127.6, 123.5, 123.0, 122.7, 121.5, 120.2, me 119.1, 112.4, 111.4, 106.7, 49.1, 47.7, 29.7, 11.3, 4.4. HRMS: (ESI) m/z calculated for nt C<sub>26</sub>H<sub>24</sub>N<sub>3</sub>[M+H]<sup>+</sup> 378.1965, observed 378.1968.

or 2-(1-(cyclopropylmethyl)-1H-indol-2-yl)-1-methyl-1H-benzo[d]imidazole (77b), a white
use solid, 0.16 g, 66 %; R<sub>f</sub> 0.53 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400MHz): δ = 7.66
thi (m, 2H, Ar-H), 7.56 ( d, J=7.4 Hz, 1H, Ar-H), 7.48 (m, 1H, Ar-H), 7.24 (m, 4H, Ar-H),
<sup>s</sup> 7.05 (m, 1H, Ar-H), 6.86 (s,1H, Ar-H), 4.53 (d, J=6.6 Hz, 2H, -CH<sub>2</sub>), 3.97 (s, 3H, N-CH<sub>3</sub>),
<sup>spa</sup> 1.33 (m, 1H, -CH), 0.35 (m, 4H, (CH<sub>2</sub>)<sub>2</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100MHz) δ 146.6, 142.9,
to

59

em

ph

- ze

preat quote from the document or use this space to emphasize a key point. To place this text box nywhere on the page, just drag

t.]

der

's

ion

att 48.3, 31.4, 11.8, 3.6. **HRMS** (ESI) *m/z* calculated for C<sub>20</sub>H<sub>20</sub>N<sub>3</sub>[M+H]<sup>+</sup> 302.1652, observed ent 302.1652.

wit *1-methyl-2-(1-(prop-2-ynyl)-1H-indol-2-yl)-1H-benzo[d]imidazole (77c)*, a white solid, h a 0.11 g, 84 %; Rf 0.71 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 7.66 (m, gre 2H, Ar-H), 7.56 ( d, *J*=7.4 Hz, 1H, Ar-H), 7.48 (m, 1H, Ar-H), 7.24 (m, 4H, Ar-H), 7.05 at (m, 1H, Ar-H), 6.86 (s,1H, Ar-H), 5.56 (s, 2H, -CH<sub>2</sub>), 4.00 (s, 3H, N-CH<sub>3</sub>), 2.21 (s, 1H, qu CCH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100MHz)  $\delta$  145.9, 142.9, 137.8, 135.9, 127.5, 127.2, 123.8, ote 123.3, 122.7, 121.5, 120.9, 120.1, 110.7, 109.8, 106.9, 78.8, 72.3, 33.9, 31.7. **HRMS** (ESI) fro *m/z* calculated for C<sub>19</sub>H<sub>16</sub>N<sub>3</sub>[M+H]<sup>+</sup> 286.1339, observed 286.1352.

<sup>m</sup> 2-(1-methyl-1H-indol-2-yl)-1-(prop-2-ynyl)-1H-benzo[d]imidazole (77d), a white solid, the 0.11 g, 71 %; R<sub>f</sub> 0.73 (1:1/Hexanes: EtOAc); <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400MHz):  $\delta$  = 7.66 (m, do 2H, Ar-H), 7.56 ( d, *J*=7.4 Hz, 1H, Ar-H), 7.48 (m, 1H, Ar-H), 7.24 (m, 4H, Ar-H), 7.05 cu (m, 1H, Ar-H), 6.86 (s,1H, Ar-H), 5.27 (s, 2H, -CH<sub>2</sub>), 4.01 (s, 3H, N-CH<sub>3</sub>), 3.57 (s, 1H, me CCH); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>, 100 MHz)  $\delta$  145.6, 142.9, 142.9, 138.7, 135.1, 127.9, 127.1, 123.7, 123.2, 121.6, 120.7, 119.9, 111.4, 111.0, 105.3, 79.2, 76.6, 34.8, 31.9. **HRMS** (ESI) or *m/z* calculated for C<sub>19</sub>H<sub>16</sub>N<sub>3</sub> [M+H]<sup>+</sup> 286.1339, observed 286.1352. use

## thi 3.2 Cell culture

HeLa (cervical cancer), A549 (lung cancer), HepG2(liver cancer), MCF-7 (human breast spa cancer), and MDAMB231 (triple negative breast cancer) cell lines in which were cultured in air-jacketed humidified incubator at 37 °C with 5% CO<sub>2</sub> in DMEM high glucose media supplemented with 10% FBS, 1% penicillin-streptomycin. 1000 cells per well in 96-well em

60

ph

ze

preat quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

t.]

der

's

att plates were plated and then compounds were added to each well at indicated concentrations ent with carrier DMSO and incubated for 72 hours followed by MTT assay. ion

#### wit 3.3. Cell viability determination (MTT assay)

h a The effect of the synthesized compounds on cellular viability was evaluated using the 3-(4,5-dimethylthiazol-2-yl)-2,5gre tetrazolium based calorimetric assay using diphenyltetrazolium bromide (MTT, Acros Organics) dye on five different cancer cell lines at qu HepG2, HeLa, MCF-7, MDA-MB231 and A549. MTT assay measures the activity of ote cellular NAD(P)H-dependent oxidoreductase enzymes reflecting the number of viable fro cells <sup>190</sup>. Briefly, cells were plated in sterile 96-well cell culture plates and allowed to reach  $\sim$ 70% confluency. Cells were then treated with benzimidazole derivatives. After 48 or 72 m the h of treatment, 0.05% v/v MTT dye was added and cells were incubated at 37 °C for 3-4 do h. Formazan crystals formed from the reduction of MTT dye were dissolved in DMSO and the absorbance <sup>191,192</sup> was measured at 570 nm using the SpectraMax M5e plate reader cii me (Molecular Devices).

#### nt

#### 3.4. Apoptosis / necrosis assay

or

A fluorescence dye-based apoptosis / necrosis assay was used to assess the mechanism of benzimidazole-induced cell death. This assay employs HO-33342 (HO), a DNA-binding dye that permeates healthy cells and propidium iodide (PI), a dye that can only penetrate cells with damaged membranes (necrotic cells). At the end of drug treatments for 24 h, spa HeLa cells were incubated with PI (25  $\mu$ g/mL) for 10 min in the dark. Next, cells were washed with PBS and incubated with HO dye (11.25  $\mu$ g/mL). Images were taken immediately using the EVOS FL Auto Imaging System (Thermo Fisher Scientific). The

61

ph

asi

preat quote from the document or the document of the document the document of the document of

der

's

att percentage of apoptotic cells were determined by the number of bright HO-stained cells as ent a fraction of total HO-stained cells whereas PI-stained cells represent total necrotic cells. ion

#### wit 3.5 Western blot analysis

h a HeLa cells treated with benzimidazole derivatives for 48 h were lysed in modified RIPA gre buffer (50 mM Tris pH 8.0, 150 mM NaCl, 1% v/v NP40, 0.5% w/v deoxycholate, 0.1% at w/v SDS, 10% v/v glycerol, 10 mM NaF, 0.4 mM EDTA) with protease inhibitors (Sigma-qu Aldrich). Lysates were collected using a scraper, transferred to microcentrifuge tubes, and ote cleared by centrifugation at 10,000g for 10 min. Next, Laemmli sample buffer containing fro SDS and  $\beta$ -mercaptoethanol was added and the samples were denatured by heating them m at 95 °C for 10 min. Subsequently, samples were separated on polyacrylamide gels and the transferred to PVDF membrane and processed for Western blot analysis with do chemiluminescence detection. Images were obtained using Azure biosystems c500 imager cu and analyzed using ImageJ 1.8.0 software.

me

#### 3.6 Human topoisomerase II assay

nt

Human Topoisomerase II (topo II) activity assay was performed according to the protocol or provided by the manufacturer (TopoGen). Briefly, the total reaction volume was fixed at use 20  $\mu$ L. For preparation of one complete reaction, 4  $\mu$ L of a 1:1 mixture of buffer A (0.5 M thi Tris-HCl, pH 8, 1.5 M NaCl, 0.1M MgCl<sub>2</sub>, 5 mM dithiothreitol, 300 µg/mL BSA) and S Buffer B (20 mM ATP) were added to 14.75 µL of dH<sub>2</sub>O and 1 µL of kinetoplast DNA spa substrate. Finally, 1  $\mu$ L of benzimidazoles was added followed by topo II (2 units). After ce 45 min of incubation at 37 °C in a water bath, the reactions were stopped by the addition to of 4  $\mu$ L of stop buffer (5% sarkosyl, 0.0025% bromophenol blue, 25% glycerol). Next, em

62

ph

ze

preat quote from the document or the document of the document of the document the document of the document of

der

's

qu

att 12.5 μL of a 125 μg/mL solution of proteinase K from *Tritirachium album* (Sigma ent Aldrich) was added to degrade the topoisomerase II from the decatenated DNA products.
ion After an additional 15 min of incubation at 37 °C in a water bath, the reactions were wit analyzed on a 1% agarose gel by running at 120 V for 30 min in TAE buffer (40 mM Tris h a base, 1.1% glacial acetic acid, 20 mM disodium EDTA dihydrate). Gels were stained with gre SYBR Safe DNA gel stain (Invitrogen) and photographed under UV illumination using a at c500 Azure Biosystems reader.

#### ote 3.7 Materials for the BMP project

fro C2C12 mouse myoblast cells were obtained from American Type Culture Collection (ATCC®, Manassas, VA, USA; ATCC® CRL-1772). Dulbecco's Modified Eagle's m the Medium (PSG), Penicillin/Streptomycin/Glutamine solution (DMEM), Penicillin/Streptomycin solution (PS), 0.25% Trypsin-EDTA were purchased from Gibco do by Life Technologies (Gaithersburg, MD, USA). Fetal Bovine Serum (FBS) was obtained cu from Atlanta Biologicals (Flowery Branch, GA, USA). Vascular Cell Basal medium, me Endothelial Cell Growth Kit – BBE, Trypsin/ EDTA solution for Primary cells, Trypsin nt Neutralizing Solution were obtained from ATCC (Manassas, VA, USA). Acridine or Orange/Propidium Iodide stain was purchased from Logos Biosystem (Annandale, VA, use USA). Recombinant BMP7 was purchased from R&D Systems (Minneapolis, MN, USA). thi TGX Fast Cast Acrylamide gel kit (12%), 4X Laemmli Sample Buffer, 10X S Tris/Glycine/SDS buffer (Running Buffer) and Precision Plus Protein Standard were spa obtained from Bio-Rad Laboratories (Hercules, California, USA). Amersham<sup>™</sup> Protran<sup>™</sup> ce 0.2 µm Nitrocellulose membrane was obtained from GE Healthcare Life Sciences to (Pittsburgh, PA, USA). Bovine Serum Albumin (30% BSA solution) was from Alfa Aesar em

63

ph

asi

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

t.]

att (Ward Hill, MA, USA). Non-fat dry milk and Goat anti-rabbit IgG-HRP: sc-2004 were ent obtained from Santa Cruz Biotechnology, Inc. (Dallas, TX, USA). GeneTex Trident Pico ion Western HRP Substrate solutions for western blot imaging were bought from GeneTex wit (Irvine, CA, USA). Ponceau S solution, Poly-D-Lysine hydrobromide (PDL) and Trish a EDTA buffer solution were procured from Sigma Aldrich (St. Louis, MO, USA). gre Paraformaldehyde stock solution (16%) was from Electron Microscopy Sciences (Hatfield, at PA, USA). Cy3-goat anti-rabbit IgG was obtained from Jackson ImmunoResearch qu Laboratories (West Grove, PA, USA). Vectashield Mounting Medium with DAPI was ote purchased from Vector Laboratories, Inc. (Burlingame, CA, USA). RealTime-Glo™ MT fro cell Viability Assay kit was purchased from Promega Corporation (Madison, WI, USA). m Phospho-Smad1/ 5(S463/465)/ 9(S465/467) rabbit monoclonal antibody (#13820S), the Smad1 (D5907) XP® rabbit monoclonal antibody (#6944), Phospho-Akt (Ser473) (D9E) do XP® rabbit monoclonal antibody (#4060), Akt (pan) (C67E7) rabbit monoclonal antibody cu (#4691), and Cell lysis Buffer (10X) were obtained from Cell Signaling Technology me (Danvers, MA, USA). The phospho-Smad 1/5/8 antibody, CU503AB rabbit polyclonal nt antibody was generously provided by the Thomas Jessell Laboratory; HHMI (Columbia or University, NY, USA).

use

#### thi 3.8 Cell culture maintenance - C2C12 cells

<sup>S</sup> C2C12 cells were cryopreserved in liquid nitrogen in Freezing Medium (Gibco by Life spa Technologies, city, state, co). When required, cells were thawed and maintained in ce Complete Growth Media (CGM) which consists of DMEM/ 10% FBS/ 1X PSG in T-75 to flasks, incubated at 37°C in 5% CO<sub>2</sub>. C2C12 cells were sub-cultured every second day by em the following protocol: a) cells were removed from the cell culture incubator and observed

64

ph

asi

preat quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

t.]

's att under a microscope to note the confluency of cells and cellular structure; b) the growth ent medium was aspirated and the cells were washed with room temperature 1X PBS; c) ion Trypsin-EDTA (5 mL, 0.25%) was added to the flask, incubated for 3 minutes at 37°C to wit detach the cells and then the flask was examined under the microscope to ensure complete h a trypsinization of the cells; d) an equal volume of CGM was added to neutralize the trypsin; gre e) the cell suspension was collected, transferred to a 15 mL conical tube and the cells were at pelleted at 1000 rpm for 3 minutes at room temperature; f) the supernatant was aspirated qu and the cell pellet was resuspended in CGM; g) the concentration of cells was determined ote using the Luna-FL<sup>™</sup> Dual Fluorescence Cell Counter from Logos Biosystem (Annandale, fro

VA, USA); and, finally, h) the cells were seeded at a 1:5 dilution into a fresh T-75 flask m and incubated at 37°C in 5% CO<sub>2</sub>.

the

der

#### do 3.9 Determination of cell concentration - C2C12 cells

cu At the time of the cell pellet resuspension (step (f)), a drop of the cell suspension was me transferred to a fresh micro-centrifuge tube to serve as a representative sample. To nt determine the cell concentration a sample of the cells (18  $\mu$ L) was mixed with 2  $\mu$ L acridine or orange/propidium iodide stain and 10  $\mu$ L of this mixture was loaded onto a LUNA cell use counting slide. The slide was inserted into the cell counter and the cell concentration and thi viability was recorded.

#### 3.10 Determining cell viability using MT-Glo assay

spa

S

C2C12 cells were seeded in a 96-well plate, 100 μL at 5 x 10<sup>5</sup> cells/mL, and incubated overnight in CGM to achieve 80-90% confluence. The medium was replaced with Serum to Starvation Medium (SSM) containing DMEM/ 1X PS and the cultures were serum starved em

ph

asi

der reat quote from he document or use this space to mphasize a key oint. To place his text box nywhere on the age, just drag t.]

's

att for at least 16 h. The test compounds were prepared in DMEM/PS at the following ent concentrations: 1 mM, 100 µM, 10 µM, 1 µM, 0.1 µM, 0.01 µM. Cells incubated in the ion presence of DMEM/PS alone served as a positive control and treatment with 78 ppm Triton wit X-100 was used as negative control. Following the serum starvation period, the cells were h a treated with controls or test compounds for 24 h. Next, the MT Glo Cell Viability substrate gre and NanoLuc® enzyme were diluted in DMEM/PS according to manufacturer's protocol at and added to the cells. The plate was then incubated for 30-40 minutes at 37°C, 5% CO<sub>2</sub>. qu Cellular luminescence was measured using the FilterMax F5 Multi-mode Microplate ote Reader (Molecular Devices, San Jose, CA, USA). Cell viability from three independent fro experiments with individual experiments carried out in triplet was determined and reported m

as a percentage of control.

the

#### **3.11 Treatment of C2C12 cells and preparation of whole cell lysates** do

C2C12 cells were seeded in 35 mm dishes at 7.5 x 10<sup>4</sup> cells/mL (3 mL) in CGM and cu incubated overnight at 37°C in 5% CO<sub>2</sub> to reach 70-80% confluence. Next, culture medium me was replaced with SSM and incubated for 16-18 hours. After serum starvation, cells were nt treated with 50 ng/mL BMP7 (positive control) and the indicated concentrations of test or compounds for 30 minutes. All samples were diluted in SSM. Following stimulation, the use treatment medium was removed, and cultures were washed with ice cold 1X TBS for 2 thi minutes. The 1X TBS was replaced with 1X Lysis Buffer supplemented with 1vmM PMSF S spa (phenylmethylsulfonyl fluoride) and incubated on ice for 10 minutes. The cells were scraped, collected into microcentrifuge tubes and incubated on ice for an additional 15 ce minutes to ensure complete cell lysis. The lysed cells were centrifuged at 15,000 rpm for to

em

ph

ze

great quote from the document or use this space to emphasize a key point. To place this text box inywhere on the page, just drag

t.]

der

's

att 20 minutes at 4°C. The supernatant (whole cell lysate) was transferred to a fresh ent y microcentrifuge tube and stored at -80°C. ion

#### wit 3.12 Determination of Total Protein Concentration: Protein Assays

h a The Amido Schwarz TCA precipitation method was used to estimate the amount of total gre protein in the whole cell lysates. First, whole cell lysates were diluted 1:100 in dH<sub>2</sub>O and at total soluble protein was precipitated using 60% TCA (Trichloroacetic acid) and 1 M qu Tris/1% SDS. Precipitated protein was collected on nitrocellulose membranes (Millipore) ote and stained with Amido Black (Sigma) reagent. Stained protein was eluted from the fro membranes using Elution Buffer (1 M NaOH/0.5 M EDTA/Absolute ethanol). Absorbance m of the eluted samples was measured at 630 nm using the Eppendorf BioSpectrometer (Lake the Forest, CA, USA). The protein concentration in the whole cell lysates was determined do using a BSA standard curve.

cu

#### 3.13 Western Blotting to determine Smad phosphorylation

me

The samples for Western blot analysis were prepared using 10 µg whole cell lysate in 1X nt Laemmli sample buffer. Samples were heated at 100°C for 5 minutes, immediately placed or on ice for 5 minutes and centrifuged at 9500 rpm for 2 minutes at room temperature. The use protein samples along with Precision Plus Protein Standard (5 µL) were loaded on 12% thi TGX Fast Cast Acrylamide gels. Protein was separated by SDS-PAGE using the Gel S Electrophoresis System from Bio-Rad (Hercules, CA, USA) at 200 V. The nitrocellulose spa membrane (GE Healthcare Life Sciences, Pittsburgh, PA, USA) was pre-incubated in cold ce 1X Transfer Buffer (10 mM Tris, 2.5 mM glycine, 20% Methanol) for 30 minutes. Next, to the separated proteins were transferred at 100 V onto the pre-equilibrated nitrocellulose em 67

ph

asi

2	der
great quote from	's
he document or	att
se this space to	membranes. To ensure equal loading of proteins in every lane, the membrane was stained
mphasize a key	with Ponceau S and subsequently washed out with 1X TBS (3X). Non-specific binding
oint. To place	sites were blocked by incubation of the membranes in BSA blocking buffer (5% BSA,
his text box	wit 0.1% Tween 20, 1X TBS) for 30 minutes followed by incubation overnight at 4°C with the
nywhere on the	h a desired primary antibody diluted in BSA blocking buffer. Next, the membranes were
age, just drag	gre washed with 1X TBST (0.1% Tween 20/TBS) for 15 minutes (3X). Membranes were then
t.]	at probed with a HRP-conjugated secondary antibody diluted in milk blocking buffer (5%
	qu non-fat milk, 0.1% Tween20, 1X TBS) for 1 hour at room temperature. The blots were then
	ote washed again with 1X TBST for 15 minutes (3X). The blots were developed using Trident
	fro Pico Western HRP substrate solutions and analysed by capturing the chemiluminescent
	m signal using the Omega Lum <sup>TM</sup> G Imaging System (San Francisco, CA, USA)
	the Overtification of the western blots was carried out using Imagel (Image Processing and
	do
	cu
	me
	nt
	or
	use
	thi
	S
	spa
	ce
	to
	em
	ph 68
	asi

ze

---

great quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

der

's

att

ent

#### **CHAPTER IV. RESULTS AND DISCUSSION**

#### ion 4.1 *O*-(benzotriazole-1-yl)-*N*,*N*,*N*',*N*'-tetramethyluronium

### wit hexafluorophosphate (HBTU) Promoted Synthesis of Benzimidazoles

<sup>h</sup> <sup>a</sup>Our goal is to develop a simple and functional group tolerant method to synthesize <sup>gre</sup> benzimidazoles. As the first step, we synthesized a valine derived amide precursor (15) to at explore a suitable mild reaction condition for the synthesis of benzimidazoles. Compound <sup>qu</sup> 15 was synthesized using commercially available Boc-Val-OH and *o*-diaminobenzene via ote a standard peptide coupling protocol. Then, we proceeded to screen various mild conditions fro to perform a dehydrative cyclization of the amide 15 into the corresponding benzimidazole m (Table 4.1). Our initial attempts to perform the dehydrative cyclization under a mild basic the or acidic condition (Table 4.1, entry 2 - 4) failed to yield the desired product. We then do realized that carbodiimides are known for their ability to promote oxidation or <sup>cu</sup> dehydration.<sup>99,193</sup> Based on this knowledge, we selected several commonly used me carbodiimide-based coupling agents, including N,N'-diisopropylcarbodiimide (DIC), 1nt ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDCI), O-(1H-6-chlorobenzotriazole-1vl)-1,1,3,3-tetramethyluronium hexafluorophosphate (HCTU) and HBTU to assess the or use dehydrative cyclization (Table 4.1, entry 5 - 9). We discovered that HBTU (1 equiv.) yielded the best conversion of 15 into the corresponding benzimidazole 15a (Table 4.1, thi S entry 7). As part of the exploration, we found that catalytic amount of HBTU (0.3 equiv.) spa is ineffective in providing the desired product in high yield. This finding suggests that HBTU is promoting the dehydrative cyclization and less likely to be acting as a catalyst. ce Although DIC, EDCI and HCTU are useful carbodiimide agents for amide formation, these to em

69

ph

asi

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

t.]

att agents did not afford the desired product. This is perhaps due to the reduced reactivity of ent these coupling agents towards activating an amide bond. ion

 $_{\rm wit}$  Table 4.1. Optimization of conditions for the conversion of aryl-amide into  $_{\rm h\ a}$  benzimidazole.

gre at		H <sub>2</sub> N HBTU reflux — 3 h Boc-t		
qu	1	5	15a	
ote	Entry	Reagent	Yield	
	1	No additives	No reaction	
fro	2	DIPEA	No reaction	
m	3	DBU	No reaction	
the	4	HC1	No reaction	
	5	DIC	No reaction	
do	6	EDCI	No reaction	
cu	7	HBTU	94%	
me	8	HCTU	5%	
	9	РуВОР	94%	
nt	10	DIPEA/HCl	No reaction	
or	11	Tetramethylurea	No reaction	

use

thi

We also discovered that (benzotriazol-1-yloxy)-tripyrrolidinophosphonium s hexafluorophosphate (PyBOP), a more reactive coupling agent compared to DIC and spa EDCI, indeed exhibited desirable reactivity towards the dehydrative cyclization. ce Phosphonium-based coupling agents are useful activating reagents for amide formation and to cyclization of thioureas.<sup>194</sup>

70

ph

- asi
- ze

reat quote from he document or use this space to mphasize a key oint. To place his text box nywhere on the age, just drag

t.]

att

der

's

#### 4.1.1 One-pot, HBTU promoted strategy for the synthesis of benzimidazoles ent

With the successful discovery of HBTU as a suitable promotor of benzimidazole synthesis, ion we quickly realized that the amide precursor was synthesized via an HBTU activated wit process as well. Hence, we attempted a one-pot strategy to form the amide precursor from h a the parent carboxylic acid and perform the subsequent benzimidazole formation by using gre 2 equivalents of HBTU. This strategy worked extremely well as a simple one-pot process. at Initial formation of aryl-amide occurred with high conversion within 4 hours at room qu temperature, and then a one-pot HBTU promoted cyclization yielded the desired product ote in less than 3 hours at refluxing temperature. The two-step, one-pot synthesis worked

extremely well, yielding the products in 80 - 99% isolated yield (Scheme 4.1).

m

fro



me Scheme 4.1 Two-step, one-pot synthesis of benzimidazole derivative.

nt

To further confirm the role of HBTU in this process, we performed a series of control or reactions. We observed that none of the components that is part of the reaction mixture, use either reagents or by-products from the amidation step promoted the benzimidazole thi synthesis (Table 4.1, entries 10 and 11). These observations led us to conclude that HBTU S is in fact the promoter of benzimidazole synthesis, and it is presumably activating the amide spa bond.

ce

to

em

ph

preat quote from the document or use this space to emphasize a key point. To place this text box inywhere on the page, just drag

der

's

at

att To investigate the versatility of solvents, we performed the benzimidazole synthesis in 1,4ent dioxane, dimethylformamide (DMF) or toluene. The coupling was performed in one of ion these solvents, and subsequently the crude reaction mixture was subjected to cyclization wit under refluxing temperature in the same solvent. Based on the high yield obtained, all three h a solvents were suitable for this operation, providing great flexibility with the choice of gre solvents.

t.]

 $_{\rm qu}~$  4.1.2 Advantages of HBTU promoted approach in the synthesis of benzimidazoles

ote In comparison to reported methods for benzimidazole syntheses, HBTU promoted fro cyclization approach highlights several important improvements and advantages. First, the m benzimidazole synthesis is a one pot process and high yielding, where current methods the require isolation of the aryl-amide prior to dehydrative cyclization. Second, there is no need do to perform the cyclization in the presence of an acid as a co-solvent, which greatly broadens cu the substrate scope and the functional group tolerability, including various protecting me groups found in amino acids and peptides. Third, the reaction works extremely well in three nt different solvents enabling synthetic flexibility for substrates with limited solubility.

or

#### 4.1.3 Substrate scope of our HBTU promoted approach

use

We proceeded to investigate the substrate scope using a small library of commercially thi available carboxylic acids (Scheme 4.2). Boc-Asp-OMe (40b) was successfully converted s to the beta-benzimidazole derivative (41b) in 92% yield, providing a unique non-protein

spa amino acid that is useful for medicinal chemistry. In addition, four different alkyl
ce carboxylic acids (40c-40f) were converted into the corresponding benzimidazoles (41c – to
41f) in high yield. As peptide substrates are of prime interest to medicinal chemists, a Cbz-

72

- em
- ph

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

att protected dipeptide (**40g**) was successfully transformed into the C-terminal benzimidazole ent derivative (**41g**) in good yield as well. This demonstrates the utility of HBTU promoted ion method for the synthesis of peptide-based benzimidazoles for drug discovery. wit



Scheme 4.2 Synthesis of alkyl benzimidazoles.

or During our investigation, we realized that indole-2-carboxylic acid is a privileged motif use and the corresponding benzimidazole is widely used in drug discovery efforts.<sup>25,195</sup> We thi successfully synthesized various indole-2-benzimidazoles (**43a–43f**), where the aryl ring s of the benzimidazole is substituted with different functional groups (**Scheme 4.3**). We spa envisioned that having halogen substitution on the aryl-ring provides a useful chemical ce handle for further structure diversification *via* metal-catalyzed cross-coupling reactions. to We also observed that electron rich 1,2-diaminobenzene derivatives yielded better yield em

73

ph

asi

reat quote from he document or att se this space to mphasize a key ion oint. To place his text box nywhere on the age, just drag

der

's

t.]

than those that are electron deficient. This may be due to change in nucleophilicity of the ent diaminobenzene.





use

To further validate the HBTU promoted method, we proceeded to synthesize an extensive thi library of alpha-amino acid derived benzimidazoles. There are two reasons for this S endeavor: (i) we wanted to access a structurally diverse collection of alpha-amino spa benzimidazoles from commercially available amino acids with suitable protecting groups; ce and (ii) we envisioned accessing alpha-amino acids precursors for the synthesis of peptideto based benzimidazoles. This library includes thirteen Boc-protected amino acids (44a - 44n,em 74

ph

asi

der reat quote from he document or use this space to ent emphasize a key point. To place his text box h a unywhere on the page, just drag at

att Table 4.2) and three Cbz-protected amino acids (48 – 50, Figure 4.1). We reacted all Bocent amino acids with 1,2-diaminobenzene under optimized, one-pot reaction condition, and ion isolated the desired benzimidazoles (45a - 45n) in excellent yield. We also learned that wit many side chains protecting groups, including benzyl ether, benzyl thioether, and benzyl h a esters are stable to the reaction condition. Additionally, the side chain of Boc-Asn-OH (441) gre required no protecting group to generate the Boc-Asn derived benzimidazole 451. It is also at interesting to note that compound 45d is structurally similar to veliparib, a poly(ADPqu ribose)polymerase (PARP) inhibitor that is in clinical trials.<sup>138</sup>

ote

 $_{\rm fro}$  Table 4.2 Investigation of substrate scope for the synthesis of amino acid based  $_{\rm m}$  benzimidazoles.



great quote from the document or use this space to emphasize a key point. To place this text box mywhere on the page, just drag





S

his

t.]

One interesting observation was made when Boc-Gln-OH (44m) was reacted to form the spa corresponding benzimidazole. Absence of protecting group on the side chain yielded an ce interesting tricyclic structure (45m). Based on a literature precedent,<sup>196</sup> we propose that the to side chain amide underwent a transamidation reaction with the benzimidazole nitrogen, em

77

- ph
- asi
- ze

preat quote from the document or the document of the document

att generating the unique tricyclic product **45m**. Since compound **45m** has an amine handle, ent and a conformationally distinct tricyclic structure, it could be a useful synthon for ion medicinal chemistry efforts. We plan to explore the medicinal chemistry potential of wit benzimidazole **45m**.

gre We also utilized the HBTU promoted approach for the synthesis of two halogenated at analogues of Boc-tyrosine derived benzimidazoles (46 and 47), and three *N*-Cbz protected qu amino acid derived benzimidazoles (48 – 50). Using these substrates, we demonstrated that ote both Boc and Cbz carbamates are tolerated. Additionally, having a halogen handle on the fro aryl ring of benzimidazole provides a new venue to diversify the benzimidazole core for m medicinal chemistry purposes (**Figure 4.1**).



der

's

h a



thi

**Figure 4.1** Structures of halogenated and *N*-Cbz protected amino acid based <sup>s</sup> benzimidazoles.

<sup>spa</sup> It is reported that amino acid derived thiazoles have shown to be potent modulators of P-<sup>ce</sup> glycoprotein, which contributes to drug resistance in cancer cells.<sup>197</sup> Since the amino acid <sup>to</sup> benzimidazoles we have generated, including **46**, **47**, **48** and **50** are isosteres of reported em

78

ph

ze

preat quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

t.]

der

's

h a

att thiazole derivatives, these compounds can be evaluated for potential P-glycoprotein ent binding affinity, and reversal of anti-cancer drug resistance. The proposed synthetic ion approach provides convenient entry to prepare and evaluate potentially bioactive Pwit glycoprotein modulators.

 $_{\rm gre}$  4.1.4 A plausible mechanism for the HBTU promoted synthesis of benzimidazoles

In addition to HBTU (52) being an effective coupling agent, we believe that HBTU is at qu playing a very important role in the cyclization process. We propose a plausible ote mechanistic pathway, which may explain the HBTU-promoted formation of the fro benzimidazole (Scheme 4.4). The intermediate aryl-amide (51) is relatively stable, and for it to undergo dehydration, it needs to be activated. We propose that HBTU helps in the m the activation of amide, where the oxygen atom of the amide reacts with the carbodiimide do motif first. Following the attack of the amide oxygen, a molecule of 1hydroxybenzotriazole (HOBt, 55) is lost from HBTU. In the subsequent step, the second cu aryl-amine motif (54) reacts to kick-out a molecule of tetramethylurea (56) and forms the me desired benzimidazole (14). Based on LC-MS and <sup>1</sup>H-NMR analyses, we confirmed the nt formation of the key by-products, HOBt (55) and tetramethylurea (56) during the or use conversation of amide substrate into benzimidazole. This finding provides experimental thi support for this mechanistic proposal.

S

spa

ce

to

em

.

ph asi

der reat quote from he document or use this space to ent ion point. To place his text box ha unywhere on the page, just drag at t.]



Scheme 4.4 Plausible mechanism of HBTU promoted cyclization. me

## nt 4.2 A second library of aryl-benzimidazoles synthesized using the synthetic or methodology developed in our lab

<sup>use</sup> Using our new synthetic method, we synthesized a second library of benzimidazoles that <sup>thi</sup> are highly useful for our medicinal chemistry efforts. During this effort, we utilized the <sup>s</sup> ability of HBTU to perform the dehydrative cyclization of the amide precursor to access <sup>spa</sup> the benzimidazoles. Using simple aryl carboxylic acid derived amide precursor **63a-63r** <sup>ce</sup> under an optimized condition, we successfully synthesized eighteen structurally distinct <sup>to</sup> benzimidazoles **64a-64r** in high yields (**Figure 4.2**). The presence of electron withdrawing <sup>em</sup> or electron donating groups on the aryl ring didn't have effect on the yield. Nowadays

ph

asi

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

att much attention has been focused on substituted 2-aryl benzimidazoles as therapeutic ent agents,<sup>198</sup> and with the use of this synthetic methodology, this type of aryl benzimidazoles ion would be easily accessible.



to Figure 4.2 Synthesis of aryl-benzimidazoles.

em

 $\mathbf{p}\mathbf{h}$ 

ze

asi

81

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

at

att Since indole based benzimidazoles is of great interest to medicinal chemists,<sup>110,199,200</sup> we ent proceeded to explore substrate scope for the synthesis of this motifs from indole amides. ion The synthesis of indole based benzimidazoles from indole amide *o*-phenylenediamine wit substrates (**Figure 4.3**) went on to afford excellent yields, **66a** – **66e**. The amide positions h a were varied on the indole core structure and these positions didn't influence the yields in gre the transformation of amide substrates to indole benzimidazoles.



use Figure 4.3 Synthesis of indole-based benzimidazoles.

thi

s 2-substituted benzimidazoles are reported to be pharmacologically active, we investigated spa the substrate scope for alkyl carboxylic acid derived amide substrates (**Figure 4.4**). The ce conversion of these substrates to benzimidazoles afforded high yields (**68a** – **68k**). Alkyl to chains with various substituents were explored with future investigation of the synthesized em compounds as potential therapeutic leads.



82

der reat quote from he document or use this space to ent emphasize a key ion ooint. To place his text box h a nywhere on the page, just drag at t.]



nt Figure 4.4 Synthesis of benzimidazoles from alkyl-amides.

# <sup>or</sup> 4.3 A Mild *N*-Alkylation Methodology for the Structure Diversification of use Indolyl-Benzimidazoles

We envisioned achieving selective reactivity for the N-alkylation based on the pKa thi differences between the indole and benzimidazole nitrogen in indole-benzimidazole hybrid S molecule (66a). Our investigation began with identifying a suitable base for the alkylation spa process. N-benzylation was performed to determine the reactivity of 2-indolylce benzimidazole 66a (Table 4.3). We selected three commonly used organic bases (1,8to Diazabicyclo[5.4.0]undec-7-ene (DBU), triethylamine (TEA) N,Nand em 83

ph

asi

preat quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

t.]

att Diisopropylethylamine (DIPEA)) and four commonly used inorganic bases (potassium ent carbonate (K<sub>2</sub>CO<sub>3</sub>), cesium carbonate (Cs<sub>2</sub>CO<sub>3</sub>), Sodium bicarbonate (NaHCO<sub>3</sub>), and ion potassium bicarbonate (KHCO<sub>3</sub>)) for our study. Our initial screening shows that all wit inorganic bases screened (Table 4.3, entries 5 - 8) are highly effective in selectively h a promoting the N-alkylation of indolyl-benzimidazoles. Under these conditions the gre alkylation occurred exclusively at the benzimidazole core. From the screening, we at observed that all the organic bases (Table 4.3, entries 2 - 4) yielded no conversion to 71a, qu perhaps due to their inability to deprotonate the benzimidazole or their reactivity towards ote benzyl bromide. As an additional component of screening, we also found that the reaction fro worked well in dimethylformamide (DMF) or acetonitrile (CH<sub>3</sub>CN).

m

der

's

the	Table 4.3	Screening	of different	conditions	for the	N-alkylation.
-----	-----------	-----------	--------------	------------	---------	---------------

do cu		BnBr (2 equiv) Base (3 equiv) ───── CH <sub>3</sub> CN, rt	
me	66a		71a Ph
nt			
or	Entry	Base	Yield (%)
	1	No base	No reaction
use	2	DBU	No reaction
thi	3	TEA	No reaction
S	4	DIPEA	No reaction
spa	5	$K_2CO_3$	90
ce	6	$Cs_2CO_3$	92
to	7	NaHCO <sub>3</sub>	86
10	8	KHCO <sub>3</sub>	85
em ——			0.4
			84

ph

asi

Ы

der
's
att
<b>4.3.1 Substrate scope evaluation of selective</b> <i>N</i> <b>-alkylation strategy</b>
Upon identifying a method to selectively alkylate the benzimidazole motif within the
indolyl-benzimidazole scaffold, we evaluated the substrate scope of this selective <i>N</i> -wit
alkylation strategy. Our initial exploration yielded a small library of <i>N</i> -benzyl
benzimidazole derivatives (Scheme 4.5). Nine different commercially available benzyl
bromides were reacted with indolyl-benzimidazole <b>66a</b> in the presence of $K_2CO_3$ (3 equiv.)
in CH <sub>3</sub> CN. Since a non-aqueous condition is used for this chemistry, we have used 3
equivalents of the base to ensure enough amount is dissolved in the reaction medium for ote
the alkylation chemistry to proceed to completion. It is worthwhile to highlight that even fro
with 3 equivalents of K <sub>2</sub> CO <sub>3</sub> , the alkylation is highly selective towards the benzimidazole
motif, providing preferential reactivity. The benzyl bromides used in this study have either the
electron withdrawing or electron donating groups at the para-position. We were delighted do
to see that all benzyl bromides yielded the desired products $(71a - 71j)$ in very high yield cu
(79-92%), and the electronic nature of the aryl ring had little effect on the reactivity.
nt
or
use
thi
S
spa
ce
to
em
ph 85
asi

der reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag t.]

's



use Scheme 4.5 Synthesis of *N*-benzylated derivatives of indolyl-benzimidazole using benzyl bromides. thi

As a next step, we further evaluated the utility of this method by alkylating 66a with five S spa different aliphatic bromides (Scheme 4.6). Despite their reduced reactivity compared to benzyl halides, we were pleasantly surprised that the alkylation reaction went smoothly to ce afford the desired products (72a - 72f) in excellent yield (88% - 94%). The alkyl donors to em vary from a simple allyl group (72d) to structurally complex cycloalkyl groups (72a, 72b 86 ph

asi

great quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

wit

att and 72c) and a long aliphatic ether (72f). The allylated product (72d) is a useful synthon ent for further derivatization *via* olefin chemistry to generate new analogs for medicinal ion chemistry.





<sup>or</sup> To study the effect of substitution on the indole-motif, we synthesized a series of alkylated <sup>use</sup> analogs using 5-methoxyindolylbenzimidazole (**43a**) as substrate (**Scheme 4.7**). <sup>thi</sup> Modification of the benzimidazole nitrogen of **43a** went smoothly under the optimized <sup>s</sup> condition. Both benzyl-derivatives (**73a** and **73b**) and aliphatic derivatives (**73c**, **73d** and <sup>spa</sup> **73e**) afforded the desired product in excellent yield (78-96%). Although the 5- <sup>ce</sup> methoxyindole unit is electronically activated in comparison to the unsubstituted indole <sup>to</sup> ring, we did not observe any trace of alkylation of the 5-methoxyindole nitrogen under the em

87

ph

asi

great quote from the document or use this space to emphasize a key point. To place this text box enywhere on the page, just drag

der

's

ion

t.]

att reported condition. This further validates that K<sub>2</sub>CO<sub>3</sub> is a very mild and highly selective ent base for the alkylation of the benzimidazole nitrogen.





use

## 4.3.2 Synthesis of symmetrical bis-*N*,*N*-alkylated 2-indolyl-benzimidazole

thi

With the successful establishment of selective *N*-alkylation of indolyl-benzimidazole, we proceeded to explore a method to alkylate the indole nitrogen. We evaluated the ability of potassium hydroxide (KOH) to assist with the alkylation of **66a**. Since KOH is a relatively estronger base than the carbonate or bicarbonate, the benzylation of **66a** is tested at 0 °C to give us a mild reaction condition. With this modified protocol, we successfully alkylated em

ph

ze

der reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag t.]

's





use



S To advance the synthetic utility of this strategy, we developed a stepwise, one-pot, selective spa method to alkylate 66a using two different alkyl bromide donors. The benzimidazole ce nitrogen in compound 66a was successfully alkylated first with an alkyl bromide, and to subsequently the indole nitrogen in 66a was alkylated with a different alkyl bromide, in a em



- asi
- ze

der reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag t.]

's

att stepwise fashion and one-pot operation (Scheme 4.9). Using this strategy, we achieved the ent selectivity and synthesized asymmetrically alkylated 2-indolylbenzimidazoles 77a - 77d. ion From a medicinal chemistry stance, this approach enables one to quickly diversify the wit indolyl-benzimidazole core to generate novel analogs for drug discovery efforts. h a



90

ze

preat quote from the document or ase this space to emphasize a key point. To place this text box anywhere on the bage, just drag

der

's

att

ent

#### 4.4 Synthesis of Bis-Benzimidazole derivatives

We used our reported HBTU promoted methodology to synthesize a series of bision benzimidazoles (Scheme 4.10). We used commercially available bis-carboxylic acid wit containing substrate linkers (78a and 78b) which underwent one pot synthesis to give the h a corresponding bis-benzimidazole products in good yields (85-93%). gre



nt

Scheme 4.10 Synthesis of bis-benzimidazole derivatives from different type of bis-dicarboxylic or acid linkers and *o*-phenylenediamine derivatives.

use

thi

We started our first strategy of synthesizing bis-benzimidazole derivatives by taking spa commercially available chelidamic acid monohydrate **78a** (Scheme 4.10), which underwent condensation-dehydration reaction with substituted *o*-phenylenediamine to vield the corresponding bis-benzimidazole derivatives **79a** - **79f** (Figure 4.5) in higher em yields of 80 - 95%. Having substitution on the benzene rings was to help us in

91

ph

ze

reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag t.]

der

's

att understanding the effect of these diverse substituent groups in modulating the ent pharmacological properties of bis-benzimidazole derivatives. It is reported that having ion substitution on the benzene rings improves the pharmacological activities,<sup>201</sup> and in wit addition, the substituents used are potential handles for further modifications. h a



spa Figure 4.5 Synthesized bis-benzimidazole derivatives using chelidamic acid monohydrate linker. ce

We proceeded to carry out the second strategy by using 2,6-pyridine dicarboxylic acid 78b to em (Scheme 4.10) to synthesize substituted bis-benzimidazole derivatives 80a – 80f (Figure 92 ph

asi

preat quote from the document or use this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

t.]

att 4.6). The only difference between the first and second strategy is the linker used and the ent aim was to evaluate the effect of the linkers in modulating physicochemical and ion pharmacological properties. In addition, the bis-benzimidazole derivatives 80a – 80f and wit their metal complexes are reported as potassium ion channel modulators<sup>202</sup>, as light h a emitting materials,<sup>203–205</sup> pestivirus inhibitors,<sup>206</sup> and apoptosis inducers in cancer cell gre lines.<sup>207,208</sup>



to **Figure 4.6** Synthesized bis-benzimidazole derivatives using 2,6-pyridine dicarboxylic acid em

93

ph

asi
reat quote from he document or se this space to ent mphasize a key oint. To place his text box nywhere on the age, just drag t.]

der

's

att

# 4.5 Synthesis of substituted phenylbenzimidazoles

We used our reported HBTU promoted methodology to synthesize a library of substituted ion phenylbenzimidazoles (Scheme 4.11 and Scheme 4.12). Commercially available benzoic wit acids with varying substitution patterns at 3, 4 and/or 5 positions were coupled with 1,2h a diaminobenzene and cyclized to form substituted phenyl benzimidazoles. gre



94

asi

der rreat quote from he document or att use this space to ent mphasize a key boint. To place his text box h a mywhere on the page, just drag at t.]



Scheme 4.12 Synthesis of 3-substituted phenyl benzimidazoles thi

### <sup>s</sup> 4.6 Biological Evaluation of benzimidazoles

<sup>spa</sup> 4.6.1 Evaluation of indole-based benzimidazoles for anti-cancer activity

<sup>ce</sup> MTT assay was done to evaluate the effect of selected indole-based benzimidazoles on cell
 <sup>to</sup> viability. There has been limited exploration of 2-indolylbenzimidazole derivatives as
 <sup>em</sup> potential chemotherapeutic leads. Our results indicate that **43a** (unsubstituted on the 95

ze

der reat quote from 's he document or att benzimidazole ring), 43b, 43c, and 43d (halogenated on the benzimidazole ring), and 43e se this space to ent (cyano group on the benzimidazole ring) showed excellent cytotoxicity with IC<sub>50</sub> values mphasize a key ion oint. To place between  $1.8 - 34 \mu M$  (Table 4.4). HepG2 cells are known to express drug metabolizing wit enzymes in high amount.<sup>209</sup> Thus, higher IC<sub>50</sub> values observed for 43d, 43e, and 43f in text box h a nywhere on the HepG2 cells could be explained by their metabolic sensitivity. Based on the IC<sub>50</sub> values, gre age, just drag all indolylbenzimidazoles showed slight selectivity towards the lung cancer cell line at (A549), compared to other cell lines tested. qu

ote Table 4.4 Data of Inhibition of Cancer Cell Proliferation for 2-indolylbenzimidazole derivatives 43a – 43f.

m				IC <sub>50</sub> (µM)	
the	Compound	Structure	HeLa	A549	HepG2
do cu	43a	MeO N H	25.4 ± 6.1	3.8 ± 1.1	8.7 ± 3.4
me nt	43b		18.1 ± 4.6	15.6±4.6	23.8±5.7
or	43c		11.1 ± 2.0	$4.0 \pm 2.4$	9.1 ± 1.0
use thi	43d		$5.8\pm0.9$	$1.8 \pm 0.6$	19.0±5.3
s spa	43e		$11.8 \pm 4.5$	$5.9 \pm 2.8$	34.0±10.1
ce to	43f	MeO	31.0±13.0	29.6±5.0	80.0±30.6
em					
ph					96
asi					

ze

fro

his

t.]

der reat quote from he document or use this space to ent emphasize a key point. To place his text box h a unywhere on the page, just drag at

## 4.6.2 Evaluation of lipid-based benzimidazoles as potential anti-cancer agents ent

Addition of lipid motifs to drug leads improves their activity, hence we investigated ion different chain length containing lipid-based benzimidazoles.<sup>210,211</sup> Lipid-based wit benzimidazole derivatives **41c - 41e** were potent in cancer cells with the lowest IC<sub>50</sub> value h a of 1.5  $\mu$ M (**Table 4.5**). The results indicate that the long lipid motif increases the antigre cancer activity as compounds **41c** (C<sub>10</sub>), **41d** (C<sub>12</sub>), and **41e** (C<sub>14</sub>) showed high anti-cancer at activity whereas compound **68c** (C<sub>8</sub>) was inactive. Compounds **68a** and **68b** were not tested because they were too hydrophobic and could not be solubilized in DMSO (**Table 4.5**).

ote

fro	Table	4.5	Data	of	Inhibition	of	Cancer	Cell	Proliferation	for	lipid-based
m	benzin	nidaz	oles								

_				IC <sub>50</sub> (µM)	
the Compo	ound	Structure	HeLa	A549	HepG2
do 410 cu	c C <sub>s</sub>	H <sub>17</sub> N H	$12.7 \pm 2.5$	$6.5\pm2.5$	$14.7 \pm 2.7$
me 410 nt	i C <sub>1</sub>	N N H H	$12.5 \pm 2.9$	$1.5 \pm 0.4$	7.3 ± 1.1
or 410 use	e C <sub>1</sub>	N 2H <sub>25</sub> N H	$16.0 \pm 5.7$	$1.7 \pm 1.0$	$5.0 \pm 1.7$
thi <b>68</b> a s	a C <sub>1</sub>	N N H <sub>33</sub> N H	NT	NT	NT
<sup>spa</sup> 681	<b>b</b> C <sub>1</sub>	AH29 NH	NT	NT	NT
to <b>68</b> 0	c C <sub>e</sub>	H <sub>13</sub> N H	>100	>100	>100
em					

97

ph

asi

reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

der

's

att

4.6.3 HO-PI Assay

t.]

ze

asi

ent Based on the MTT cell viability assay results, experiments were performed to determine ion the cell death mechanism for selected compounds from our library, representing each class wit of synthesized benzimidazoles at their IC<sub>50</sub> values such as indole based **43c**, **43e**, and alkylh a based 68c, 41c. Many anti-cancer drugs and experimental benzimidazole compounds are gre known to cause apoptosis.<sup>212,213</sup> Apoptosis was assayed using the double staining of at Hoechst 33342 (HO) and propidium iodide (PI) dye and observation by fluorescence qu microscope. The blue fluorescent HO is a cell permeable nucleic acid dye that shows bright ote fluorescence in apoptotic cells due to chromatin condensation. The red-fluorescent fro propidium iodide is a cell impermeable DNA-binding dye, which can only stain the cells m in situations where there is loss of plasma membrane integrity such as necrosis. Etoposide, the an anticancer agent known to cause apoptosis, was used as a positive control in the assay do <sup>214</sup>. The control cells appeared to be intact and the nuclei were stained with a less bright cu blue fluorescence, and the absence of red fluorescence also indicated regular, intact cells. me Cells treated with select benzimidazole compounds exhibited bright blue fluorescence nt denoting apoptotic cells (white arrows, Figure 4.7). Necrotic cells show bright red nuclei. or We found that most of the selected compounds induced apoptotic cell death from the HOuse PI assay data (Figure 4.7). Compounds 43e, 68c, and 41c showed significant apoptosis thi which correlates to the cell viability assay. Thus, apoptosis was observed as a primary S mechanism of cell death for the majority of benzimidazole compounds tested. spa ce to em ph

der reat quote from 's he document or att se this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the gre age, just drag at t.] qu



use

ote

fro

m

the

do

cu

me

nt

or

thi Figure 4.7 Apoptosis/necrosis assay using fluorescence microscopy. Representative images of HeLa cells stained with HO33342 and PI dyes after 24h of treatment with benzimidazole derivatives obtained by fluorescence microscopy (200X magnification). White arrows indicate bright blue apoptotic cells.



to

em

ph

asi

	der
reat quote from	uei
-	's
he document or	
se this space to	att
	ent
mphasize a key	ion
oint. To place	1011
	wit
his text box	
1 .1	h a
nywhere on the	ara
age, just drag	gic
	at
t.]	
	qu
	ote

4.6.4 Evaluation of bis-benzimidazole derivatives for anti-cancer activity
ent
The synthesized bis-benzimidazole derivatives $79a - 79f$ and $80a - 80e$ were evaluated for ion
anti-cancer activity against HeLa, and MDA-MB231 cancer cell lines (Table 4.6, Figur wit
<b>4.8</b> ) in order to determine their cytotoxicity. Doxorubicin was used as a positive control h a
Bis-benzimidazole <b>80e</b> was not tested since it was insoluble in DMSO. Overall the result
indicated that bis-benzimidazole derivatives $79a - 79d$ , and $80a - 80d$ were promising
anti-proliferative agents with IC <sub>50</sub> values between 2.72 – 28.53 $\mu$ M ( <b>Table 4.6</b> ). Bis
u benzimidazoles <b>79e</b> , <b>79f</b> and <b>80f</b> were inactive which may be attributed to the effect of
substituent groups on the benzene rings (Table 4.6). Compound 79a had better activit
fro compared to its analogs, <b>79b</b> - <b>79c</b> , implying that having substitution on the benzine ring
m has little effect in improving the activity, except for <b>79d</b> whose activity was within th
the range with <b>79a</b> . Having fluorine and chlorine substitution on <b>80a</b> , improves the activity
do Compounds 80b and 80c showed improved activity against the cancer cell line teste
cu compared to <b>80a</b> . Compound <b>80d</b> had comparable activity with <b>80a</b> , hence having bromin
me as a substituent on the benzene ring has little effect in improving the activity
nt
or
use
thi
S

- spa
- ce
- to
- em
- ph

asi

ze

..... great quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

der

's

t.]

att Table 4.6 Data of Inhibition of Cancer Cell Proliferation for bis-benzimidazole ent derivatives

ion			
•.		IC <sub>50</sub>	(µM)
wit	Compound	HeLa	MDA-MB231
h a	79a	$480 \pm 0.98$	6 38 + 0 61
gre	//u	1.00 ± 0.90	0.50 ± 0.01
at	79b	$28.53\pm9.64$	$40.57\pm3.07$
at	79c	$23.52\pm3.48$	$17.44\pm2.38$
qu	79d	8 89 + 2 02	10 25 + 2 52
ote	r / u	0.09 - 2.02	10.20 - 2.02
fro	<b>79e</b>	NA	NA
	<b>79f</b>	NA	NA
m	80a	$24.51 \pm 3.54$	$18.74 \pm 4.70$
the			
do	80b	$2.72 \pm 0.50$	$4.62 \pm 0.91$
	80c	$6.97\pm0.87$	$8.84 \pm 0.36$
cu	80d	$18.80 \pm 2.53$	$24.66 \pm 3.21$
me	000		
nt	80f	NA	NA
	Doxorubicin	$0.08\pm0.01$	$0.11\pm0.02$
or			

use thi S spa ce to

em

ph

asi

der reat quote from he document or use this space to ent emphasize a key boint. To place his text box h a nywhere on the page, just drag at t.]



do Figure 4.8 Cell viability graphs of 79a, 80b and 80c against HeLa and MDA-MB231 tumor cell cu lines.

me

#### nt 4.6.5 Evaluation of bis-benzimidazole derivatives as topoisomerase II inhibitors

or Human DNA topoisomerases has been recognized as an attractive target for developing use anti-cancer drugs.<sup>188</sup> Two types of topoisomerase exist, namely, type I topoisomerase thi (Topo I) and type II topoisomerase (Topo II).<sup>215</sup> Topo I, introduces single-strand breaks in s DNA, whereas Topo II introduces double-strand breaks and requires ATP for full spa activity.<sup>188</sup> Both isomers are nuclear enzymes that are crucial in resolving topological ce challenges that occur during DNA transcription, replication, and chromosome to segregation.<sup>216</sup>

102

em

ph

ze

preat quote from the document or ase this space to emphasize a key point. To place this text box anywhere on the page, just drag

t.]

der

's

att

Topo II is the specific target of some of the most active anti-cancer drugs such as etoposide, ent doxorubicin, mitoxantrone, amonafide, and amsacrine.<sup>217</sup> However, Topo II inhibitors have ion some therapeutic limitations because of their serious side effects during cancer wit chemotherapy.<sup>218</sup> Thus, development of new anti-cancer Topo II inhibitors is necessary for h a improving cancer treatment.<sup>187</sup> Several benzimidazole derivatives are reported as novel gre Topo II inhibitors.<sup>188,189</sup> Topo II-mediated DNA relaxation assay was performed for bisat benzimidazole **79a**. The work is done in collaboration with the Low lab and performed qu with the assistance of Shilpa. To determine Topo II inhibitory activity of bis-benzimidazole ote 79a, doxorubicin was used as a positive control and the assay done at different fro concentration of 10, 20 and 30 µM for both bis-benzimidazole 79a (Figure 4.9A and 4.9B) m and doxorubicin. the



**Figure 4.9A** Topo II agarose gel assay results. Lane 1 is decatenated DNA (control), Lane 2 is Linear DNA (control), Lane 3 is kinetoplast DNA (kDNA-control), Lane 4 is kDNA with Topo II enzyme, Lane 5 is kDNA with enzyme and DMSO (negative control), Lane 6 is kDNA with enzyme and doxorubicin (10  $\mu$ M) positive control, Lane 7 corresponds to kDNA with enzyme and bis-benzimidazole 79a (30  $\mu$ M), Lane 8 corresponds to kDNA with enzyme and bis-benzimidazole 79a (20  $\mu$ M) and Lane 9 corresponds to kDNA SPa with enzyme and bis-benzimidazole 79a (10  $\mu$ M). **Figure 4.9B** Topo II agarose gel assay results for bis-benzimidazole 79a and doxorubicin at 20  $\mu$ M. Lane 1 is decatenated DNA (control), lane 2 is linear DNA (control), lane 3 is kinetoplast DNA (kDNA-control), lane 4 is kDNA with Topo II enzyme, lane 5 is kDNA with enzyme and DMSO (negative control), lane 6 is kDNA with enzyme and doxorubicin (20  $\mu$ M)-positive control and lane 7 corresponds to kDNA with enzyme and bis-benzimidazole 79a (20  $\mu$ M).

em

preat quote from the document or ase this space to emphasize a key point. To place this text box anywhere on the page, just drag

der

's

att

t.]

ent not bis-benzimidazole 79a. At 20 µM, there was complete inhibition for both bision benzimidazole 79a and doxorubicin (Figure 4.9A and 4.9B). Top II assay was repeated at wit  $20 \,\mu\text{M}$  to ascertain the results (Figure 4.9B). We hypothesize that since bis-benzimidazole h a derivatives 79d, 80b and 80c are analogs of compound 79a, have similar Topo II inhibition gre potential properties. Bis-benzimidazole derivatives are reported to mediate Topo I and II at activity<sup>219,220,220-228</sup> and their ability to interact with DNA,<sup>229,230</sup> have made them an qu effective precursors for a wide range of drugs targeting DNA and DNA related processes. ote Bis-benzimidazole derivative 80a is reported to have a wide range of properties such as fro binding to DNA,<sup>231-234</sup> an efficient anion receptor,<sup>235</sup> as a luminescent complex m ligand,<sup>203,236</sup> a receptor for urea recognition<sup>237,238</sup>, a chemo-sensor for ions,<sup>239-241</sup> a the coordinating ligand,<sup>204,242-247</sup> induces cancer cell apoptosis,<sup>207,248</sup> detects toxic benzene do metabolites,<sup>249</sup> as a potent and selective inhibitor of small conductance calcium-activated cu potassium channels (SK),<sup>202</sup> and as a potent transmembrane anion transporter.<sup>250</sup> me

After performing Topo II assay, at 10  $\mu$ M, there was slight inhibition by doxorubicin but

### nt 4.6.6 Evaluation of aryl benzimidazoles as BMPs agonists.

### or 4.6.6.1 MT-Glo Assay

use The synthesized aryl benzimidazole derivatives (**Scheme 4.11** and **4.12**) were evaluated thi for cell cytotoxicity using MT-Glo assay in C2C12 cell lines (**Table 4.7** and **Table 4.8**). s Compounds targeting BMPs pathways are supposed to be non-toxic to the cells and this spa explains why carrying out cell cytotoxicity test is important. MT-Glo assay results indicates ce that majority of the compounds were non-toxic to cells. Compounds **841** and **84n** (**Table 4.8**) were cytotoxic to the cells and there deemed not good to be evaluated for further BMP

em

ph

104

ze

great quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

t.]

der

's

ze

 $h_a$  C2C12 cell lines

att tests. Compounds 82c, 84a, 84h, and 84i had IC<sub>50</sub> values in the range of 29 - 40 µM (Table ent 4.7 and Table 4.8), not too toxic to prevent them from further BMP testing. ion

wit Table 4.7 MT-Glo analysis for cell cytotoxicity of the screened compounds against

Compound	Structure	IC <sub>50</sub> (µM)
82a		>100
82b		>100
82c		29.24
82d		>100
82e		70.2
QJE		>100
821	OMe OMe	~100
82g	И С С С С С С С С С С С С С С С С С С С	66.85
82h	C N N OH	>100
	ОН	
	Compound         82a         82b         82c         82d         82d         82e         82f         82g         82h	CompoundStructure82a $( ) = ( ) $

der great quote from he document or use this space to ent smphasize a key ion point. To place his text box h a nywhere on the page, just drag at t.]

asi

Table	4.8 MT Glo cell cy	totoxicity test for 3-substitu	ted phenyl benzimidazoles
ent	Compound	Structure	IC <sub>50</sub> (µM)
ion	940	HO H	40.14

wit	042		40.14
h a gre	84b	Br H N	93.29
at			
qu	84c		>500
ote	84d	H <sub>3</sub> C	>500
fro	07u		~ 500
m	810	O <sub>2</sub> N H	34 58
the	040		54.56
do	<b>8</b> /1f	F H	>500
cu	041	N	~ 500
		<b>.</b>	
me	<b>84</b> σ		>1000
me nt	84g	CI H N	>1000
me nt or	84g		>1000
me nt or use	84g		>1000
me nt or use thi	84g		>1000
me nt or use thi s	84g		>1000
me nt or use thi s spa	84g		>1000
me nt or use thi s spa ce	84g		>1000
me nt or use thi s spa ce to	84g		>1000
me nt or use thi s spa ce to em	84g		>1000
me nt nt or use thi s spa ce to em ph	84g		>1000

der reat quote from 's he document or att se this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the gre age, just drag at t.]

ent	Compound	Structure	IC <sub>50</sub> (µM)
ion wit	84h	H <sub>3</sub> CS N	36.53
h a gre	84i	H <sub>3</sub> CO	38.09
at qu	84j	F <sub>3</sub> C N	>100
ote fro	84k	H <sub>3</sub> C H <sub>3</sub> C H <sub>3</sub> C	>500
m the	841	H <sub>3</sub> COOC	4.38
do cu me	84m	H <sub>3</sub> C-S	>100
nt or use	84n		13.23

thi

#### 4.6.6.2 Smad-Phosphorylation Assay S

spa After performing cell cytotoxicity assay, compounds were selected for smadphosphorylation experiments to determine their contribution to BMPs signaling pathways. ce The expression of smads were assessed using Immunofluorescent labelling and Western to em blotting experiments. The work is done in collaboration with the Perron lab and performed

107

ph

asi

great quote from	der 's
he document or	att
se this space to	by Dr. Perron. BMP7 was used as a positive control. Treatments of <b>82c</b> , <b>82e</b> , <b>84a</b> , and <b>84h</b> ent
mphasize a key	in the immunofluorescence labelling assay (Figure 4.10 and Figure 4.11) caused
oint. To place	translocation of pSmads into the nucleus. Compounds 82c and compounds 82e showed the wit
his text box	most increased Smads phosphorylation compared to the positive control BMP7 (Figure
nywhere on the	<ul> <li><b>4.11</b>). Among the 3-substituted phenyl benzimidazoles, 82g, 82h, 84a, 84b and 84i, only</li> </ul>
oage, just drag	gre compounds <b>82h</b> and <b>84a</b> showed to some extend Smad phosphorylation but lower than that at
t.]	of compounds 82c and 82e (Figure 4.10). Compound 82a had no substitution (Scheme
	<b>4.11</b> ) and didn't cause phosphorylation of Smads. Likewise, compound <b>82b</b> with a
	ote substitution at para position ( <b>Scheme 4.11</b> ) didn't cause phosphorylation of smads ( <b>Figure</b>
	<b>4.10</b> ). Di-substitution methoxy's ( <b>82c</b> and <b>82e</b> ) on aryl ring at positions 3 and 4 or at 3 and
	m 5 is crucial for pSmad activity than mono-substitution (82b and 84i, Scheme 4.11). We
	the took one of the most active compounds, <b>82e</b> , and performed western blot analysis.
	Compound <b>82e</b> increased the expression of pSmad in a concentration dependent manner
	cu and the results validates immunoblotting observations. Overall our results suggest that di-
	me methoxy-benzimidazoles <b>82c</b> and <b>82e</b> from a library of small molecules aryl
	nt benzimidazoles were identified as promising compounds for further evaluation
	or

use

thi

S

ce

spa

to

em

108

ph asi

der reat quote from he document or use this space to ent emphasize a key boint. To place his text box h a unywhere on the page, just drag at t.]



S

ce

spa

to

em

em

ph

asi

..... der great quote from he document or att se this space to mphasize a key oint. To place his text box h nywhere on the age, just drag t.]

's

ent		pSma	d	pSn	nad	DA	PI	pSmad	DAPI
ion	itrol					46	• Rectang	ular Ship	
wit	C01			÷:				1	
h a									
gre	MP7						Carta a		
at	B		G	-	6		<i>A</i>		
qu							10		
ote	82c						1		
fro									
m	ĺ	dille.							
the	82e	- 4698 	Zir.	1. ja	Altere			2.73	
do			24		1999 - C.				

cu Figure 4.11 Fluorescence imaging. Treatment of 82c and 82e in C2C12 cells caused me translocation of pSmad into the nucleus.

nt or use thi S spa ce to em ph asi

ze

reat quote from he document or use this space to mphasize a key oint. To place his text box nywhere on the age, just drag

der

's

att

ent

t.]

CONCLUSIONS

Benzimidazole is a privileged, and routinely used pharmacophore lead in the drug ion discovery process. We have reported two convenient and mild methodologies for the wit synthesis of benzimidazoles: (a) a dehydrative cyclization strategy for the synthesis of h a benzimidazoles using HBTU promoted approach from aryl/alkyl amide substrates and (b) gre an efficient one-pot conversion of carboxylic acids into benzimidazoles using HBTU at promoted approach. These synthetic approaches are high yielding, acid free and tolerates qu various common functional groups. We have also reported a simple and highly effective ote synthetic method to chemo-selectively alkylate the indolyl-benzimidazole scaffold using a fro wide range of benzyl and alkyl aliphatic bromides. These reported methods are more m convenient for the synthesis and structural diversification of benzimidazole derivatives and the

will enable medicinal chemists to explore the SAR of benzimidazole-based drug leads. do

We successfully used the reported synthetic methodologies to synthesize different classes cu me of benzimidazoles and selected a representative of each class for bioassay evaluation. nt Indole-based and lipid-based benzimidazoles were tested for cell viability and results revealed that they are potential anticancer agents. The primary mechanism for cancer cell or use death induced by the tested compounds is apoptosis. Bis-benzimidazole derivatives were thi also evaluated for anti-cancer and topoisomerase II inhibition properties. Bisbenzimidazoles 79a, 79b, 80a, and 80c showed promising anti-cancer activities against the S spa cell line tested. Compound **79a** is a potential Topo II inhibitor.

ce

We also tested aryl benzimidazoles against bone morphogenetic proteins (BMPs) signaling to pathways. Our results suggest that bis disubstituted methoxy compounds 82c and 82e from em

111

ph

ze

der reat quote from 's he document or att a library of small molecules aryl benzimidazoles are promising BMP receptor agonist se this space to ent where they stimulated downstream cascade canonical pSmad-signaling pathways in mphasize a key ion oint. To place C2C12 cells. The reported benzimidazole derivatives possess anti-cancer and bone wit morphogenetic protein (BMPs) agonist properties. Our findings suggest that further his text box h a development of these scaffolds could provide drug leads towards new chemotherapeutics nywhere on the gre age, just drag and agonists for BMP signaling pathway. at t.] qu ote fro m the do cu me nt or use thi S spa ce to em 112 ph asi

der reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

- t.]

's att REFERENCES ent Walsh, C. T. Nature Loves Nitrogen Heterocycles. *Tetrahedron Lett.* **2015**, *56* (23), (1)ion 3075-3081. wit Kaur, N. Review on the Synthesis of Six-Membered N,N-Heterocycles by Microwave (2)h a Irradiation. Synth. Commun. 2015, 45 (10), 1145–1182. gre (3) Asif, M. A Mini Review: Biological Significances of Nitrogen Hetero Atom at Containing Heterocyclic Compounds. Int. J. Bioorg. Chem. 2017, 2 (3), 146-152. qu

- Garuti, L.; Roberti, M.; Pizzirani, D. Nitrogen-Containing Heterocyclic Quinones: A (4)ote Class of Potential Selective Antitumor Agents. Mini Rev. Med. Chem. 2007, 7 (5), fro 481 - 489.
- m
- Kaur, N. Multiple Nitrogen-Containing Heterocycles: Metal and Non-Metal Assisted (5) the Synthesis. Synth. Commun. 2019, 49 (13), 1633–1658. do
- Thansandote, P.; Lautens, M. Construction of Nitrogen-Containing Heterocycles by (6) cu C-H Bond Functionalization. Chem. Eur. J. 2009, 15 (24), 5874–5883.
- me Vitaku, E.; Smith, D. T.; Njardarson, J. T. Analysis of the Structural Diversity, (7)nt Substitution Patterns, and Frequency of Nitrogen Heterocycles among U.S. FDA or Approved Pharmaceuticals: Miniperspective. J. Med. Chem. 2014, 57 (24), 10257use 10274.
- thi

Zhang, B.; Studer, A. Recent Advances in the Synthesis of Nitrogen Heterocycles via (8)S Radical Cascade Reactions Using Isonitriles as Radical Acceptors. Chem. Soc. Rev. spa **2015**, *44* (11), 3505–3521.

- ce
- Morgenthaler, M.; Schweizer, E.; Hoffmann-Röder, A.; Benini, F.; Martin, R. E.; (9) to Jaeschke, G.; Wagner, B.; Fischer, H.; Bendels, S.; Zimmerli, D.; Schneider, J.; em

113

- ph

asi

	der
reat quote from	•
he document or	S
	att
ise this space to	
mphasize a key	ent
	ion
oint. To place	
	wit
his text box	ha
nywhere on the	па
and just drag	gre
age, just diag	at
t.]	
	qu
	ote

Diederich, F.; Kansy, M.; Müller, K. Predicting and Tuning Physicochemical
Properties in Lead Optimization: Amine Basicities. ChemMedChem 2007, 2 (8)
1100–1115.

- (10) Anand, K.; Wakode, S. Development of Drugs Based on Benzimidazole Heterocycle:
   a Recent Advancement and Insights. *Int. J. Chem. Stud.* 2017, 5 (2), 350 362.
- (11) Sun, X.; Lv, X.-H.; Ye, L.-M.; Hu, Y.; Chen, Y.-Y.; Zhang, X.-J.; Yan, M. Synthesis at of Benzimidazoles via Iridium-Catalyzed Acceptorless Dehydrogenative Coupling.
   *Qu* Org. Biomol. Chem. 2015, 13 (27), 7381–7383.
- 01
- fro

(12) Bhattacharya, S.; Chaudhuri, P. Medical Implications of Benzimidazole Derivatives as Drugs Designed for Targeting DNA and DNA Associated Processes. *Curr. Med. Chem.* 2008, *15* (18), 1762 - 1777.

the

m

- (13) Hirashima, S.; Suzuki, T.; Ishida, T.; Noji, S.; Yata, S.; Ando, I.; Komatsu, M.; Ikeda, do
  S.; Hashimoto, H. Benzimidazole Derivatives Bearing Substituted Biphenyls as
  cu
  Hepatitis C Virus NS5B RNA-Dependent RNA Polymerase Inhibitors:
  me
  Structure–Activity Relationship Studies and Identification of a Potent and Highly
  nt
  Selective Inhibitor JTK-109. *J. Med. Chem.* 2006, 49 (15), 4721–4736.
- or (14) Chu, B.; Liu, F.; Li, L.; Ding, C.; Chen, K.; Sun, Q.; Shen, Z.; Tan, Y.; Tan, C.; Jiang, use Y. A Benzimidazole Derivative Exhibiting Antitumor Activity Blocks EGFR and thi HER2 Activity and Upregulates DR5 in Breast Cancer Cells. *Cell Death Dis*. **2015**, 6 s
  - (3), e1686–e1686.
- spa
- (15) Cevik, U. A.; Saglik, B. N.; Ozkay, Y.; Canturk, Z.; Bueno, J.; Demirci, F.; Koparal,
   A. S. Synthesis of New Fluoro-Benzimidazole Derivatives as an Approach towards to

- em
- ph
- asi
- ze

der reat quote from 's he document or att se this space to ent mphasize a key ion oint. To place his text box nywhere on the gre age, just drag

t.]

the Discovery of Novel Intestinal Antiseptic Drug Candidates. *Curr. Pharm. Des.* **2017**, *23* (15), 2276 - 2286.

- (16) Yadav, G.; Ganguly, S. Structure Activity Relationship (SAR) Study of wit Benzimidazole Scaffold for Different Biological Activities: A Mini-Review. *Eur. J.*h a *Med. Chem.* 2015, 97, 419–443.
- (17) Galal, S. A.; Abdelsamie, A. S.; Shouman, S. A.; Attia, Y. M.; Ali, H. I.; Tabll, A.;
  at El-Shenawy, R.; El Abd, Y. S.; Ali, M. M.; Mahmoud, A. E.; Abdel-Halim, A. H.;
  qu Fyiad, A. A.; Girgis, A. S.; El-Diwani, H. I. Part I: Design, Synthesis and Biological ote Evaluation of Novel Pyrazole-Benzimidazole Conjugates as Checkpoint Kinase 2
  fro (Chk2) Inhibitors with Studying Their Activities Alone and in Combination with m Genotoxic Drugs. *Eur. J. Med. Chem.* 2017, *134*, 392–405.
- the
- (18) Adegboye, A. A.; Khan, K. M.; Salar, U.; Aboaba, S. A.; Kanwal; Chigurupati, S.;
  do Fatima, I.; Taha, M.; Wadood, A.; Mohammad, J. I.; Khan, H.; Perveen, S. 2-Aryl cu Benzimidazoles: Synthesis, In Vitro α-Amylase Inhibitory Activity, and Molecular me Docking Study. *Eur. J. Med. Chem.* 2018, *150*, 248–260.
- nt (19) Shin, Y.; Suchomel, J.; Cardozo, M.; Duquette, J.; He, X.; Henne, K.; Hu, Y.-L.; or Kelly, R. C.; McCarter, J.; McGee, L. R.; Medina, J. C.; Metz, D.; San Miguel, T.; use Mohn, D.; Tran, T.; Vissinga, C.; Wong, S.; Wannberg, S.; Whittington, D. A.; thi Whoriskey, J.; Yu, G.; Zalameda, L.; Zhang, X.; Cushing, T. D. Discovery, Optimization, and in Vivo Evaluation of Benzimidazole Derivatives AM-8508 and AM-9635 as Potent and Selective PI3Kδ Inhibitors. *J. Med. Chem.* 2016, *59* (1), 431–ce 447.

115

- to
- em
- em
- ph
- ze

der reat quote from 's he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag at t.]

> qu ote fro

att (20) Tatani, K.; Hiratochi, M.; Kikuchi, N.; Kuramochi, Y.; Watanabe, S.; Yamauchi, Y.; ent Itoh, F.; Isaji, M.; Shuto, S. Identification of Adenine and Benzimidazole Nucleosides ion as Potent Human Concentrative Nucleoside Transporter 2 Inhibitors: Potential wit Treatment for Hyperuricemia and Gout. J. Med. Chem. 2016, 59 (8), 3719–3731. h a (21) Lapierre, J. M.; Eathiraj, S.; Vensel, D.; Liu, Y.; Bull, C. O.; Cornell-Kennon, S.; gre

- Iimura, S.; Kelleher, E. W.; Kizer, D. E.; Koerner, S.; Makhija, S.; Matsuda, A.; Moussa, M.; Namdev, N.; Savage, R. E.; Szwaya, J.; Volckova, E.; Westlund, N.; Wu, H.; Schwartz, B. Discovery of 3-(3-(4-(1-Aminocyclobutyl)Phenyl)-5-Phenyl-3 H -Imidazo[4,5- b |Pyridin-2-Yl)Pyridin-2-Amine (ARQ 092): An Orally Bioavailable, Selective, and Potent Allosteric AKT Inhibitor. J. Med. Chem. 2016, 59 (13), 6455-6469.
- the

m

(22) Muth, A.; Subramanian, V.; Beaumont, E.; Nagar, M.; Kerry, P.; McEwan, P.; do Srinath, H.; Clancy, K.; Parelkar, S.; Thompson, P. R. Development of a Selective cu Inhibitor of Protein Arginine Deiminase 2. J. Med. Chem. 2017, 60 (7), 3198–3211. me (23) Hoyt, S. B.; Park, M. K.; London, C.; Xiong, Y.; Tata, J.; Bennett, D. J.; Cooke, A.; nt Cai, J.; Carswell, E.; Robinson, J.; MacLean, J.; Brown, L.; Belshaw, S.; Clarkson, or T. R.; Liu, K.; Liang, G.-B.; Struthers, M.; Cully, D.; Wisniewski, T.; Ren, N.; Bopp, use C.; Sok, A.; Cai, T.-Q.; Stribling, S.; Pai, L.-Y.; Ma, X.; Metzger, J.; Verras, A.; thi McMasters, D.; Chen, Q.; Tung, E.; Tang, W.; Salituro, G.; Buist, N.; Kuethe, J.; S Rivera, N.; Clemas, J.; Zhou, G.; Gibson, J.; Maxwell, C. A.; Lassman, M.; spa McLaughlin, T.: Castro-Perez, J.: Szeto, D.: Forrest, G.: Haidu, R.: Rosenbach, M.: ce Ali, A. Discovery of Benzimidazole CYP11B2 Inhibitors with in Vivo Activity in to Rhesus Monkeys. ACS Med. Chem. Lett. 2015, 6 (5), 573-578. em

116

- ph

- ze

der reat quote from 's he document or att se this space to mphasize a key oint. To place text box his nywhere on the gre age, just drag

- t.]
- (24) Parks, D. J.; Parsons, W. H.; Colburn, R. W.; Meegalla, S. K.; Ballentine, S. K.; Illig, ent C. R.; Qin, N.; Liu, Y.; Hutchinson, T. L.; Lubin, M. L.; Stone, D. J.; Baker, J. F.; ion Schneider, C. R.; Ma, J.; Damiano, B. P.; Flores, C. M.; Player, M. R. Design and wit Optimization of Benzimidazole-Containing Transient Receptor Potential Melastatin h a 8 (TRPM8) Antagonists. J. Med. Chem. 2011, 54 (1), 233-247.
- (25) Venable, J. D.; Cai, H.; Chai, W.; Dvorak, C. A.; Grice, C. A.; Jablonowski, J. A.; at Shah, C. R.; Kwok, A. K.; Ly, K. S.; Pio, B.; Wei, J.; Desai, P. J.; Jiang, W.; Nguyen, qu S.; Ling, P.; Wilson, S. J.; Dunford, P. J.; Thurmond, R. L.; Lovenberg, T. W.; ote Karlsson, L.; Carruthers, N. I.; Edwards, J. P. Preparation and Biological Evaluation fro of Indole, Benzimidazole, and Thienopyrrole Piperazine Carboxamides: Potent m Human Histamine H<sub>4</sub> Antagonists. J. Med. Chem. 2005, 48 (26), 8289–8298.
- the
- (26) Sørensen, U. S.; Strøbæk, D.; Christophersen, P.; Hougaard, C.; Jensen, M. L.; do Nielsen, E. Ø.; Peters, D.; Teuber, L. Synthesis and Structure–Activity Relationship cu Studies of 2-(N-Substituted)-Aminobenzimidazoles as Potent Negative Gating me Modulators of Small Conductance Ca<sup>2+</sup> -Activated K <sup>+</sup> Channels. J. Med. Chem. nt 2008, 51 (23), 7625-7634.
- or

(27) Balboni, G.; Trapella, C.; Sasaki, Y.; Ambo, A.; Marczak, E. D.; Lazarus, L. H.; use Salvadori, S. Influence of the Side Chain Next to C-Terminal Benzimidazole in thi Opioid Pseudopeptides Containing the Dmt-Tic Pharmacophore. J. Med. Chem. 2009, S 52 (17), 5556–5559.

spa

(28) Kishore Babu, P. N.; Ramadevi, B.; Poornachandra, Y.; Ganesh Kumar, C. Synthesis, ce Antimicrobial, Anticancer Evaluation Novel 2-(3and of to

- em
- ph
- asi
- ze

der reat quote from 's he document or att se this space to ent mphasize a key ion oint. To place his text box h a nywhere on the gre age, just drag t.]

- Methylindolyl)Benzimidazole Derivatives. *Med. Chem. Res.* **2014**, *23* (9), 3970– ent 3978. ion (29) Ramanjulu, J. M.; Pesiridis, G. S.; Yang, J.; Concha, N.; Singhaus, R.; Zhang, S.-Y.; wit Tran, J.-L.; Moore, P.; Lehmann, S.; Eberl, H. C.; Muelbaier, M.; Schneck, J. L.; h a Clemens, J.; Adam, M.; Mehlmann, J.; Romano, J.; Morales, A.; Kang, J.; Leister, L.;
- Graybill, T. L.; Charnley, A. K.; Ye, G.; Nevins, N.; Behnia, K.; Wolf, A. I.;
  at Kasparcova, V.; Nurse, K.; Wang, L.; Puhl, A. C.; Li, Y.; Klein, M.; Hopson, C. B.;
  qu Guss, J.; Bantscheff, M.; Bergamini, G.; Reilly, M. A.; Lian, Y.; Duffy, K. J.; Adams, ote J.; Foley, K. P.; Gough, P. J.; Marquis, R. W.; Smothers, J.; Hoos, A.; Bertin, J.
  fro Design of Amidobenzimidazole STING Receptor Agonists with Systemic Activity.
  m Nature 2018, 564 (7736), 439–443.
- the
- (30) Dokla, E. M. E.; Abutaleb, N. S.; Milik, S. N.; Li, D.; El-Baz, K.; Shalaby, M.-A. W.;
  do
  Al-Karaki, R.; Nasr, M.; Klein, C. D.; Abouzid, K. A. M.; Seleem, M. N.
  cu
  Development of Benzimidazole-Based Derivatives as Antimicrobial Agents and
  me
  Their Synergistic Effect with Colistin against Gram-Negative Bacteria. *Eur. J. Med.*nt *Chem.* 2020, 186, 111850.
- or
- (31) Jeyakkumar, P.; Liu, H. B.; Gopala, L.; Cheng, Y.; Peng, X. M.; Geng, R. X.; Zhou, use
  C. H. Novel Benzimidazolyl Tetrahydroprotoberberines: Design, Synthesis, thi
  Antimicrobial Evaluation and Multi-Targeting Exploration. *Bioorg. Med. Chem. Lett.*s
- spa

**2017**, *27* (8), 1737–1743.

(32) El-Gohary, N. S.; Shaaban, M. I. Synthesis, Antimicrobial, Antiquorum-Sensing and
 Antitumor Activities of New Benzimidazole Analogs. *Eur. J. Med. Chem.* 2017, *137*,
 to
 439–449.

- em
- ph
- asi
- ze

reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

t.]

der 's att ent ion

(33) Agh-Atabay, N. Synthesis and Investigation of Antimicrobial Activity of Some Bisbenzimidazole-Derived Chelating Agents. Eur. J. Med. Chem. 2003, 38 (10), 875-881.

(34) Göker, H.; Özden, S.; Yıldız, S.; Boykin, D. W. Synthesis and Potent Antibacterial h a Activity against MRSA of Some Novel 1,2-Disubstituted-1H-Benzimidazole-Ngre Alkylated-5-Carboxamidines. Eur. J. Med. Chem. 2005, 40 (10), 1062–1069.

- (35) Ramprasad, J.; Nayak, N.; Dalimba, U.; Yogeeswari, P.; Sriram, D.; Peethambar, S. qu K.; Achur, R.; Kumar, H. S. S. Synthesis and Biological Evaluation of New ote Imidazo[2,1-b][1,3,4]Thiadiazole-Benzimidazole Derivatives. Eur. J. Med. Chem. fro **2015**, *95*, 49–63.
- m

wit

at

(36) Zhang, H. Z.; He, S. C.; Peng, Y. J.; Zhang, H. J.; Gopala, L.; Tangadanchu, V. K. the R.; Gan, L.-L.; Zhou, C.-H. Design, Synthesis and Antimicrobial Evaluation of Novel do Benzimidazole-Incorporated Sulfonamide Analogues. Eur. J. Med. Chem. 2017, 136, cu 165–183.

me

(37) Abraham, R.; Prakash, P.; Mahendran, K.; Ramanathan, M. A Novel Series of Nnt Acyl Substituted Indole-Linked Benzimidazoles and Naphthoimidazoles as Potential or Anti Inflammatory, Anti Biofilm and Anti Microbial Agents. Microb. Pathog. 2018, use 114, 409–413.

- thi
- (38) Song, D.; Ma, S. Recent Development of Benzimidazole-Containing Antibacterial S Agents. ChemMedChem 2016, 11 (7), 646–659.

(39) Janupally, R.; Jeankumar, V. U.; Bobesh, K. A.; Soni, V.; Devi, P. B.; Pulla, V. K.;

Survadevara, P.; Chennubhotla, K. S.; Kulkarni, P.; Yogeeswari, P.; Sriram, D.

Structure-Guided Design and Development of Novel Benzimidazole Class of

- spa
- ce
- to
- em
- ph

119

ze

der reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag at

t.]

's att ent ion

Compounds Targeting DNA GyraseB Enzyme of Staphylococcus Aureus. *Bioorg*. Med. Chem. 2014, 22 (21), 5970-5987.

- (40) Picconi, P.; Hind, C.; Jamshidi, S.; Nahar, K.; Clifford, M.; Wand, M. E.; Sutton, J. wit M.; Rahman, K. M. Triaryl Benzimidazoles as a New Class of Antibacterial Agents h a against Resistant Pathogenic Microorganisms. J. Med. Chem. 2017, 60 (14), 6045gre 6059.
- (41) Monforte, A. M.; Ferro, S.; De Luca, L.; Lo Surdo, G.; Morreale, F.; Pannecouque, qu C.; Balzarini, J.; Chimirri, A. Design and Synthesis of N1-Aryl-Benzimidazoles 2ote Substituted as Novel HIV-1 Non-Nucleoside Reverse Transcriptase Inhibitors. fro Bioorg. Med. Chem. 2014, 22 (4), 1459–1467.
- m (42) Miao, T. T.; Tao, X. B.; Li, D. D.; Chen, H.; Jin, X. Y.; Geng, Y.; Wang, S. F.; Gu, the W. Synthesis and Biological Evaluation of 2-Aryl-Benzimidazole Derivatives of do Dehydroabietic Acid as Novel Tubulin Polymerization Inhibitors. RSC Adv. 2018, 8 cu

(31), 17511-17526.

- me (43) Li, P.; Zhang, W.; Jiang, H.; Li, Y.; Dong, C.; Chen, H.; Zhang, K.; Du, Z. Design, nt Synthesis and Biological Evaluation of Benzimidazole-Rhodanine Conjugates as or Potent Topoisomerase II Inhibitors. Med. Chem. Commun. 2018, 9 (7), 1194–1205.
- use (44) Hegde, M.; Sharath Kumar, K. S.; Thomas, E.; Ananda, H.; Raghavan, S. C.; thi Rangappa, K. S. A Novel Benzimidazole Derivative Binds to the DNA Minor Groove S and Induces Apoptosis in Leukemic Cells. RSC Adv. 2015, 5 (113), 93194–93208.
- spa (45) Alpan, A. S.; Zencir, S.; Zupkó, I.; Coban, G.; Réthy, B.; Gunes, H. S.; Topcu, Z. ce Biological Activity of Bis-Benzimidazole Derivatives on DNA Topoisomerase I and to HeLa, MCF7 and A431 Cells. J. Enzyme Inhib. Med. Chem. 2009, 24 (3), 844–849.

120

em

ph

asi

- reat quote from he document or use this space to mphasize a key oint. To place his text box nywhere on the age, just drag
- t.]
- wit h a gre

at

qu

der

's

att

- (46) Alkahtani, H. M.; Abbas, A. Y.; Wang, S. Synthesis and Biological Evaluation of ent Benzo[d]Imidazole Derivatives as Potential Anti-Cancer Agents. *Bioorg. Med. Chem.* ion Lett. 2012, 22 (3), 1317–1321.
  - (47) Nayak, V. L.; Nagesh, N.; Ravikumar, A.; Bagul, C.; Vishnuvardhan, M. V. P. S.; Srinivasulu, V.; Kamal, A. 2-Aryl Benzimidazole Conjugate Induced Apoptosis in Human Breast Cancer MCF-7 Cells through Caspase Independent Pathway. Apoptosis 2017, 22 (1), 118–134.
- (48) Darwish, S. A.; Elbayaa, R. Y.; Ashour, H. M.; Khalil, M. A.; Badawey, E. A. ote Potential Anticancer Agents: Design, Synthesis of New Pyrido [1,2-a] Benzimidazoles fro and Related Derivatives Linked to Alkylating Fragments. *Med. chem.* **2018**, *08* (04).
- m (49) Rashedy, A.; Aboul-Enein, H. Benzimidazole Derivatives Potential as the Chemotherapeutic Agents. Curr. Drug Ther. 2013, 8 (1), 1-14.
- do
- (50) Refaat, H. M. Synthesis and Anticancer Activity of Some Novel 2-Substituted cu Benzimidazole Derivatives. Eur. J. Med. Chem. 2010, 45 (7), 2949–2956.
- me
- (51) Oksuzoglu, E.; Tekiner-Gulbas, B.; Alper, S.; Temiz-Arpaci, O.; Ertan, T.; Yildiz, I.; nt Diril, N.; Sener-Aki, E.; Yalcin, I. Some Benzoxazoles and Benzimidazoles as DNA or

Topoisomerase I and II Inhibitors. J. Enzyme Inhib. Med. Chem. 2008, 23 (1), 37–42.

- use
- (52) Chahrour, O.; Abdalla, A.; Lam, F.; Midgley, C.; Wang, S. Synthesis and Biological thi Evaluation of Benzyl Styrylsulfonyl Derivatives as Potent Anticancer Mitotic S Inhibitors. Bioorg. Med. Chem. Lett. 2011, 21 (10), 3066–3069.
- spa
- (53) Wang, Z.; Deng, X.; Xiong, S.; Xiong, R.; Liu, J.; Zou, L.; Lei, X.; Cao, X.; Xie, Z.; ce Chen, Y.; Liu, Y.; Zheng, X.; Tang, G. Design, Synthesis and Biological Evaluation to
- em

121

ph

asi

	der
reat quote from	uer
	's
he document or	
	att
ise this space to	~
mphasize a kev	ent
inphasize a key	ion
oint. To place	
	wit
his text box	1
nywhere on the	n a
ing where on the	gre
age, just drag	U
-	at
t.]	
	qu

of Chrysin Benzimidazole Derivatives as Potential Anticancer Agents. *Nat. Prod. Res.* **2018**, *32* (24), 2900–2909.

- (54) Yadav, S.; Narasimhan, B.; kaur, H. Perspectives of Benzimidazole Derivatives as wit
   Anticancer Agents in the New Era. *Anti-Cancer Agents Med. Chem.* 2016, 16 (11), h a
   1403 1425.
- (55) Gaba, M.; Gaba, P.; Uppal, D.; Dhingra, N.; Bahia, M. S.; Silakari, O.; Mohan, C. at
   Benzimidazole Derivatives: Search for GI-Friendly Anti-Inflammatory Analgesic
   qu
   Agents. Acta Pharm. Sin. B 2015, 5 (4), 337–342.
- ote

 (56) Achar, K. C. S.; Hosamani, K. M.; Seetharamareddy, H. R. In-Vivo Analgesic and fro Anti-Inflammatory Activities of Newly Synthesized Benzimidazole Derivatives. *Eur*.
 *Med. Chem.* 2010, 45 (5), 2048–2054.

- the
- (57) Ganie, A. M.; Dar, A. M.; Dar\*, F. A. K. and B. A. Benzimidazole Derivatives as do
  Potential Antimicrobial and Antiulcer Agents: *A Mini-Rev. Med. Chem.* 2019, 19 cu
  (16), 1292 1297.
- me

(58) Alpan, A. S.; Parlar, S.; Carlino, L.; Tarikogullari, A. H.; Alptüzün, V.; Güneş, H. S. nt
Synthesis, Biological Activity and Molecular Modeling Studies on 1H-or
Benzimidazole Derivatives as Acetylcholinesterase Inhibitors. *Bioorg. Med. Chem.*use
2013, 21 (17), 4928–4937.

thi

 (59) Desai, N. C.; Shihory, N. R.; Kotadiya, G. M.; Desai, P. Synthesis, Antibacterial and Antitubercular Activities of Benzimidazole Bearing Substituted 2-Pyridone Motifs.

 spa
 *Eur. J. Med. Chem.* 2014, 82, 480–489.

- ce
- to
- em
- ph asi
- ze

	der
reat quote from	's
he document or	
	att
se this space to	
	ent
emphasize a key	•
point To place	10n
onit. To place	wit
his text box	wite
	h a
nywhere on the	
	gre
bage, just drag	
с <b>Т</b>	at
t.j	

(60) Starčević, K.; Kralj, M.; Ester, K.; Sabol, I.; Grce, M.; Pavelić, K.; Karminskient Zamola, G. Synthesis, Antiviral and Antitumor Activity of 2-Substituted-5-Amidinoion Benzimidazoles. *Bioorg. Med. Chem.* 2007, *15* (13), 4419–4426.
wit (61) Valdez, J.; Cedillo, R.; Hernández-Campos, A.; Yépez, L.; Hernández-Luis, F.; h a Navarrete-Vázquez, G.; Tapia, A.; Cortés, R.; Hernández, M.; Castillo, R. Synthesis gre

Lett. 2002, 12 (16), 2221–2224.

J. Org. Chem. 2011, 7, 442–495.

qu

(62) Mayence, A.; Pietka, A.; Collins, M. S.; Cushion, M. T.; Tekwani, B. L.; Huang, T. ote
L.; Vanden Eynde, J. J. Novel Bisbenzimidazoles with Antileishmanial Effectiveness.
fro *Bioorg. Med. Chem. Lett.* 2008, 18 (8), 2658–2661.

and Antiparasitic Activity of 1H-Benzimidazole Derivatives. Bioorg. Med. Chem.

- m
  (63) Farahat, A. A.; Ismail, M. A.; Kumar, A.; Wenzler, T.; Brun, R.; Paul, A.; Wilson, the
  W. D.; Boykin, D. W. Indole and Benzimidazole Bichalcophenes: Synthesis, DNA
- do Binding and Antiparasitic Activity. *Eur. J. Med. Chem.* **2018**, *143*, 1590–1596.
- (64) Townsend, L. B.; Wise, D. S. The Synthesis and Chemistry of Certain Anthelmintic me
   Benzimidazoles. *Parasitol. Today* 1990, 6 (4), 107–112.
- nt (65) Salahuddin; Shaharyar, M.; Mazumder, A. Benzimidazoles: A Biologically Active or Compounds. *Arabian J. Chem.* **2017**, *10*, S157–S173.
- use

cu

(66) Kumar, J. R.; Jawahar L., J.; Pathak, D. P. Synthesis of Benzimidazole Derivatives:
thi
As Anti-Hypertensive Agents. *E-J. Chem.* 2006, 3 (4), 278–285.

(67) Baumann, M.; Baxendale, I. R.; Ley, S. V.; Nikbin, N. An Overview of the Key

Routes to the Best Selling 5-Membered Ring Heterocyclic Pharmaceuticals. Beilstein

- S
- spa
- ce
- to
- em
- nh
- ph

asi

ze

	der
great quote from	
	's
he document or	
	att
se this space to	
	ent
mphasize a key	
	ion
oint. To place	
	wit
his text box	
1 (1	ha
nywhere on the	
and inst dues	gre
age, just drag	
+ ]	at
ւ.յ	

- (68) LaBarbera, D. V.; Skibo, E. B. Synthesis of Imidazo[1,5,4-de]Quinoxalin-9-Ones, Benzimidazole Analogues of Pyrroloiminoquinone Marine Natural Products. *Bioorg*. *Med. Chem.* 2005, *13* (2), 387–395.
- (69) Martins, P.; Jesus, J.; Santos, S.; Raposo, L. R.; Roma-Rodrigues, C.; Baptista, P. V.;
  h a Fernandes, A. R. Heterocyclic Anticancer Compounds: Recent Advances and the gre Paradigm Shift towards the Use of Nanomedicine's Tool Box. *Molecules* 2015, 20 at (9), 16852–16891.
- -

(70) Rajam, S.; Ranjith, R. The Chemistry and Biological Significance of Imidazole, ote
Benzimidazole, Benzoxazole, Tetrazole and Quinazolinone Nucleus. J. Chem. fro *Pharm. res.* 2016, 8 (5), 505 - 526.

- (71) Gandhi, P.; Schmitt, E. K.; Chen, C.-W.; Samantray, S.; Venishetty, V. K.; Hughes, the
   D. Triclabendazole in the Treatment of Human Fascioliasis: A Review. *Trans. R. Soc.* do
  - *Trop. Med. Hyg.* **2019**, *113* (12), 797 804.

2005, 46 (31), 5195-5197.

cu

qu

m

- (72) Alaqeel, S. I. Synthetic Approaches to Benzimidazoles from O-Phenylenediamine: A me
   Literature Review. J. Saudi Chem. Soc. 2017, 21 (2), 229–237.
- nt
- (73) Mazurov, A. Traceless Synthesis of Benzimidazoles on Solid Supporty. *Bioorg. Med.* or *Chem. Lett.* 2000, 4.
- use
- (74) Kilburn, J. P.; Lau, J.; Jones, R. C. F. Solid-Phase Synthesis of Substituted 2thi Aminomethylbenzimidazoles. *Tetrahedron Lett*. **2000**, *41* (28), 5419–5421.

(75) Vaidyanathan, S.; Surber, B. W. Microwave Mediated Hydrogen Deuterium

Exchange: A Rapid Synthesis of 2H-Substituted Benzimidazole. Tetrahedron Lett.

124

S

spa

- ce
- to
- em
- ph
- Ρu

asi

- der reat quote from 's he document or att use this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the gre age, just drag t.]
  - (78) Perkins, J. J.; Zartman, A. E.; Meissner, R. S. Synthesis of 2-at (Alkylamino)Benzimidazoles. *Tetrahedron Lett.* 1999, 40 (6), 1103–1106.
    (79) Singhal, S.; Khanna, P.; Panda, S. S.; Khanna, L. Recent Trends in the Synthesis of ote Benzimidazoles From o -Phenylenediamine via Nanoparticles and Green Strategies fro Using Transition Metal Catalysts. J. Heterocyclic. Chem. 2019, 56 (10), 2702–2729.

(76) Wang, Y.; Sarris, K.; Sauer, D. R.; Djuric, S. W. A Simple and Efficient One Step

(77) Das, B.; Holla, H.; Srinivas, Y. Efficient (Bromodimethyl)Sulfonium Bromide

Mediated Synthesis of Benzimidazoles. *Tetrahedron Lett.* **2007**, 48 (1), 61–64.

*Lett.* **2006**, *47* (28), 4823–4826.

Synthesis of Benzoxazoles and Benzimidazoles from Carboxylic Acids. Tetrahedron

- m (80) Rithe, S. R.; Jagtap, R. S.; Ubarhande, S. S. One Pot synthesis of Substituted the Benzimidazole Derivatives and their Characterization. *Rasayan J. Chem.* **2015**, *8* (2), do 213 - 217.
- cu

 (81) Samanta, D.; Rana, A.; Bats, J. W.; Schmittel, M. A One-Pot Multistep Cyclization me Yielding Thiadiazoloimidazole Derivatives. *Beilstein J. Org. Chem.* 2014, *10*, 2989–
 nt
 2996.

or

(82) Herrera Cano, N.; Uranga, J. G.; Nardi, M.; Procopio, A.; Wunderlin, D. A.; Santiago, use
A. N. Selective and Eco-Friendly Procedures for the Synthesis of Benzimidazole thi
Derivatives. The Role of the Er(OTf) 3 Catalyst in the Reaction Selectivity. *Beilstein J. Org. Chem.* 2016, *12*, 2410–2419.

- spa
- (83) Venkateswarlu, Y.; Kumar, S.; Leelavathi, P. Facile and Efficient One-Pot Synthesis
   of Benzimidazoles Using Lanthanum Chloride. Org. Med. Chem. Lett. 2013, 3 (1), 7.
   to

125

- em
- ph

asi

- reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag
- t.]
- der 's att h a
  - (84) Lai, T. T.; Xie, D.; Zhou, C. H.; Cai, G. X. Copper-Catalyzed Inter/Intramolecular N ent -Alkenylation of Benzimidazoles via Tandem Processes Involving Selectively Mild ion Iodination of Sp<sup>3</sup> C–H Bond at α-Position of Ester. J. Org. Chem. 2016, 81 (19), wit 8806-8815.
  - (85) Zhang, R.; Qin, Y.; Zhang, L.; Luo, S. Oxidative Synthesis of Benzimidazoles, gre Quinoxalines, and Benzoxazoles from Primary Amines by Ortho -Quinone Catalysis. at Org. Lett. 2017, 19 (20), 5629–5632.
  - qu (86) Adharvana Chari, M.; Shobha, D.; Sasaki, T. Room Temperature Synthesis of ote Benzimidazole Derivatives Using Reusable Cobalt Hydroxide (II) and Cobalt Oxide fro (II) as Efficient Solid Catalysts. *Tetrahedron Lett.* **2011**, *52* (43), *5575–5580*.
  - m (87) Cimarelli, C.; Di Nicola, M.; Diomedi, S.; Giovannini, R.; Hamprecht, D.; Properzi, the R.; Sorana, F.; Marcantoni, E. An Efficient One-Pot Two Catalyst System in the do Construction of 2-Substituted Benzimidazoles: Synthesis of Benzimidazo[1,2cu c]Quinazolines. Org. Biomol. Chem. 2015, 13 (48), 11687–11695.
  - me
  - (88) Carvalho, L. C. R.; Fernandes, E.; Marques, M. M. B. Developments Towards nt Regioselective Synthesis of 1,2-Disubstituted Benzimidazoles. Chem. Eur. J. 2011, or 17 (45), 12544–12555.
  - use
  - (89) Chen, L. H.; Chang, C. M.; Salunke, D. B.; Sun, C. M. Divergent Synthesis of thi Unsymmetrical Annulated Biheterocyclic Compound Libraries: Benzimidazole S Linked Indolo-Benzodiazepines/Quinoxaline. ACS Comb. Sci. 2011, 13 (4), 391–398. spa (90) Bahrami, K.; Khodaei, M. M.; Naali, F. Mild and Highly Efficient Method for the

Synthesis of 2-Arylbenzimidazoles and 2-Arylbenzothiazoles. J. Org. Chem. 2008,

- ce
- to

em

73 (17), 6835-6837.

ph

126

ze

	der	
great quote from	's	
he document or	3	
se this space to	att (91)	Bahra
mphasize a key	ent	Benzi
oint. To place	ion	SDS N
his text box	wit (92)	Bastug
nywhere on the	h a	Buildi
page, just drag	gre	2012.
t]	at (93)	Phillir
	qu	Reacti
	ote	Keucii
	fro	conrr4
	(94) m	Lin, S
	111	Benzi
	the	4319.
	do (95)	Aggar
	cu	via El
	me	7653
	nt (96)	Mame
	(90) or	wiaiiie
		Rearra

- (91) Bahrami, K.; Khodaei, M. M.; Nejati, A. Synthesis of 1,2-Disubstituted t
   Benzimidazoles, 2-Substituted Benzimidazoles and 2-Substituted Benzothiazoles in
   SDS Micelles. *Green Chem.* 2010, *12* (7), 1237.
- (92) Bastug, G.; Eviolitte, C.; Markó, I. E. Functionalized Orthoesters as Powerful Building Blocks for the Efficient Preparation of Heteroaromatic Bicycles. Org. Lett.
   2012, 14 (13), 3502–3505.
- (93) Phillips-Ladenburg Benzimidazole Synthesis. In Comprehensive Organic Name
   Reactions and Reagents; John Wiley & Sons, Inc.: Hoboken, NJ, USA, 2010; p
   ote
- (94) Lin, S.; Yang, L. A Simple and Efficient Procedure for the Synthesis of
   Benzimidazoles Using Air as the Oxidant. *Tetrahedron Lett.* 2005, 46 (25), 4315–
   the
   4319.
- (95) Aggarwal, T.; Kumar, S.; Verma, A. K. Iodine-Mediated Synthesis of Heterocycles cu via Electrophilic Cyclization of Alkynes. *Org. Biomol. Chem.* 2016, *14* (32), 7639–me
   7653
- (96) Mamedov, V. A. Recent Advances in the Synthesis of Benzimidazol(on)Es via or Rearrangements of Quinoxalin(on)Es. RSC Adv. 2016, 6 (48), 42132–42172.
- use (97) Wright, J. B. The Chemistry of the Benzimidazoles. *Chem. Rev.* **1951**, *48* (3), 397–thi 541.
- S
- (98) Sontakke, V. A.; Ghosh, S.; Lawande, P. P.; Chopade, B. A.; Shinde, V. S. A Simple,
   spa
   Efficient Synthesis of 2-Aryl Benzimidazoles Using Silica Supported Periodic Acid
   ce
   Catalyst and Evaluation of Anticancer Activity. *ISRN Org. Chem.* 2013, 2013, 1–7.

- to
- em
- ph
  - ••
- asi ze

- reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag
- t.]
- der 's att (99) Pfitzner, K. E.; Moffatt, J. G. A New and Selective Oxidation of Alcohols. J. Am. ent ion (100) Peng, J.; Ye, M.; Zong, C.; Hu, F.; Feng, L.; Wang, X.; Wang, Y.; Chen, C. Copperwit h a
  - qu ote fro
    - (102) Chen, C.; Chen, C.; Li, B.; Tao, J.; Peng, J. Aqueous Synthesis of 1-H-2-Substituted m Benzimidazoles via Transition-Metal-Free Intramolecular Amination of Aryl Iodides. the Molecules 2012, 17 (11), 12506–12520.

Ammonia

Iodophenylcarbamates. J. Org. Chem. 2009, 74 (20), 7974–7977.

Catalyzed Intramolecular C-N Bond Formation: A Straightforward Synthesis of

Benzimidazoles and 1,3-Dihydrobenzimidazol-2-Ones via CuI/ L -Proline Catalyzed

with

2-Iodoacetanilides

and

2-

128

Benzimidazole Derivatives in Water. J. Org. Chem. 2011, 76 (2), 716-719.

(101) Diao, X.; Wang, Y.; Jiang, Y.; Ma, D. Assembly of Substituted 1 H -

Chem. Soc. 1963, 85 (19), 3027-3028.

do

gre

at

Coupling

of

Aqueous

- (103) Moorthy, J. N.; Neogi, I. IBX-Mediated One-Pot Synthesis of Benzimidazoles from cu Primary Alcohols and Arylmethyl Bromides. Tetrahedron Lett. 2011, 52 (30), 3868me 3871.
- nt
- (104) Wray, B. C.; Stambuli, J. P. Synthesis of N Arylindazoles and Benzimidazoles or from a Common Intermediate. Organic Lett. 2010, 12 (20), 4576–4579.
- (105) Wang, Q.; Schreiber, S. L. Copper-Mediated Amidation of Heterocyclic and thi Aromatic C-H Bonds. Org. Lett. 2009, 11 (22), 5178-5180.
- S

use

- (106) Cui, W.; Kargbo, R. B.; Sajjadi-Hashemi, Z.; Ahmed, F.; Gauuan, J. F. Efficient spa One-Pot Synthesis of 2-Substituted Benzimidazoles from Triacyloxyborane ce Intermediates. Synlett. 2012, 2012 (02), 247–250.
- to
- em
- ph
- ze

reat quote from	
ieat quote nom	's
he document or	att
se this space to	(107) Trivedi, R.; De, S. K.; Gibbs, R. A. A Convenient One-Pot Synthesis of 2-
mphasize a key	Substituted Benzimidazoles. J. Mol. Catal. A Chem. 2006, 245 (1–2), 8 - 11.
oint. To place	(108) Zhang, Z. H.; Yin, L.; Wang, Y. M. An Expeditious Synthesis of Benzimidazole
his text box	Wit Derivatives Catalyzed by Lewis Acids. Catal. Commun. 2007, 8 (7), 1126–1131.
nywhere on the	h a (109) Cescon, L. A.; Day, A. R. Preparation of Some Benzimidazolylamino Acids.
age, just drag	gre Reactions of Amino Acids with o-Phenylenediamines. J. Org. Chem. 1962, 27 (2),
t.]	at 581–586.
	qu (110) Singla, R.; Gupta, K. B.; Upadhyay, S.; Dhiman, M.; Jaitak, V. Design, Synthesis
	ote and Biological Evaluation of Novel Indole-Benzimidazole Hybrids Targeting
	fro Estrogen Receptor Alpha (ER-α). Eur. J. Med. Chem. 2018, 146, 206–219.
	m (111) Mondal, S.; Thompson, P. R. Protein Arginine Deiminases (PADs): Biochemistry
	the and Chemical Biology of Protein Citrullination. Acc. Chem. Res. 2019, 52 (3), 818–
	do 832.
	cu (112) Lewis, H. D.; Liddle, J.; Coote, J. E.; Atkinson, S. J.; Barker, M. D.; Bax, B. D.;
	me Bicker, K. L.; Bingham, R. P.; Campbell, M.; Chen, Y. H.; Chung, C.; Craggs, P. D.;
	nt Davis, R. P.; Eberhard, D.; Joberty, G.; Lind, K. E.; Locke, K.; Maller, C.; Martinod.
	or K.: Patten, C.: Polyakova, O.: Rise, C. E.: Rüdiger, M.: Sheppard, R. J.: Slade, D. J.:
	use Thomas P: Thorpe I: Yao G: Drewes G: Wagner D D: Thompson P R:
	thi Priniba R K : Wilson D M Inhibition of $PAD4$ Activity Is Sufficient to Disrupt
	s Net Cham District Net Cham Dist 2015, 11, 190
	spa
	(113) Fuhrmann, J.; Thompson, P. R. Protein Arginine Methylation and Citrullination in ce
	Epigenetic Regulation. ACS Chem. Biol. 2016, 11 (3), 654–668. to
	em
	ph 129
	ası

der
	der
great quote from	's
he document or	att
se this space to	(114) Kenner, G. W.; McCombie, S. W.; Smith, K. M. Pyrroles and Related Compounds.
mphasize a key	Part XXIV. Separation and Oxidative Degradation of Chlorophyll Derivatives. J.
oint. To place	Chem. Soc., Perkin Trans. 1 <b>1973</b> , 2517.
his text box	(115) Chakraborty, A.; Debnath, S.; Ghosh, T.; Maiti, D. K.; Majumdar, S. An Efficient
nywhere on the	h a Strategy for N-Alkylation of Benzimidazoles/Imidazoles in SDS-Aqueous Basic
oage, just drag	gre Medium and N-Alkylation Induced Ring Opening of Benzimidazoles. Tetrahedron
t.]	at <b>2018</b> , <i>74</i> (40), 5932–5941.
	qu (116) Karchava, A. V.; Melkonyan, F. S.; Yurovskaya, M. A. New Strategies for the
	ote Synthesis of N-Alkylated Indoles (Review). Chem. Heterocycl. Comp. 2012, 48 (3),
	fro 391–407.
	m (117) Earle, M. J.; McCormac, P. B.; Seddon, K. R. Regioselective Alkylation in Ionic
	the Liquids. Chem. Commun. 1998, No. 20, 2245–2246.
	do (118) Le, ZG.; Chen, ZC.; Hu, Y.; Zheng, QG. Organic Reactions in Ionic Liquids:
	cu A Simple and Highly Regioselective N-Substitution of Pyrrole. Synth. 2004, 2004
	me (12), 1951–1954.
	nt (119) Siegel R. L.: Miller K. D.: Jemal A. Cancer Statistics 2019 CA A Cancer J Clin
	or $2019\ 60\ (1)\ 7\ -\ 34$
	Use (120) Street W. Cancer Facts & Figures 2020 1930 76
	(120) Street, W. Cancer Facts & Figures 2020. $1550$ , 70. thi (121) Singel P. L.: Miller K. D.: Jamel A. Concer Statistics 2017. <i>CA</i> : A Concer L
	(121) Slegel, R. L.; Miller, K. D.; Jemai, A. Cancer Statistics, 2017. CA: A Cancer J.
	spa (122) $T_{1} = 50.$
	(122) Types of Cancer Treatment https://www.cancer.gov/about-cancer/treatment/types. ce
	to
	em
	ph 130
	asi

reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the age, just drag

t.]

's att (123) Liang, X. J.; Chen, C.; Zhao, Y.; Wang, P. C. Circumventing Tumor Resistance to ent Chemotherapy by Nanotechnology. In Multi-Drug Resist. Cancer. 2010, 596, 467 ion

488.

(124) Holohan, C.; Van Schaeybroeck, S.; Longley, D. B.; Johnston, P. G. Cancer Drug h a Resistance: An Evolving Paradigm. Nat. Rev. Cancer 2013, 13 (10), 714 - 726.

gre (125) Markman, J. L.; Rekechenetskiy, A.; Holler, E.; Ljubimova, J. Y. Nanomedicine at Therapeutic Approaches to Overcome Cancer Drug Resistance. Adv. Drug Deliv. Rev. qu 2013, 65 (13–14), 1866 - 1879.

ote

wit

der

(126) Krzyszczyk, P.; Acevedo, A.; Davidoff, E. J.; Timmins, L. M.; Marrero-Berrios, I.; fro Patel, M.; White, C.; Lowe, C.; Sherba, J. J.; Hartmanshenn, C.; O'Neill, K. M.;

m Balter, M. L.; Fritz, Z. R.; Androulakis, I. P.; Schloss, R. S.; Yarmush, M. L. The the Growing Role of Precision and Personalized Medicine for Cancer Treatment. do *Technology* **2018**, *06* (03n04), 79–100.

cu

(127) Mokhtari, R. B.; Homayouni, T. S.; Baluch, N.; Morgatskaya, E.; Kumar, S.; Das, me B.; Yeger, H. Combination Therapy in Combating Cancer. *Oncotarget* 2017, 8 (23).

(128) Reese, M. J.; Knapp, D. W.; Anderson, K. M.; Mund, J. A.; Case, J.; Jones, D. R.;

nt

or

Packer, R. A. In Vitro Effect of Chlorambucil on Human Glioma Cell Lines (SF767 use and U87-MG), and Human Microvascular Endothelial Cell (HMVEC) and thi Endothelial Progenitor Cells (ECFCs), in the Context of Plasma Chlorambucil S Concentrations in Tumor-Bearing Dogs. PLoS ONE 2018, 13 (9), e0203517.

(129) Di Antonio, M.; McLuckie, K. I. E.; Balasubramanian, S. Reprogramming the

Am. Chem. Soc. 2014, 136 (16), 5860-5863.

Mechanism of Action of Chlorambucil by Coupling to a G-Quadruplex Ligand. J.

- spa
- ce
- to
- em
- ph
- asi ze

131

	der
reat quote from	's
he document or	att
use this space to	(130) Panasci, L.; Paiement, JP.; Christodoulopoulos, G.; Belenkov, A.; Malapetsa, A.;
mphasize a key	Aloyz, R. Chlorambucil Drug Resistance in Chronic Lymphocytic Leukemia: The
oint. To place	Emerging Role of DNA Repair. <i>Clin. Cancer Res.</i> <b>2001</b> , 7(3),454-461.
his text box	(131) Longley, D. B.; Harkin, D. P.; Johnston, P. G. 5-Fluorouracil: Mechanisms of
nywhere on the	h a Action and Clinical Strategies. <i>Nat. Rev. Cancer</i> <b>2003</b> , <i>3</i> (5), 330–338.
oage, just drag	gre (132) Mader, R. M.; Müller, M.; Steger, G. G. Resistance to 5-Fluorouracil. <i>General</i>
t.]	at Pharmacology: The Vascular System <b>1998</b> , 31 (5), 661–666.
	qu (133) Pommier, Y.; Leo, E.; Zhang, H.; Marchand, C. DNA Topoisomerases and Their
	ote Poisoning by Anticancer and Antibacterial Drugs. <i>Chem. Bio.</i> <b>2010</b> , <i>17</i> (5), 421–433.
	fro (134) Helmbach, H.; Kern, M. A.; Rossmann, E.; Renz, K.; Kissel, C.; Gschwendt, B.;
	m Schadendorf, D. Drug Resistance Towards Etoposide and Cisplatin in Human
	the Melanoma Cells Is Associated with Drug-Dependent Apoptosis Deficiency. J.
	do Investi. Derma. 2002, 118 (6), 923 – 932.
	cu (135) Kanwal, A.; Saddique, F. A.; Aslam, S.; Ahmad, M.; Zahoor, A. F.; Mohsin, NA.
	me Benzimidazole Ring System as a Privileged Template for Anticancer Agents. <i>Pharm</i> .
	nt Chem. J. <b>2018</b> , 51 (12), 1068–1077.
	or (136) Purushottamachar, P.; Ramalingam, S.; C.O. Njar, V. Development of
	use Benzimidazole Compounds for Cancer Therapy. In <i>Chemistry and Applications of</i>
	thi <i>Benzimidazole and its Derivatives</i> ; Marinescu, M., Ed.; IntechOpen, <b>2019</b> .
	s (137) Hoy S M Bendamustine: A Review of Its Use in the Management of Chronic
	spa
	ce
	Lymphoma and Multiple Myeloma. <i>Drugs</i> <b>2012</b> , <i>72</i> (14), 1929–1950. to
	em
	ph 132
	-

asi

	der
reat quote from	's
he document or	
	att
se this space to	
	ent
emphasize a key	:
oint To place	1011
	wit
his text box	
	h a
nywhere on the	
	gre
bage, just drag	
. 1	at

- (138) Wagner, L. Profile of Veliparib and Its Potential in the Treatment of Solid Tumors.
   *Onco Targets Ther.* 2015, 1931.
- (139) Decaudin, D.; El Botty, R.; Diallo, B.; Massonnet, G.; Fleury, J.; Naguez, A.; wit
  Raymondie, C.; Davies, E.; Smith, A.; Wilson, J.; Howes, C.; Smith, P. D.; Cassoux, h a
  N.; Piperno-Neumann, S.; Roman-Roman, S.; Némati, F. Selumetinib-Based Therapy gre
  in Uveal Melanoma Patient-Derived Xenografts. *Oncotarget* 2018, 9 (31), 21674–
  at
- qu
- (140) McClurg, U. L.; Azizyan, M.; Dransfield, D. T.; Namdev, N.; Chit, N. C. T. H.;
   ote Nakjang, S.; Robson, C. N. The Novel Anti-Androgen Candidate Galeterone Targets fro Deubiquitinating Enzymes, USP12 and USP46, to Control Prostate Cancer Growth m and Survival. *Oncotarget* 2018, 9 (38), 24992–25007.
- the
- (141) Eilers, U.; Klumperman, J.; Hauri, H. P. Nocodazole, a Microtubule-Active Drug,
   do Interferes with Apical Protein Delivery in Cultured Intestinal Epithelial Cells (Cacocu
   2). J. Cell Bio. 1989, 108 (1), 13–22.
- me
- (142) Hranjec, M.; Kralj, M.; Piantanida, I.; Sedić, M.; Šuman, L.; Pavelić, K.; nt Karminski-Zamola, G. Novel Cyano- and Amidino-Substituted Derivatives of Styrylor
  2-Benzimidazoles and Benzimidazo[1,2-a]Quinolines. Synthesis, Photochemical use Synthesis, DNA Binding, and Antitumor Evaluation, Part 3. J. Med. Chem. 2007, 50 thi
- S
- (143) Bansal, Y.; Silakari, O. The Therapeutic Journey of Benzimidazoles: A Review.
   spa
   *Bioorg. Med. Chem.* 2012, 20 (21), 6208–6236.

133

- ce
- to
- em
- ph

asi

der reat quote from he document or att use this space to mphasize a key oint. To place his text box nywhere on the age, just drag at

's

(144) Carreira, A. C.; Alves, G. G.; Zambuzzi, W. F.; Sogayar, M. C.; Granjeiro, J. M. ent Bone Morphogenetic Proteins: Structure, Biological Function and Therapeutic ion Applications. Arch. Biochem. Biophys. 2014, 561, 64 - 73. wit

(145) Wang, R. N.; Green, J.; Wang, Z.; Deng, Y.; Qiao, M.; Peabody, M.; Zhang, Q.; h a Ye, J.; Yan, Z.; Denduluri, S.; Idowu, O.; Li, M.; Shen, C.; Hu, A.; Haydon, R. C.; gre Kang, R.; Mok, J.; Lee, M. J.; Luu, H. L.; Shi, L. L. Bone Morphogenetic Protein (BMP) Signaling in Development and Human Diseases. Genes & Diseases 2014, 1

t.]

qu (1), 87-105.

ote

- (146) Prakash, C. A.; Parthiban, J.; Balakrishnan, R.; Anandh, B.; Lokesh, B. Bone fro Morphogenetic Proteins-An Update. Biomed. Pharmacol. J. 2015, 8, 329-333.
- m (147) Genthe, J. R.; Min, J.; Farmer, D. M.; Shelat, A. A.; Grenet, J. A.; Lin, W.; the Finkelstein, D.; Vrijens, K.; Chen, T.; Guy, R. K.; Clements, W. K.; Roussel, M. F.
- do Ventromorphins: A New Class of Small Molecule Activators of the Canonical BMP cu Signaling Pathway. ACS Chem. Biol. 2017, 12 (9), 2436–2447.
- me
- (148) Katagiri, T.; Watabe, T. Bone Morphogenetic Proteins. Cold Spring Harb. nt Perspect. Biol. 2016, 8 (6), a021899.
- or (149) Chen, D.; Zhao, M.; Mundy, G. R. Bone Morphogenetic Proteins. Growth Factors use **2004**, *22* (4), *233–241*.
- thi
  - (150) Grinspan, J. B. Bone Morphogenetic Proteins. Vitam. Horm. 2015, 99, 195 222.
- S
- (151) Bradford, S. T. J.; Ranghini, E. J.; Grimley, E.; Lee, P. H.; Dressler, G. R. Highspa Throughput Screens for Agonists of Bone Morphogenetic Protein (BMP) Signaling ce Identify Potent Benzoxazole Compounds. J. Biol. Chem. 2019, 294 (9), 3125–3136.

134

- to
- em
- ph

- ze

	der
great quote from	
	's
he document or	
	att
se this space to	
	ent
mphasize a key	
	ion
oint. To place	
	wit
his text box	
	h a
nywhere on the	
	gre
bage, just drag	
	at

- t.]
- (154) Peng, J.; Li, Q.; Wigglesworth, K.; Rangarajan, A.; Kattamuri, C.; Peterson, R. T.;
   at Eppig, J. J.; Thompson, T. B.; Matzuk, M. M. Growth Differentiation Factor 9: Bone qu
   Morphogenetic Protein 15 Heterodimers Are Potent Regulators of Ovarian Functions.
   ote *Proc. Natl. Acad. Sci.* 2013, *110* (8), E776 E785.

(152) Xiao, Y. T.; Xiang, L. X.; Shao, J. Z. Bone Morphogenetic Protein. Biochem.

(153) Chenard, K. E.; Teven, C. M.; He, T. C.; Reid, R. R. Bone Morphogenetic Proteins

Personalized Stem Cell Therapy. J. Biomed. Biotechnol. 2012, 2012, 1 - 14.

in Craniofacial Surgery: Current Techniques, Clinical Experiences, and the Future of

Biophys. Re. Commun. 2007, 362 (3), 550 - 553.

fro

(155) Rengachary, S. S. Bone Morphogenetic Proteins: Basic Concepts. Neurosurg.
 m Focus 2002, 13 (6), 1–6.

- the
- (156) Miyazono, K. Signal Transduction by Bone Morphogenetic Protein Receptors:
   do
   Functional Roles of Smad Proteins. *Bone* 1999, 25 (1), 91–93.
- cu

(157) Cecchi, S.; Bennet, S. J.; Arora, M. Bone Morphogenetic Protein-7: Review of me
 Signalling and Efficacy in Fracture Healing. J. Orthop. Translat. 2016, 4, 28 - 34.

- nt
- (158) Oryan, A.; Alidadi, S.; Moshiri, A.; Bigham-Sadegh, A. Bone Morphogenetic or
   Proteins: A Powerful Osteoinductive Compound with Non-Negligible Side Effects use
   and Limitations: Bone Morphogenetic Proteins in Bone Healing. *BioFactors* 2014,
- thi
- 40 (5), 459–481.
- S
- (159) Heldin, C.-H.; Moustakas, A. Signaling Receptors for TGF-β Family Members.
   spa
   *Cold Spring Harb. Perspect. Biol.* 2016, 8 (8), a022053.
- ce
- (160) Chen, G.; Deng, C.; Li, Y.-P. TGF-β and BMP Signaling in Osteoblast to
   Differentiation and Bone Formation. *Int. J. Biol. Sci.* 2012, 8 (2), 272–288.
- em
- ph

135

ze

	der
reat quote from	
	's
he document or	
	att
se this space to	
	ent
mphasize a key	
	ion
oint. To place	
	wit
his text box	
1 .1	h a
nywhere on the	
···· · ··· · · · · · ·	gre
bage, just drag	
( )	at

t.]

 (161) Chan, C. K. Y.; Mason, A.; Cooper, C.; Dennison, E. Novel Advances in the ent Treatment of Osteoporosis. 2016, 119, 13.
 ion

- (162) Wang, N.; Zhao, G.; Zhang, Y.; Wang, X.; Zhao, L.; Xu, P.; Shou, D. A Network wit
  Pharmacology Approach to Determine the Active Components and Potential Targets
  h a
  of Curculigo Orchioides in the Treatment of Osteoporosis. *Med. Sci. Monit.* 2017, 23,
  gre 5113–5122.
- (163) Diwan, A. D.; Leong, A.; Appleyard, R.; Bhargav, D.; Fang, Z. M.; Wei, A. Bone
   qu
   Morphogenetic Protein-7 Accelerates Fracture Healing in Osteoporotic Rats. *Indian* ote
   J. Orthop. 2013, 47 (6), 540–546.
- fro

(164) Varanasi, S. S.; Tuck, S. P.; Mastana, S. S.; Dennison, E.; Cooper, C.; Vila, J.;
m Francis, R. M.; Datta, H. K. Lack of Association of Bone Morphogenetic Protein 2 the Gene Haplotypes with Bone Mineral Density, Bone Loss, or Risk of Fractures in Men.
do J. Osteoporos 2011, 2011.

- J. Osleoporos 20
- cu

(165) Odén, A.; McCloskey, E. V.; Kanis, J. A.; Harvey, N. C.; Johansson, H. Burden of me
 High Fracture Probability Worldwide: Secular Increases 2010-2040. Osteoporos Int.
 nt
 2015, 26 (9), 2243–2248.

or

(166) Yp, Z.; Ry, X.; B, Z.; F, Z.; Xs, Z.; Ll, Z.; H, L. Gender Differences on Osteoporosis use
 Health Beliefs and Related Behaviors in Non-Academic Community Chinese. J
 thi
 Community Health 2014, 39 (3), 545–551.

S

(167) Khosla, S.; Hofbauer, L. C. Osteoporosis Treatment: Recent Developments and spa
 Ongoing Challenges. *The Lancet Diabetes & Endocrinology* 2017, 5 (11), 898–907.
 ce

136

- to
- em
- ph

- 70
- ze

reat quote from	der
ne document or	's
se this space to	att (168) Zhao, B.; Xing, G.; Wang, A. The BMP Signaling Pathway Enhances the
mphasize a key	ent Osteoblastic Differentiation of Bone Marrow Mesenchymal Stem Cells in Rats with
oint. To place	ion Osteoporosis. J. Orthop. Surg. Res. <b>2019</b> , 14 (1), 462.
nis text box	wit (169) Kanakaris, N. K.; Petsatodis, G.; Tagil, M.; Giannoudis, P. V. Is There a Role for
nywhere on the	h a Bone Morphogenetic Proteins in Osteoporotic Fractures? <i>Injury</i> <b>2009</b> , <i>40</i> , S21–S26.
age, just drag	gre (170) Khosla, S.; Westendorf, J. J.; Oursler, M. J. Building Bone to Reverse Osteoporosis
.]	at and Repair Fractures. J. Clin. Invest. 2008, 118 (2), 421–428.
	qu (171) Yang, J. Bone Morphogenetic Proteins: Relationship between Molecular Structure
	ote and Their Osteogenic Activity. Food Science and Human Wellness 2014, 9.
	fro (172) Dumic-Cule, I.; Peric, M.; Kucko, L.; Grgurevic, L.; Pecina, M.; Vukicevic, S.
	m Bone Morphogenetic Proteins in Fracture Repair. International Orthopaedics
	the ( <i>SICOT</i> ) <b>2018</b> , <i>42</i> (11), 2619–2626.
	do (173) Feng, L.; Cook, B.; Tsai, SY.; Zhou, T.; LaFlamme, B.; Evans, T.; Chen, S.
	cu Discovery of a Small-Molecule BMP Sensitizer for Human Embryonic Stem Cell
	me Differentiation. <i>Cell Rep.</i> <b>2016</b> , <i>15</i> (9), 2063–2075.
	nt (174) Vrijens, K.; Lin, W.; Cui, J.; Farmer, D.; Low, J.; Pronier, E.; Zeng, FY.; Shelat,
	or A. A.; Guy, K.; Taylor, M. R.; Chen, T.; Roussel, M. F. Identification of Small
	use Molecule Activators of BMP Signaling. <i>PLoS ONE</i> <b>2013</b> , 8 (3), e59045.
	thi (175) Yu, P. B.; Hong, C. C.; Sachidanandan, C.; Babitt, J. L.; Deng, D. Y.; Hoyng, S.
	s A.; Lin, H. Y.; Bloch, K. D.; Peterson, R. T. Dorsomorphin Inhibits BMP Signals
	spa Required for Embryogenesis and Iron Metabolism. Nat. Chem. Biol. 2008, 4 (1), 33–
	ce 41.
	to
	em 107
	ph 137
	asi

most mosts from	der
reat quote from	's
he document or	att
se this space to	(176) Pigeon, C.; Ilyin, G.; Courselaud, B.; Leroyer, P.; Turlin, B.; Brissot, P. A New
mphasize a key	Mouse Liver-Specific Gene, Encoding a Protein Homologous to Human
oint. To place	Antimicrobial Peptide Hepcidin, Is Overexpressed during Iron Overload. 10.
his text box	(177) Fraenkel, P. G.; Traver, D.; Donovan, A.; Zahrieh, D.; Zon, L. I. Ferroportin1 Is
nywhere on the	h a Required for Normal Iron Cycling in Zebrafish. J. Clin. Invest. 2005, 115 (6), 1532–
oage, just drag	1541.
t.]	at (178) Silvestri, L.; Nai, A.; Dulja, A.; Pagani, A. Hepcidin and the BMP-SMAD Pathway:
	qu An Unexpected Liaison. In Vitamins and Hormones; Elsevier, 2019; Vol. 110, pp 71–
	ote 99.
	fro (179) Nemeth, E. Hepcidin Regulates Cellular Iron Efflux by Binding to Ferroportin and
	m Inducing Its Internalization. Science 2004, 306 (5704), 2090–2093.
	the (180) Anderson, G. J.; Darshan, D. Small-Molecule Dissection of BMP Signaling. <i>Nat</i> .
	do <i>Chem. Biol.</i> <b>2008</b> , <i>4</i> (1), 15–16.
	cu (181) Cuny, G. D.; Yu, P. B.; Laha, J. K.; Xing, X.; Liu, JF.; Lai, C. S.; Deng, D. Y.;
	me Sachidanandan, C.; Bloch, K. D.; Peterson, R. T. Structure–Activity Relationship
	nt Study of Bone Morphogenetic Protein (BMP) Signaling Inhibitors <i>Bioorg Med</i>
	or <i>Cham Latt</i> <b>2008</b> <i>18</i> (15) <i>4</i> 388 <i>4</i> 392
	Use $(100)$ H = C = C = H = D = A = 15 = 1 = C = 0.000 H = D = D = D = D = D = D = D = D = D =
	(182) Hong, C. C.; Yu, P. B. Applications of Small Molecule BMP Inhibitors in thi
	Physiology and Disease. Cytokine & Growth Factor Reviews 2009, 20 (5–6), 409– s
	418. spa
	(183) Baek, S.; Choi, SW.; Park, SJ.; Lee, SH.; Chun, HS.; Kim, S. H. Quinoline
	Compound KM11073 Enhances BMP-2-Dependent Osteogenic Differentiation of
	to
	em
	ph 138
	asi

der reat quote from he document or se this space to mphasize a key oint. To place his text box nywhere on the gre age, just drag

t.]

's att ent ion

C2C12 Cells via Activation of P38 Signaling and Exhibits In Vivo Bone Forming Activity. PLoS ONE 2015, 10 (3), e0120150.

- (184) Cao, Y.; Wang, C.; Zhang, X.; Xing, G.; Lu, K.; Gu, Y.; He, F.; Zhang, L. Selective wit Small Molecule Compounds Increase BMP-2 Responsiveness by Inhibiting Smurf1h a Mediated Smad1/5 Degradation. Sci. Rep. 2015, 4 (1), 4965.
- (185) Beaulieu, P. L.; Gillard, J.; Jolicoeur, E.; Duan, J.; Garneau, M.; Kukolj, G.; at Poupart, M.-A. From Benzimidazole to Indole-5-Carboxamide Thumb Pocket I qu Inhibitors of HCV NS5B Polymerase. Part 1: Indole C-2 SAR and Discovery of ote Diamide Derivatives with Nanomolar Potency in Cell-Based Subgenomic Replicons. fro Bioorg. Med. Chem. Lett. 2011, 21 (12), 3658–3663.
- m
- (186) Brands, M.; Ergüden, J.-K.; Hashimoto, K.; Heimbach, D.; Schröder, C.; Siegel, the S.; Stasch, J.-P.; Weigand, S. Novel, Selective Indole-Based ECE Inhibitors: Lead do Optimization via Solid-Phase and Classical Synthesis. Bioorg. Med. Chem. Lett. cu **2005**, *15* (19), 4201–4205.
- me

(187) Zhuo, S.-T.; Li, C.-Y.; Hu, M.-H.; Chen, S.-B.; Yao, P.-F.; Huang, S.-L.; Ou, T.nt M.; Tan, J.-H.; An, L.-K.; Li, D.; Gu, L.-Q.; Huang, Z.-S. Synthesis and Biological or Evaluation of Benzo[a]Phenazine Derivatives as a Dual Inhibitor of Topoisomerase I use and II. Org. Biomol. Chem. 2013, 11 (24), 3989.

thi

(188) Li, P.-H.; Zeng, P.; Chen, S.-B.; Yao, P.-F.; Mai, Y.-W.; Tan, J.-H.; Ou, T.-M.; S Huang, S.-L.; Li, D.; Gu, L.-Q.; Huang, Z.-S. Synthesis and Mechanism Studies of spa 1,3-Benzoazolyl Substituted Pyrrolo[2,3- b]Pyrazine Derivatives as Nonintercalative ce Topoisomerase II Catalytic Inhibitors. J. Med. Chem. 2016, 59 (1), 238-252. to

139

- em
- ph

asi

- der reat quote from 's he document or att use this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the gre age, just drag at t.] qu ote fro
  - (189) Li, P.; Zhang, W.; Jiang, H.; Li, Y.; Dong, C.; Chen, H.; Zhang, K.; Du, Z. Design, ent
    Synthesis and Biological Evaluation of Benzimidazole–Rhodanine Conjugates as ion
    Potent Topoisomerase II Inhibitors. *MedChemComm* 2018, 9 (7), 1194–1205.
  - (190) Mosmann, T. Rapid Colorimetric Assay for Cellular Growth and Survival:
     h a Application to Proliferation and Cytotoxicity Assays. J. Immunol. Methods 1983, 65 gre
     (1-2), 55-63.
    - (1-2), 55-05.
  - (191) Refaat, H. M. Synthesis and Anticancer Activity of Some Novel 2-Substituted
     gu
     Benzimidazole Derivatives. *Eur. J. Med. Chem.* 2010, 45 (7), 2949–2956.
  - (192) Ozden, S.; Karataş, H.; Yildiz, S.; Göker, H. Synthesis and Potent Antimicrobial fro
     Activity of Some Novel 4-(5, 6-Dichloro-1H-Benzimidazol-2-Yl)-N-Substituted
     m
     Benzamides. Arch. Pharm. (Weinheim) 2004, 337 (10), 556–562.
  - the
  - (193) Stevens, C. L.; Singhal, G. H.; Ash, A. B. Carbodiimides. Dehydration of Ureas. J. do
    Org. Chem. 1967, 32 (9), 2895–2895.
  - cu

(194) Wan, Z.-K.; Ousman, E. F.; Papaioannou, N.; Saiah, E. Phosphonium-Mediated me
 Cyclization of N-(2-Aminophenyl)Thioureas: Efficient Synthesis of 2-nt
 Aminobenzimidazoles. *Tetrahedron Lett.* 2011, 52 (32), 4149–4152.

- or (195) Kaushik, N. K.; Kaushik, N.; Attri, P.; Kumar, N.; Kim, C. H.; Verma, A. K.; Choi, use
  - E. H. Biomedical Importance of Indoles. *Molecules* **2013**, *18* (6), 6620–6662.
- thi
- (196) Maekawa, K.; Ohtani, J. Synthesis of 1,2,3,4-Tetrahydro-1-Oxo-Pyrido[1,2-s
   a]Benzimidazole. Agric. Bio. Chem. 1978, 42 (2), 483 484.
- spa
- (197) Singh, S.; Prasad, N. R.; Chufan, E. E.; Patel, B. A.; Wang, Y.-J.; Chen, Z.-S.;
  Ambudkar, S. V.; Talele, T. T. Design and Synthesis of Human ABCB1 (P-to
  Glycoprotein) Inhibitors by Peptide Coupling of Diverse Chemical Scaffolds on

140

- em
- ph
- ze

	der
reat quote from	's
he document or	att
se this space to	Carboxyl and Amino Termini of (S)-Valine-Derived Thiazole Amino Acid. J. Med.
mphasize a key	<i>Chem.</i> <b>2014</b> , <i>57</i> (10), 4058–4072.
oint. To place	(198) Akhtar, W.; Khan, M. F.; Verma, G.; Shaquiquzzaman, M.; Rizvi, M. A.; Mehdi,
his text box	S. H.; Akhter, M.; Alam, M. M. Therapeutic Evolution of Benzimidazole Derivatives
nywhere on the	h a in the Last Quinquennial Period. Eur. J. Med. Chem. 2017, 126, 705–753.
oage, just drag	gre (199) Ateş-Alagöz, Z.; Kuş, C.; Çoban, T. Synthesis and Antioxidant Properties of Novel
t.]	at Benzimidazoles Containing Substituted Indole or 1,1,4,4-Tetramethyl-1,2,3,4-
	qu Tetrahydro-Naphthalene Fragments. J. Enzyme Inhib. Med. Chem. 2005, 20 (4), 325–
	ote 331.
	fro (200) Lee V. T.: Chiu F. V.: Barve I. I.: Sun C. M. Microwave-Assisted Synthesis of
	m
	Benzimidazole-Linked Indoline and Indole Hybrids from C-2 Linked ( $o$ - the
	Aminobenzyl)Benzimidazoles. Adv. Synth. Catal. 2018, 360 (3), 502–512.
	(201) Yu, X. H.; Hong, X. Q.; Chen, W. H. Fluorinated Bisbenzimidazoles: A New Class
	of Drug-like Anion Transporters with Chloride-Mediated, Cell Apoptosis-Inducing
	me Activity. Org. Biomol. Chem. <b>2019</b> , 17 (6), 1558–1571.
	nt (202) Simó-Vicens, R.; Bomholtz, S. H.; Sørensen, U. S.; Bentzen, B. H. 2.6-Bis(2-
	Or Benzimidazolul)Puridine (BBP) Is a Potent and Selective Inhibitor of Small
	use
	Conductance Calcium-Activated Potassium (SK) Channels. Front. Pharmacol. 2018, thi
	<i>9</i> , 1409.
	(203) Petoud, S.; Bünzli, J. C. G.; Schenk, K. J.; Piguet, C. Luminescent Properties of
	spa Lanthanide Nitrato Complexes with Substituted Bis(Benzimidazolyl)Pyridines.
	ce Inorg. Chem. <b>1997</b> , 36 (7), 1345–1353.
	to
	em
	ph 141
	asi

great quote from	der
he document or	's
se this space to	att (204) Wei, S. Y.; Wang, J. L.; Zhang, C. S.; Xu, X. T.; Zhang, X. X.; Wang, J. X.; Xing,
mphasize a key	ent YH. D <sup>7</sup> /d <sup>8</sup> Metal Complexes Constructed from 2,6-Bis(2-Benzimidazolyl)Pyridyl
oint. To place	ion or 2,6-Di-(Pyrazol-3-Yl)Pyridine Derivatives: Synthesis, Structure, Characterization,
his text box	wit and Photocatalytic Activity. <i>ChemPlusChem</i> <b>2015</b> , 80 (3), 549–558.
nywhere on the	h a (205) Tam, A. YY.; Lam, W. H.; Wong, K. MC.; Zhu, N.; Yam, V. WW.
oage, just drag	gre Luminescent Alkynylplatinum(II) Complexes of 2,6-Bis(N-Alkylbenzimidazol-2'-
t.]	at Yl)Pyridine-Type Ligands with Ready Tunability of the Nature of the Emissive States
	qu by Solvent and Electronic Property Modulation. Chem. Eur. J. 2008, 14 (15), 4562–
	ote 4576.
	fro (206) Musiu, S.; Pürstinger, G.; Stallinger, S.; Vrancken, R.; Haegeman, A.; Koenen, F.;
	m Leyssen, P.; Froeyen, M.; Neyts, J.; Paeshuyse, J. Substituted 2,6-Bis(Benzimidazol-
	the 2-YI)Pyridines: A Novel Chemical Class of Pestivirus Inhibitors That Targets a Hot
	do Spot for Inhibition of Pestivirus Replication in the RNA-Dependent RNA
	cu Polymerase. Antiviral Research 2014, 106, 71–79.
	me (207) Li, L.; Cao, W.; Zheng, W.; Fan, C.; Chen, T. Ruthenium Complexes Containing
	nt 2,6-Bis(Benzimidazolyl)Pyridine Derivatives Induce Cancer Cell Apoptosis by
	or Triggering DNA Damage-Mediated P53 Phosphorylation. Dalton Trans. 2012, 41
	use (41), 12766.
	thi (208) Liu, S.; Cao, W.; Yu, L.; Zheng, W.; Li, L.; Fan, C.; Chen, T. Zinc(Ii) Complexes
	s Containing Bis-Benzimidazole Derivatives as a New Class of Apoptosis Inducers
	spa That Trigger DNA Damage-Mediated P53 Phosphorylation in Cancer Cells. <i>Dalton</i>
	<i>Trans.</i> <b>2013</b> , <i>42</i> (16), 5932.
	to
	em 142
	ph
	asi

	der
great quote from	's
he document or	
se this space to	att (209) Hewitt, N. J.; Hewitt, P. Phase I and II Enzyme Characterization of Two Sources
mphasize a key	of HepG2 Cell Lines. <i>Xenobiotica</i> <b>2004</b> , <i>34</i> (3), 243–256.
oint. To place	(210) Yarlagadda, V.; Akkapeddi, P.; Manjunath, G. B.; Haldar, J. Membrane Active
his text box	Vancomycin Analogues: A Strategy to Combat Bacterial Resistance. J. Med. Chem.
nywhere on the	h a <b>2014</b> , <i>57</i> (11), 4558–4568.
oage, just drag	gre (211) Yoganathan, S.; Miller, S. J. Structure Diversification of Vancomycin through
t.]	at Peptide-Catalyzed, Site-Selective Lipidation: A Catalysis-Based Approach To
	Qu Combat Glycopeptide-Resistant Pathogens. J. Med. Chem. 2015, 58 (5), 2367–2377.
	ote (212) Yadav, S.; Narasimhan, B.; Kaur, H. Perspectives of Benzimidazole Derivatives as
	Anticancer Agents in the New Era. Anti-cancer Agents Med. Chem. 2016, 16 (11),
	m 1403–1425.
	the (213) El Rashedy, A. A.; Aboul-Enein, H. Y. Benzimidazole Derivatives as Potential
	do Anticancer Agents. <i>Mini Rev. Med. Chem.</i> <b>2013</b> , <i>13</i> (3), 399–407.
	cu (214) Willmore, E.; Frank, A. J.; Padget, K.; Tilby, M. J.; Austin, C. A. Etoposide Targets
	me Topoisomerase IIalpha and IIbeta in Leukemic Cells: Isoform-Specific Cleavable
	nt Complexes Visualized and Quantified in Situ by a Novel Immunofluorescence
	or Technique. Mol. Pharmacol. <b>1998</b> , 54 (1), 78–85.
	use (215) Shrestha, A.; Jo, H.; Kwon, Y.; Lee, ES. Design, Synthesis, and Structure-Activity
	thi Relationships of New Benzofuro[3,2-b]Pyridin-7-Ols as DNA Topoisomerase II
	s Inhibitors. <i>Bioorg. Med. Chem. Lett.</i> <b>2018</b> , <i>28</i> (4), 566–571.
	spa (216) Vos, S. M.; Tretter, E. M.; Schmidt, B. H.; Berger, J. M. All Tangled up: How Cells
	ce Direct, Manage and Exploit Topoisomerase Function. <i>Nat. Rev. Mol. Cell Biol.</i> 2011,
	to <i>12</i> (12), 827–841.
	em 143
	ph

asi

reat quote from	der
he document or	's
se this space to	att (217) Atwell, G. J.; Rewcastle, G. W.; Baguley, B. C.; Denny, W. A. Potential Antitumor
mphasize a key	Agents. 50. In Vivo Solid-Tumor Activity of Derivatives of N-[2-
oint. To place	ion (Dimethylamino)Ethyl]Acridine-4-Carboxamide. J. Med. Chem. <b>1987</b> , 30 (4), 664–
his text box	wit 669.
nywhere on the	h a (218) Ortega, J. A.; Riccardi, L.; Minniti, E.; Borgogno, M.; Arencibia, J. M.; Greco, M.
age, just drag	gre L.; Minarini, A.; Sissi, C.; De Vivo, M. Pharmacophore Hybridization To Discover
t.]	at Novel Topoisomerase II Poisons with Promising Antiproliferative Activity. J. Med.
	qu <i>Chem.</i> <b>2018</b> , <i>61</i> (3), 1375–1379.
	ote (219) Beerman, T. A.; McHugh, M. M.; Sigmund, R.; Lown, J. W.; Rao, K. E.; Bathini,
	fro Y. Effects of Analogs of the DNA Minor Groove Binder Hoechst 33258 on
	m Topoisomerase II and I Mediated Activities. <i>Biochim. Biophys. Acta</i> 1992, 1131 (1),
	the 53 - 61.
	do (220) Hasinoff, B. B.; Wu, X.; Nitiss, J. L.; Kanagasabai, R.; Yalowich, J. C. The
	cu Anticancer Multi-Kinase Inhibitor Dovitinib Also Targets Topoisomerase I and
	me Topoisomerase II. <i>Pharmacol.</i> <b>2012</b> , <i>84</i> (12), 1617 - 1626.
	nt (221) Seaton, A.; Higgins, C.; Mann, J.; Baron, A.; Bailly, C.; Neidle, S.; van den Berg,
	or H. Mechanistic and Anti-Proliferative Studies of Two Novel, Biologically Active Bis-
	use Benzimidazoles, <i>Eur. I. Cancer</i> <b>2003</b> 39 (17) 2548–2555.
	thi (222) Tolner, B.: Hartley, J. A.: Hochhauser, D. Transcriptional Regulation of
	S Topoisomeroso II at Confluence and Dharmacological Modulation of Expression by
	spa
	Bis-Benzimidazole Drugs. <i>Mol. Pharmacol.</i> <b>2001</b> , 59 (4), 699 - 706. ce
	to
	em
	ph 144
	asi

der reat quote from 's he document or att use this space to ent mphasize a key ion oint. To place wit his text box nywhere on the age, just drag at t.]

(223) Chamberlin, J.; Story, S.; Ranjan, N.; Chesser, G.; Arya, D. P. Gram-Negative Synergy and Mechanism of Action of Alkynyl Bisbenzimidazoles. Sci. Rep. 2019, 9 (1), 14171.

(224) Bell, C. A.; Dykstra, C. C.; Naiman, N. A.; Cory, M.; Fairley, T. A.; Tidwell, R. R. h a Structure-Activity Studies of Dicationically Substituted Bis-Benzimidazoles against gre Giardia Lamblia: Correlation of Antigiardial Activity with DNA Binding Affinity and Giardial Topoisomerase II Inhibition. Antimicrob. Agents Chemother. 1993, 37 (12), qu 2668 - 2673.

ote

(225) Dale, A. G.; Hinds, J.; Mann, J.; Taylor, P. W.; Neidle, S. Symmetric Bisfro Benzimidazoles Are Potent Anti-Staphylococcal Agents with Dual Inhibitory m Mechanisms against DNA Gyrase. Biochemistry 2012, 51 (29), 5860-5871.

the

(226) Nimesh, H.; Sur, S.; Sinha, D.; Yadav, P.; Anand, P.; Bajaj, P.; Virdi, J. S.; Tandon, do V. Synthesis and Biological Evaluation of Novel Bisbenzimidazoles as Escherichia cu

Coli Topoisomerase IA Inhibitors and Potential Antibacterial Agents. J. Med. Chem. me **2014**, *57* (12), *5238–5257*.

nt

(227) Smith, P. J.; Anderson, C. O. Modification of the Radiation Sensitivity of Human or Tumour Cells by a Bis-Benzimidazole Derivative. Int. J. Radiat. Biol. 1984, 46 (4), use 331 - 344.

thi

(228) Tawar, U.; Jain, A. K.; Dwarakanath, B. S.; Chandra, R.; Singh, Y.; Chaudhury, N. S K.; Khaitan, D.; Tandon, V. Influence of Phenyl Ring Disubstitution on spa Bisbenzimidazole and Terbenzimidazole Cytotoxicity: Synthesis and Biological ce Evaluation as Radioprotectors. J. Med. Chem. 2003, 46 (18), 3785–3792.

145

- to
- em
- ph
- ze

	der
great quote from	uer
he document or	's
ne document of	att
se this space to	ont
mphasize a key	ent
point To place	ion
onn. 10 place	wit
his text box	1
nywhere on the	n a
and just drag	gre
age, just drag	at
t.]	
	qu
	ote
	fro

5324-5330.

use

the

do

cu

nt

or

- thi
- S
- spa
- ce
- to
- ph
- ze

asi

(229) Bhattacharya, S.; Chaudhuri, P. Medical Implications of Benzimidazole Derivatives as Drugs Designed for Targeting DNA and DNA Associated Processes. Curr. Med. Chem. 2008, 15 (18), 1762–1777.

(230) Darii, M. V.; Rakhimova, A. R.; Tashlitsky, V. N.; Kostyuk, S. V.; Veiko, N. N.; Ivanov, A. A.; Zhuze, A. L.; Gromova, E. S. Dimeric Bisbenzimidazoles: Cytotoxicity and Effects on DNA Methylation in Normal and Cancer Human Cells. Mol. Biol. 2013, 47 (2), 259–266.

- (231) Wu, H.; Huang, X.; Yuan, J.; Kou, F.; Chen, G.; Jia, B.; Yang, Y.; Lai, Y. Synthesis, Crystal Structure and DNA-Binding Properties of a Nickel(II) Complex with 2, 6-Bis(2-Benzimidazolyl)Pyridine. Zeitschrift für Naturforschung B 2010, 65 (11), m
  - 1334–1340. (232) Wu, H.; Huang, X.; Yuan, J.; Kou, F.; Jia, F.; Liu, B.; Wang, K. A V-Shaped Ligand 2,6-Bis(2-Benzimidazolyl)Pyridine and Its Picrate Mn(II) Complex: Synthesis,
- Crystal Structure and DNA-Binding Properties. Eur. J. Med. Chem. 2010, 45 (11), me

  - (233) Wu, H.-L.; Huang, X.; Liu, B.; Kou, F.; Jia, F.; Yuan, J.; Bai, Y. Copper(II) Complex Based on a V-Shaped Ligand, 2,6- Bis (2-Benzimidazolyl)Pyridine: Synthesis, Crystal Structure, DNA-Binding Properties, and Antioxidant Activities. J.
    - Coord. Chem. 2011, 64 (24), 4383 4396.
  - (234) Wu, H.-L.; Yuan, J.-K.; Huang, X.-C.; Kou, F.; Liu, B.; Jia, F.; Wang, K.-T.; Bai,
  - Y. Two Zinc(II) and Cadmium(II) Complexes Based on the V-Shaped Ligand 2,6-Bis(2-Benzimidazolyl)Pyridine: Synthesis, Crystal Structure, DNA-Binding Properties and Antioxidant Activities. Inorg. Chim. Acta 2012, 390, 12 - 21.
- em

146

- der reat quote from 's he document or att use this space to ent mphasize a key ion oint. To place wit his text box h a nywhere on the age, just drag t.]
  - 2004, 705 (1–3), 153 157.
    gre

    (237) Chetia, B.; Iyer, P. K. 2,6-Bis(2-Benzimidazolyl)Pyridine Receptor for Urea at Recognition. *Tetrahedron Lett.* 2006, 47 (46), 8115–8117.
    qu

    (238) Chetia, B.; Goutam, P. J.; Chipem, F. A. S.; Iyer, P. K. Thiourea Recognition by

(235) Saikia, E.; Dutta, P.; Chetia, B. A Novel Benzimidazolyl-Based Receptor for the

Recognition of Fluoride and Cyanide Anion. J. Chem. Sci. 2017, 129 (1), 1-7.

(236) Liu, S.-G.; Zuo, J.-L.; Li, Y.-Z.; You, X.-Z. Syntheses, Crystal Structures of Blue

Luminescent Complexes Based on 2,6-Bis(Benzimidazolyl) Pyridine. J. Mol. Struct.

- ote 2,6-Bis(2-Benzimidazolyl)Pyridine Using Spectroscopic Techniques and DFT. J. fro *Mol. Struct.* **2013**, *1042*, 32–36.
- m
- (239) Chetia, B.; Iyer, P. K. 2,6-Bis(2-Benzimidazolyl)Pyridine as a Chemosensor for the Fluoride Ions. *Tetrahedron Lett.* **2008**, *49* (1), 94–97.
- do
- (240) Badiei, A.; Razavi, B. V.; Goldooz, H.; Mohammadi Ziarani, G.; Faridbod, F.;
  cu
  Ganjali, M. R. A Novel Fluorescent Chemosensor Assembled with 2,6-Bis(2-me
  Benzimidazolyl)Pyridine-Functionalized Nanoporous Silica-Type SBA-15 for
  nt
  Recognition of Hg2+ Ion in Aqueous Media. *Int. J. Environ. Res.* 2018, *12* (1), 109–
  or
  115.
- use
- (241) Vosough Razavi, B.; Badiei, A.; Lashgari, N.; Mohammadi Ziarani, G. 2,6-Bis(2-thi Benzimidazolyl)Pyridine Fluorescent Red-Shifted Sensor for Recognition of Zinc(II)
   and a Calorimetric Sensor for Iron Ions. J. Fluoresc. 2016, 26 (5), 1723–1728.
- spa
- (242) Rajnák, C.; Titiš, J.; Fuhr, O.; Ruben, M.; Boča, R. Low Spin Fe(II) Complexes
   Formed of Monosubstitued 2,6-Bis(2-Benzimidazolyl)Pyridine Ligands. *Polyhedron* to
   2017, 123, 122–131.

em

ph

147

ze

der reat quote from he document or att se this space to ent mphasize a key ion oint. To place wit box his text h a nywhere on the age, just drag t.]

's

- (243) Appukuttan, V.; Zhang, L.; Ha, J. Y.; Chandran, D.; Bahuleyan, B. K.; Ha, C.-S.; Kim, I. Stereospecific Polymerizations of 1,3-Butadiene Catalyzed by Co(II) Complexes Ligated by 2,6-Bis(Benzimidazolyl)Pyridines. J. Mol. Catal. A Chem. 2010, 325 (1-2), 84 - 90.
- (244) Gerber, T. I. A.; Mayer, P.; Tshentu, Z. R. Imidazolate Coordination of 2,6-Bis(2gre Benzimidazolyl) Pyridine in a Dimeric Rhenium(V) Complex. J. Coord. Chem. 2005, at 58 (15), 1271 - 1277.
- qu (245) Liu, S.; Pan, R.; Li, G.; Su, W.; Ni, C. Synthesis of Ruthenium Complex Based on ote 2,6-Bis(1-(Phenyl)-1H-Benzo[d]Imidazol-2-Yl)Pyridine and 2-(1-Phenyl-1Hfro Benzo[d]Imidazol-2-Yl)Benzoate and Catalytical Oxidation Property of 1-(1Hm Benzo[d]Imidazol-2-YI)Ethanol to 1-(1H-Benzo[d]Imidazol-2-YI)Ethanone with H<sub>2</sub> the O<sub>2</sub>. J. Chem. 2017, 2017, 1–7.
- do
- (246) Wojtecki, R. J.; Wu, Q.; Johnson, J. C.; Ray, D. G.; Korley, L. T. J.; Rowan, S. J. cu Optimizing the Formation of 2,6-Bis(N-Alkyl-Benzimidazolyl)Pyridine-Containing me [3]Catenates through Component Design. Chem. Sci. 2013, 4 (12), 4440.
- nt (247) Zhang, W.; Sun, W.-H.; Zhang, S.; Hou, J.; Wedeking, K.; Schultz, S.; Fröhlich, or R.; Song, H. Synthesis, Characterization, and Ethylene Oligomerization and use Polymerization of [2,6-Bis(2-Benzimidazolyl)Pyridyl]Chromium Chlorides. thi Organometallics 2006, 25 (8), 1961–1969.
- S

(248) Deng, Z.; Yu, L.; Cao, W.; Zheng, W.; Chen, T. Rational Design of Ruthenium spa Complexes Containing 2,6-Bis(Benzimidazolyl)Pyridine Derivatives with ce Radiosensitization Activity by Enhancing P53 Activation. ChemMedChem 2015, 10 to (6), 991–998.

- em
- ph

148

ze

reat quote from	der
ha dagumant an	's
ne document or	att
se this space to	(249) Chetia, B.; Iyer, P. K. Utilization of 2,6-Bis(2-Benzimidazolyl)Pyridine to Detect ent
mphasize a key	Toxic Benzene Metabolites. <i>Tetrahedron Lett.</i> <b>2007</b> , <i>48</i> (1), 47–50.
oint. To place	(250) Peng, CC.; Li, Z.; Deng, LQ.; Ke, ZF.; Chen, WH. 2,6-Bis(Benzimidazol-2- wit
his text box	Y1)Pyridine as a Potent Transmembrane Anion Transporter. <i>Bioorg. Med. Chem. Lett.</i>
nywhere on the	<b>2016</b> , <i>26</i> (10), 2442–2445.
oage, just drag	gre
t.]	at
	qu
	ote
	fro
	m
	the
	do
	cu
	me
	nt
	or
	use
	thi
	5
	spa
	ce
	to
	em 140
	ph 149
	asi
	ze

......

---

## Vita

Name

Baccalaureate Degree

Date Graduated

Other Degrees and Certificates

Leonard Barasa

Bachelor of Education (Science), Egerton University, Nakuru Major: Mathematics/Chemistry

November, 2001

Master of Science, University of Nairobi, Nairobi, Major: Chemistry

Date Graduated

September, 2011